



H2020-MSCA ITN  
Grant n. 956099



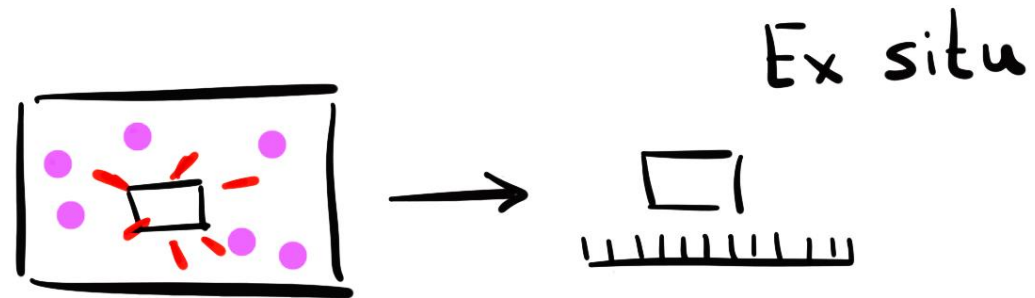
*Nan*ED



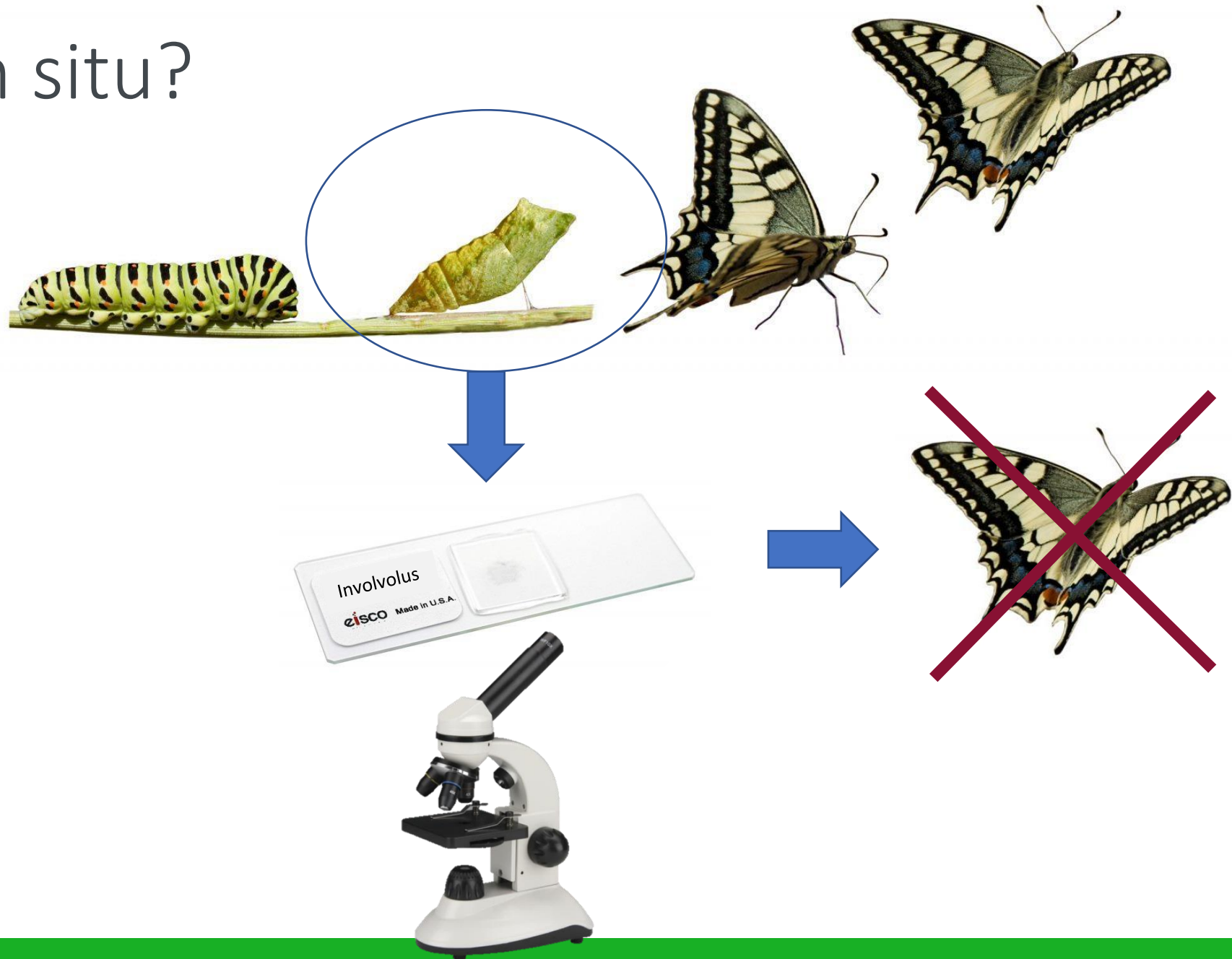
In situ 3D electron diffraction for following reactions in gas and electrochemical environment

Joke Hadermann - University of Antwerp

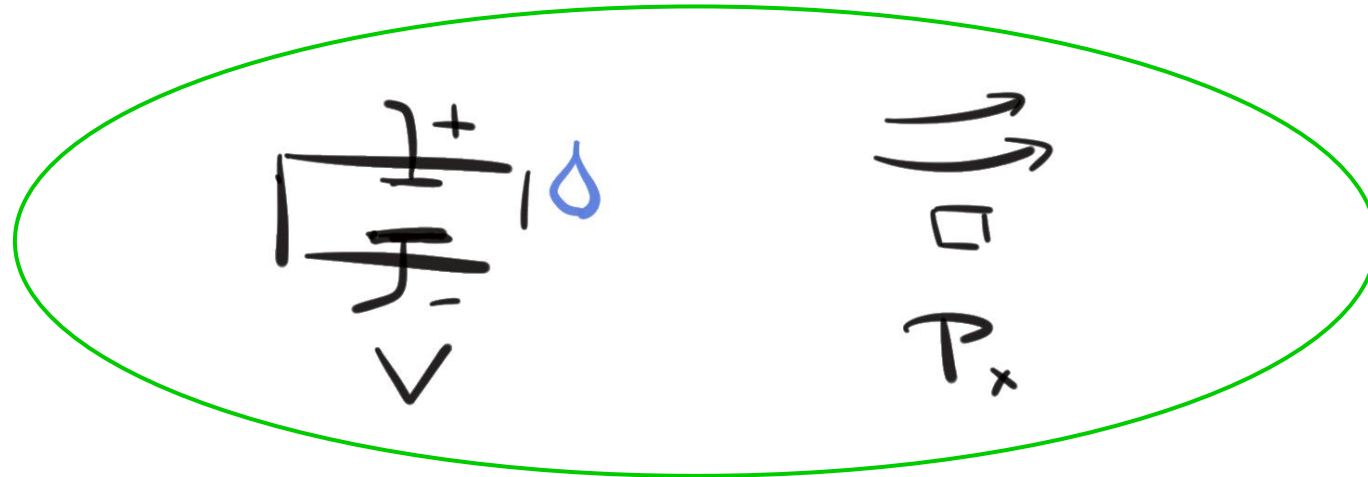
# Ex situ - in situ - operando



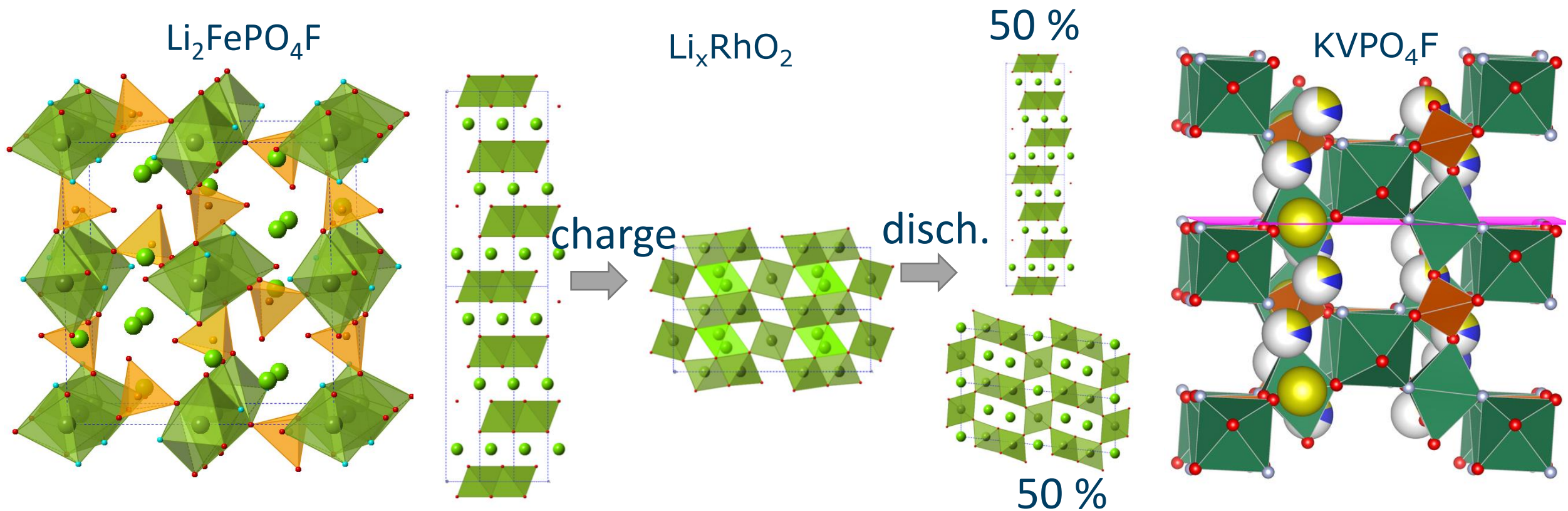
# Why in situ?



Ex situ - in situ - operando



# Ex situ results asking for in situ...



O.M. Karakulina et al. *Chem. Mater.* 2016, 28 (21), 7578–7581.

D. Mikhailova et al. *Inorg. Chem.* 2016, 55 (14), 7079–7089.

S.S. Fedotov et al. *Chemistry of Materials*, 28, 411–415 (2016).



# Possible effects before measurements

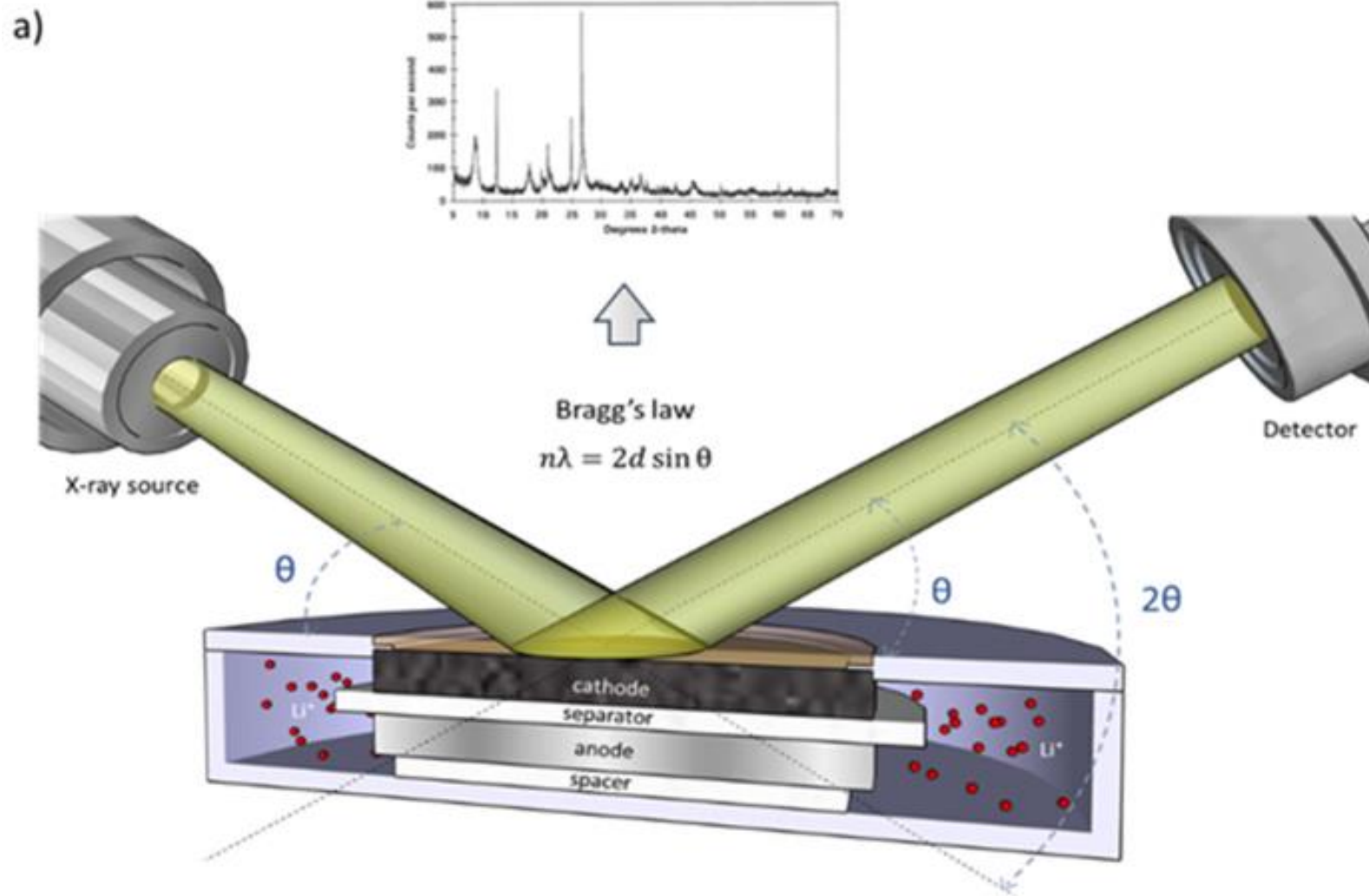


## In situ

- ~~• relaxation time~~
- ~~• strong reducing effect of  $e^-$  beam in vacuum~~
- ~~• small number of cycling stages~~
- ~~• each measurement on a different crystal~~

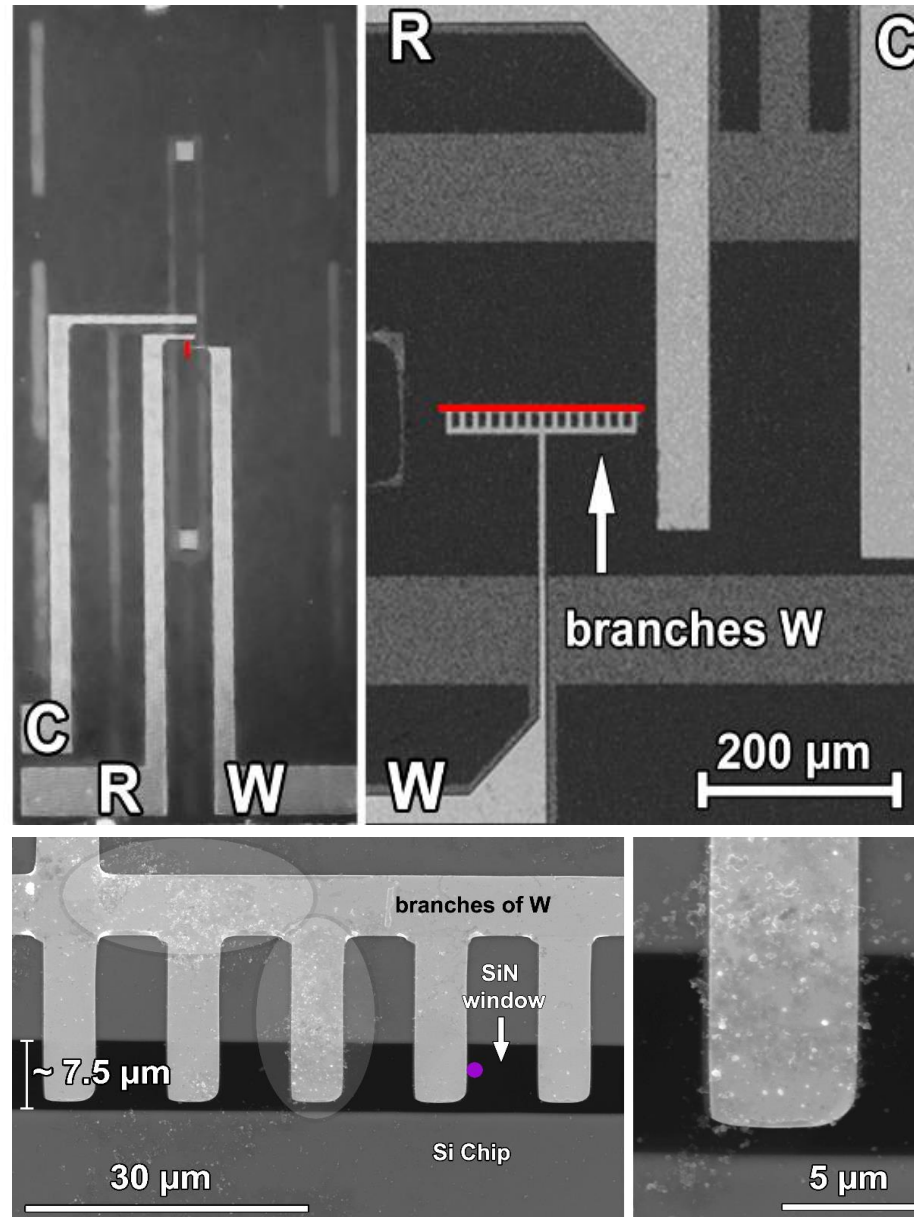


# In situ XRD- in situ ND



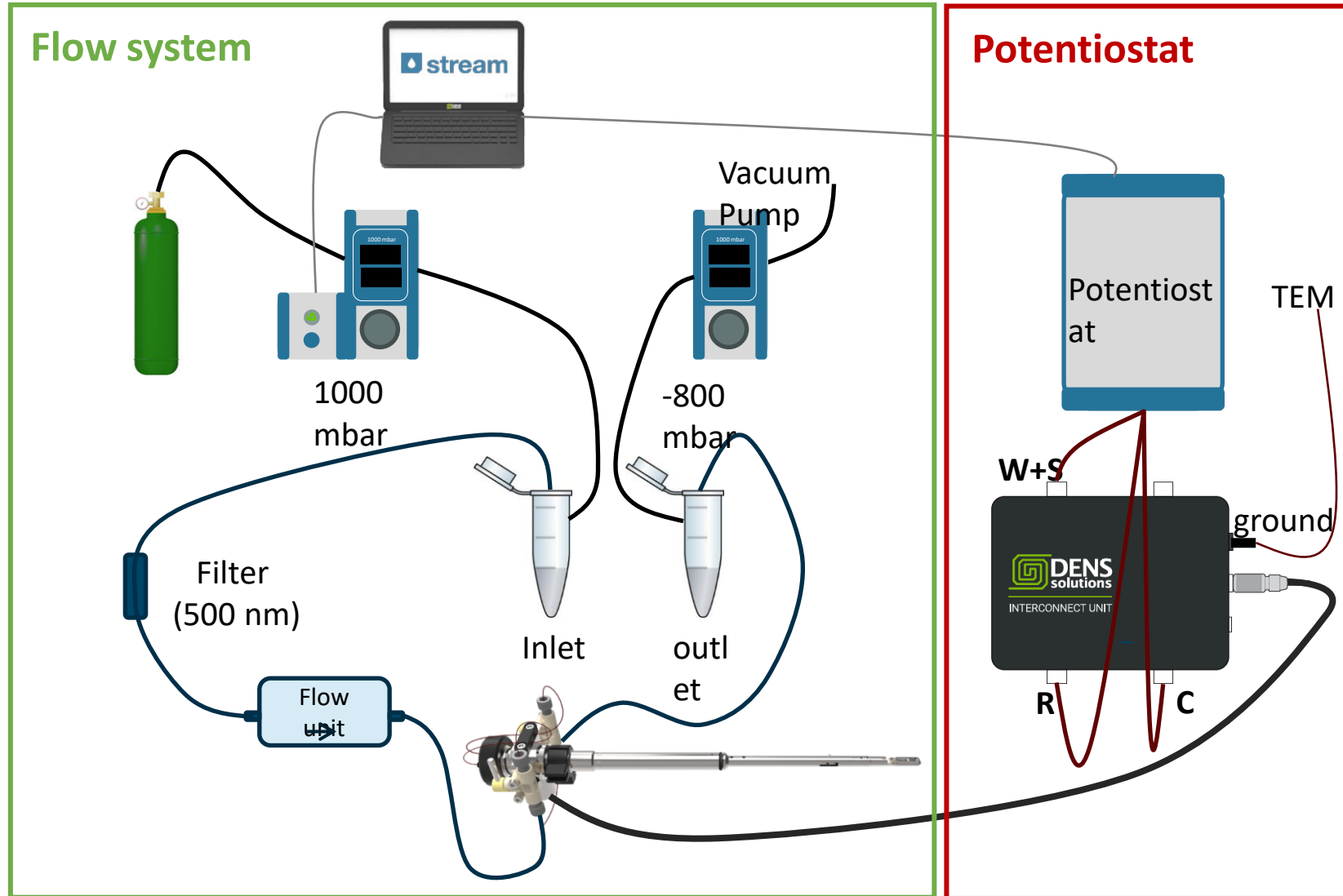


# In situ TEM

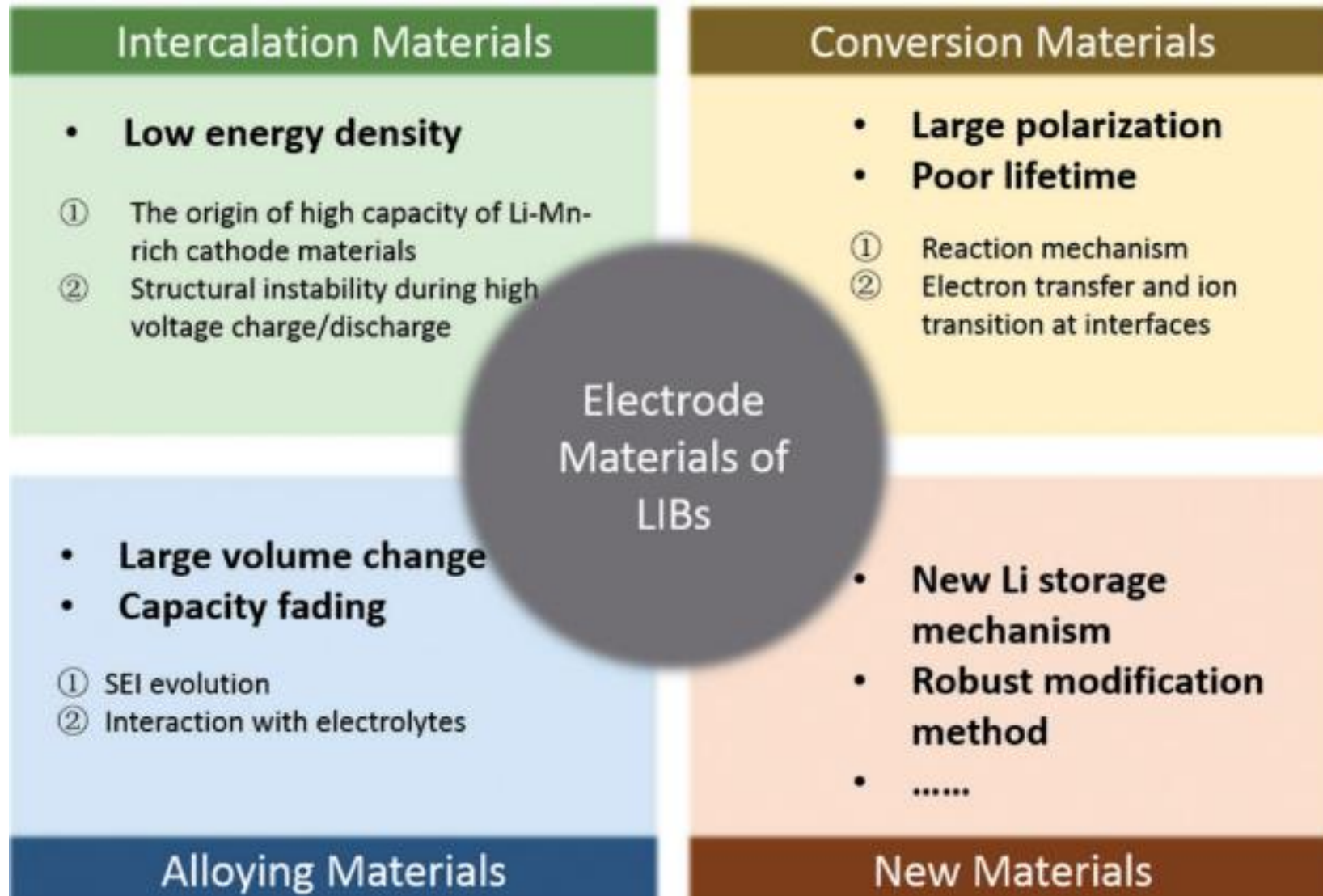




# Setup in situ electrochemistry in TEM



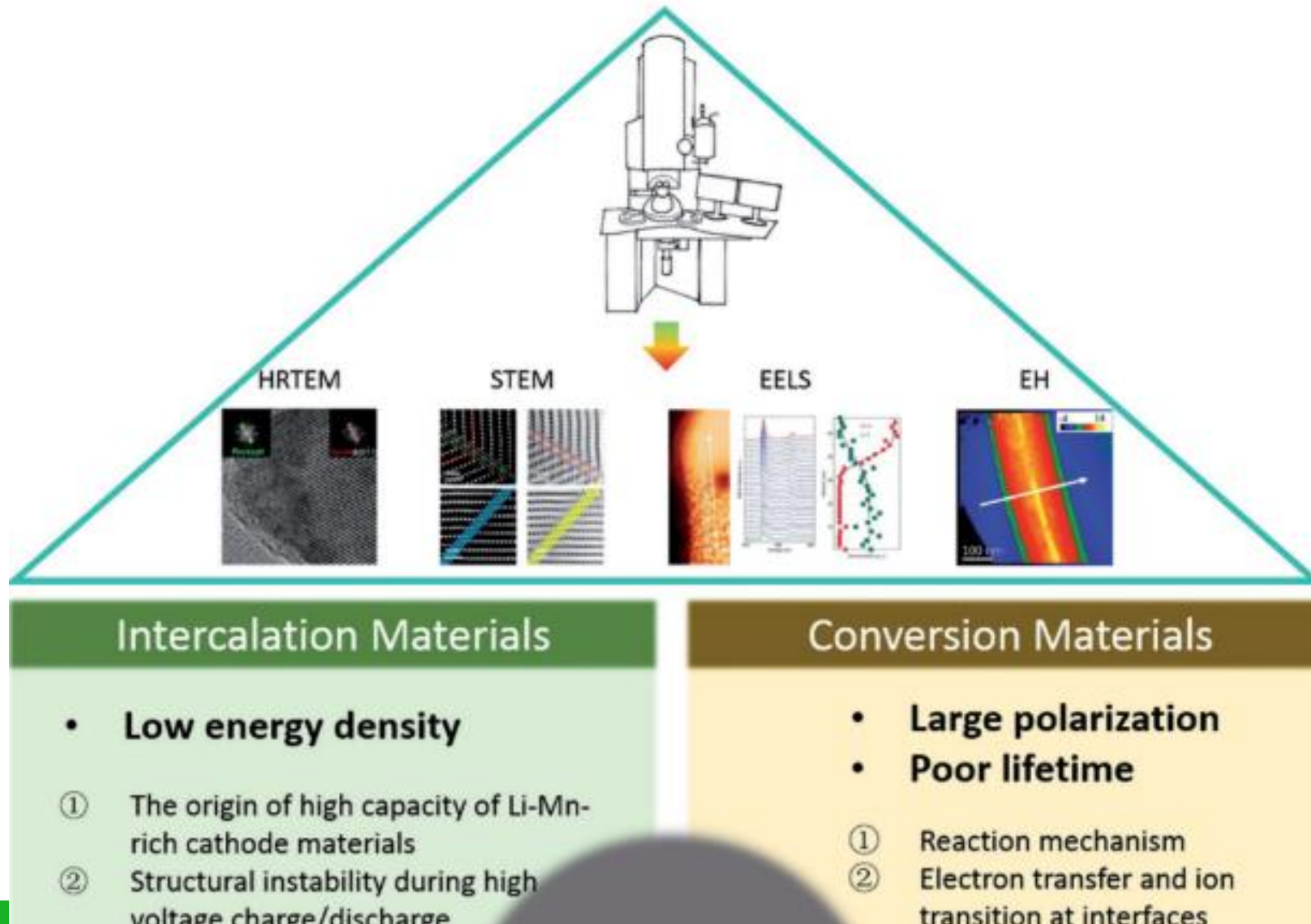
# In situ TEM in literature



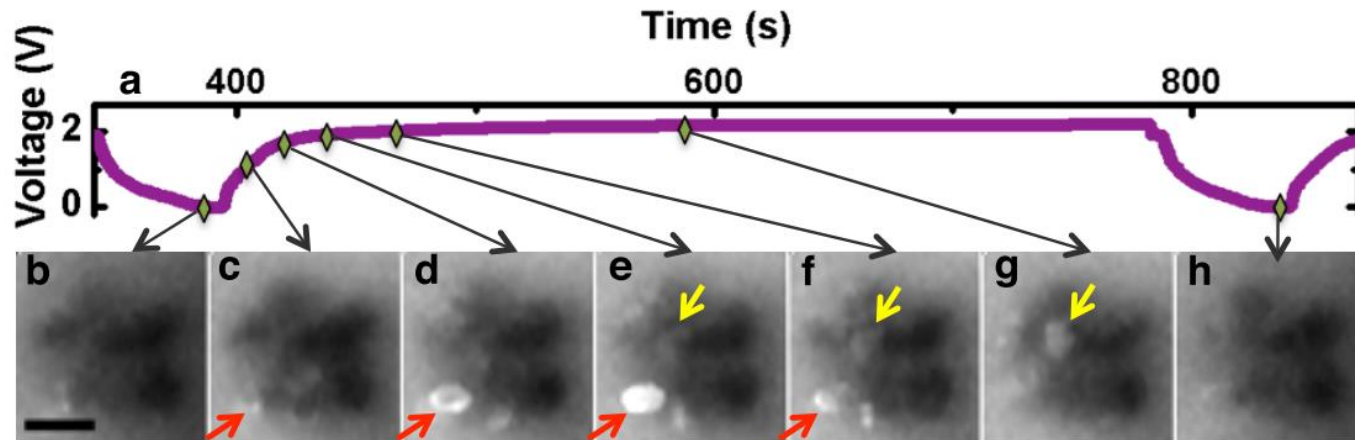
Shang, T., Wen, Y., Xiao, D., Gu, L., Hu, Y. S. & Li, H. (2017). *Adv. Energy Mater.* **1700709**, 1–16.



# In situ TEM in literature



# In situ TEM imaging in an EC cell

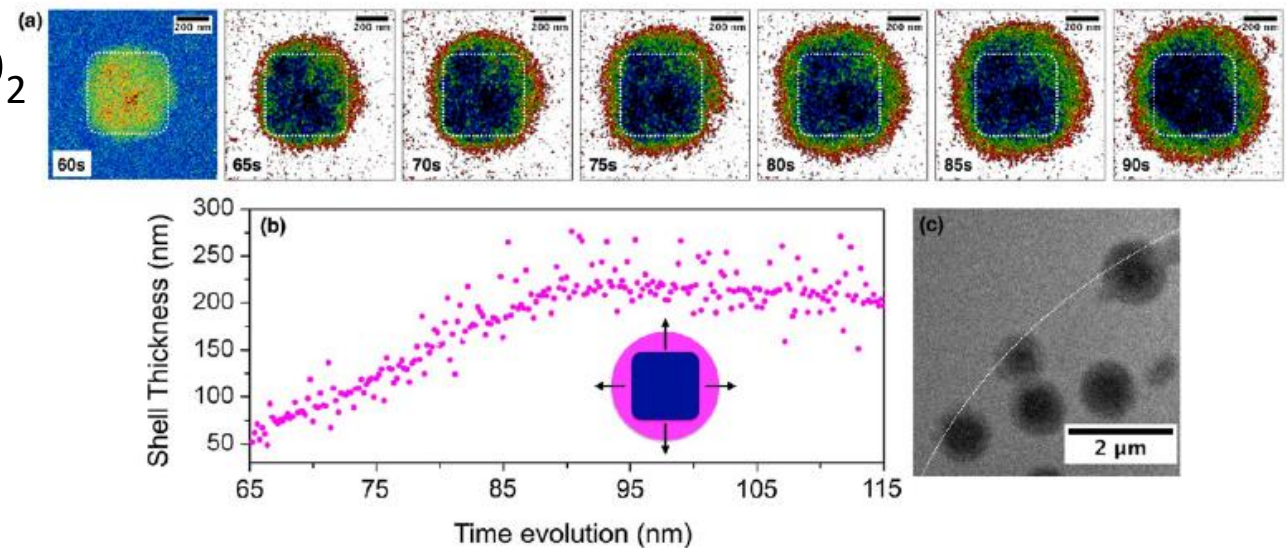


The evolution of a cluster of  $\text{LiFePO}_4$  /  $\text{FePO}_4$  during one charge/discharge cycle

M. Holtz et al. (2014)  
*Nano Lett.* **14**

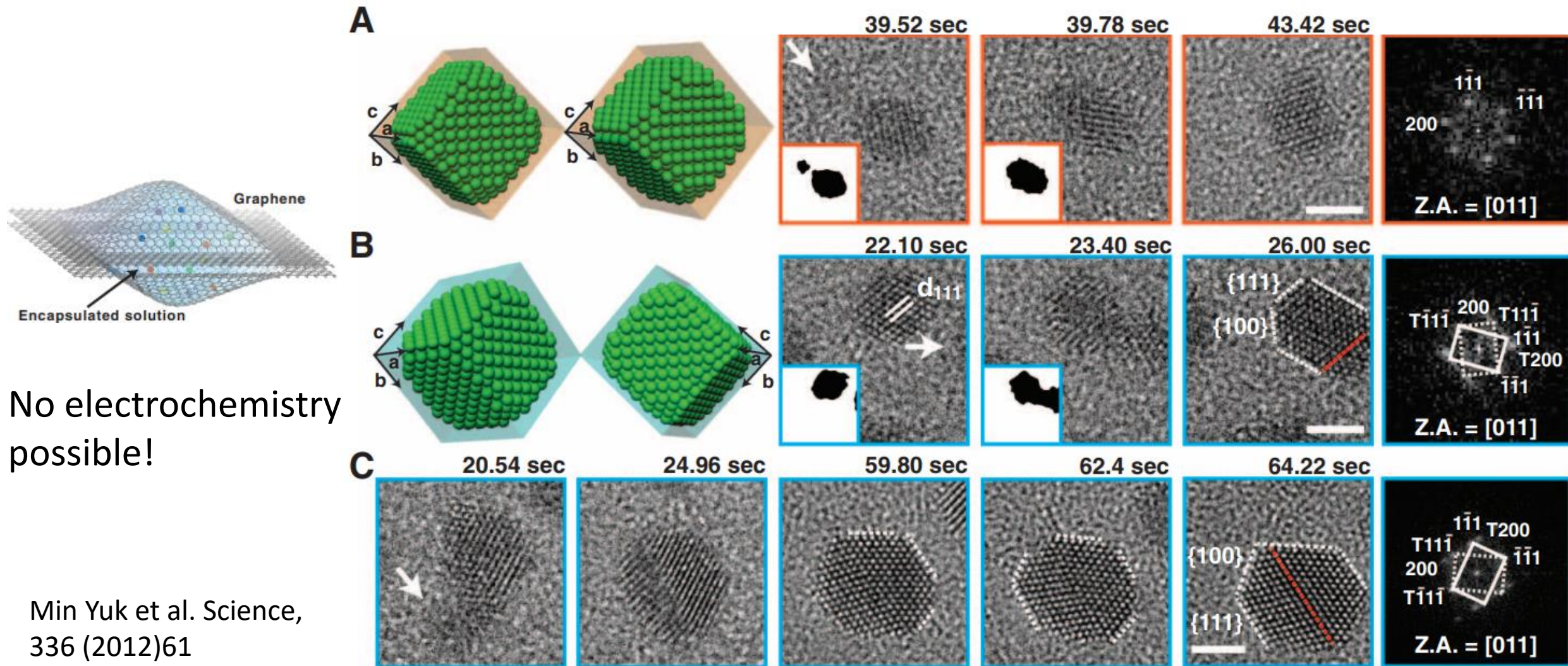
The growth of nanocubes in  $\text{Na-O}_2$  batteries.

L. Lutz et al. (2018) *Nano Lett.* **18**





# In situ TEM imaging in a graphene cell





# State-of-the art of electron diffraction in electrochemical TEM cells

NANO LETTERS

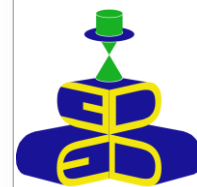
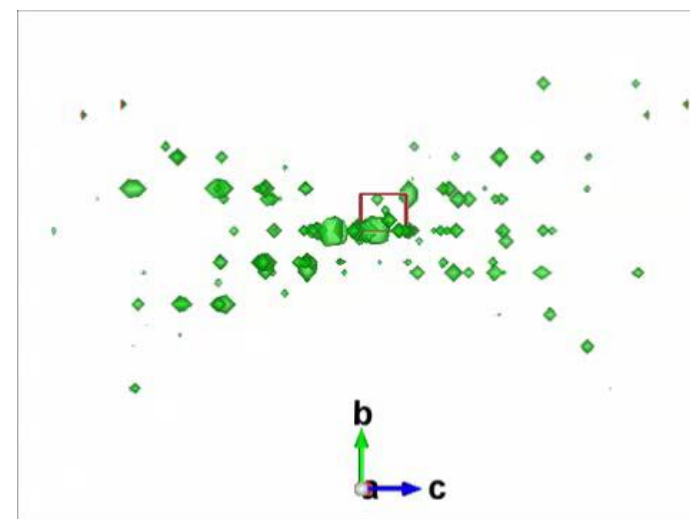
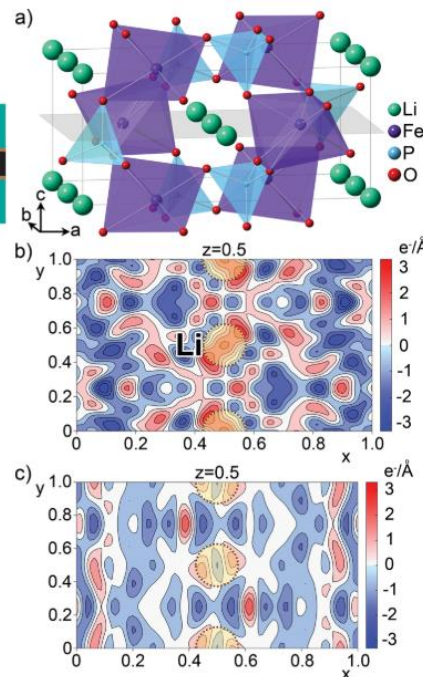
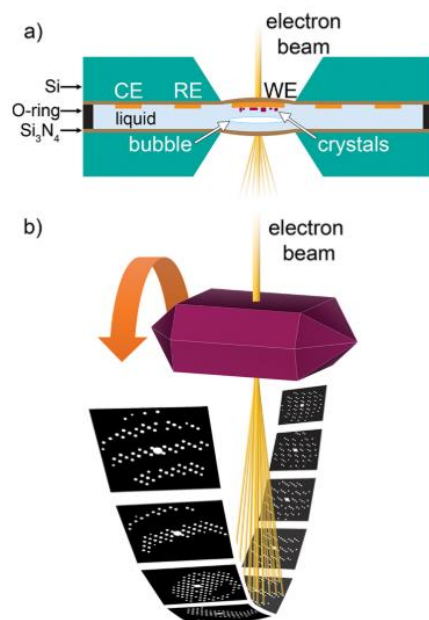
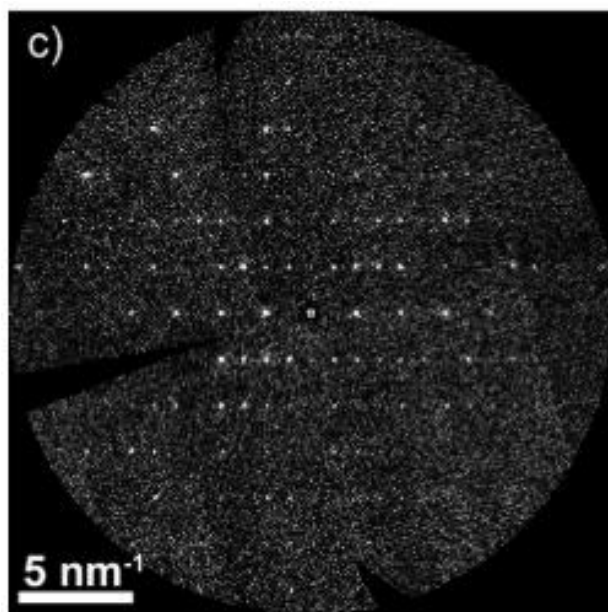
Cite This: *Nano Lett.* 2018, 18, 6286–6291

Letter

[pubs.acs.org/NanoLett](https://pubs.acs.org/NanoLett)

## In Situ Electron Diffraction Tomography Using a Liquid-Electrochemical Transmission Electron Microscopy Cell for Crystal Structure Determination of Cathode Materials for Li-Ion batteries

Olesia M. Karakulina,<sup>†</sup> Arnaud Demortière,<sup>\*,‡,§</sup> Walid Dachraoui,<sup>§</sup> Artem M. Abakumov,<sup>||</sup> and Joke Hadermann<sup>\*,†</sup>





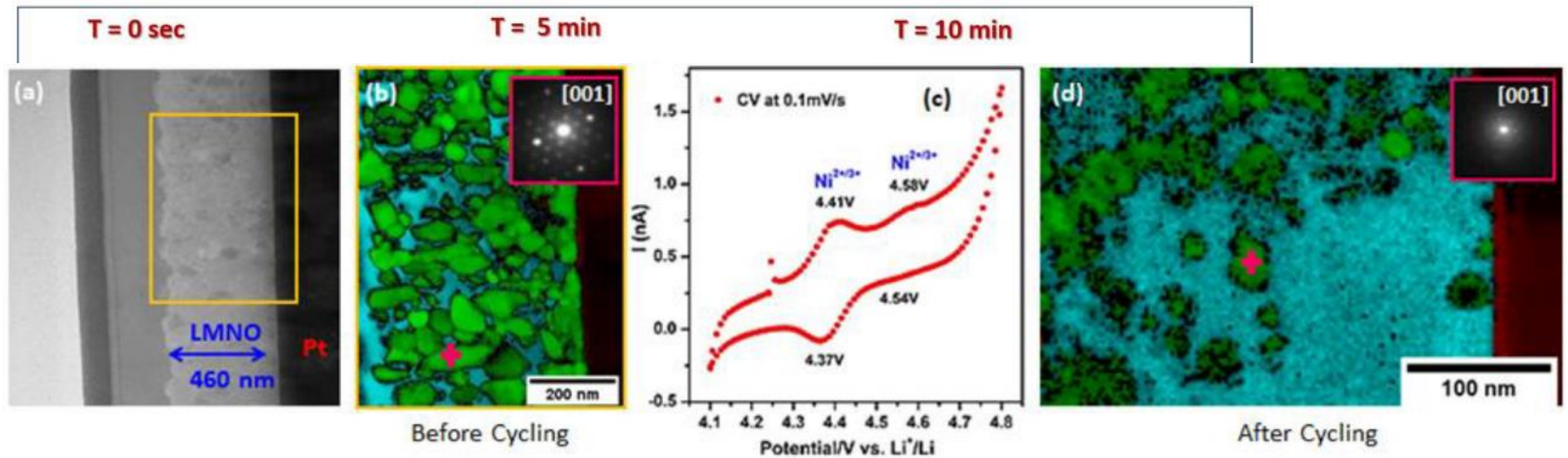
# Lessons learned

Need to avoid intense beam → Need thinner cell

→ DensSolutions Stream 200 nm instead of 500 nm

**Confinement issues?**

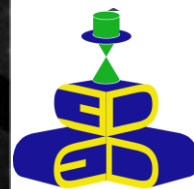
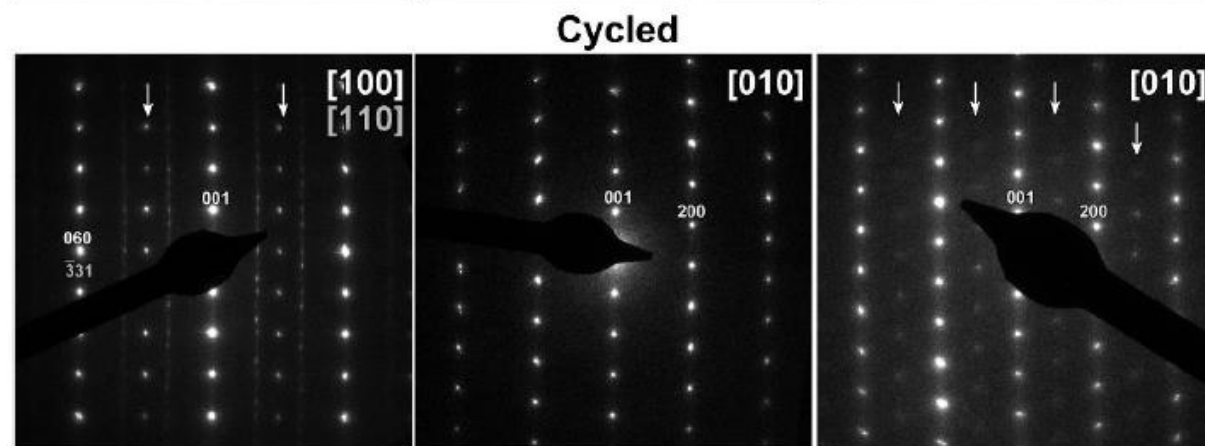
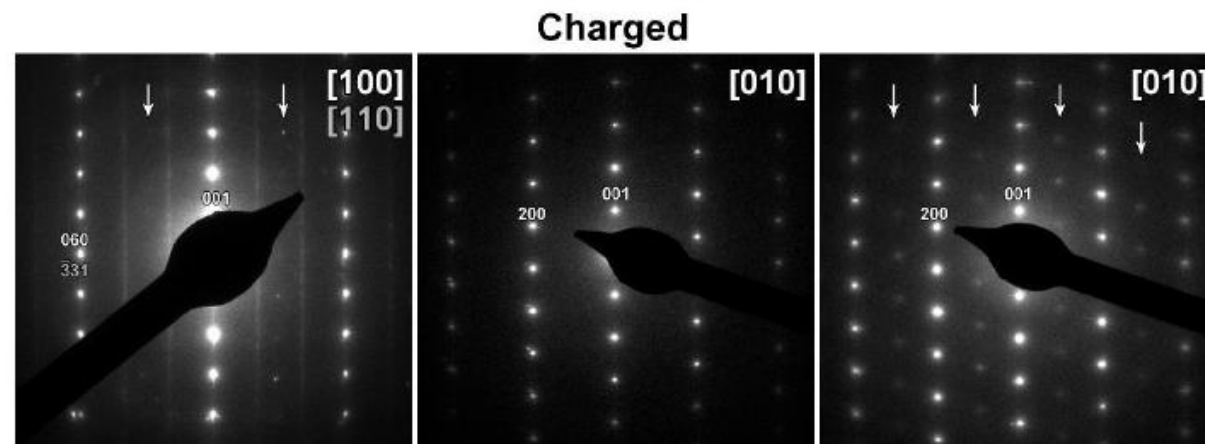
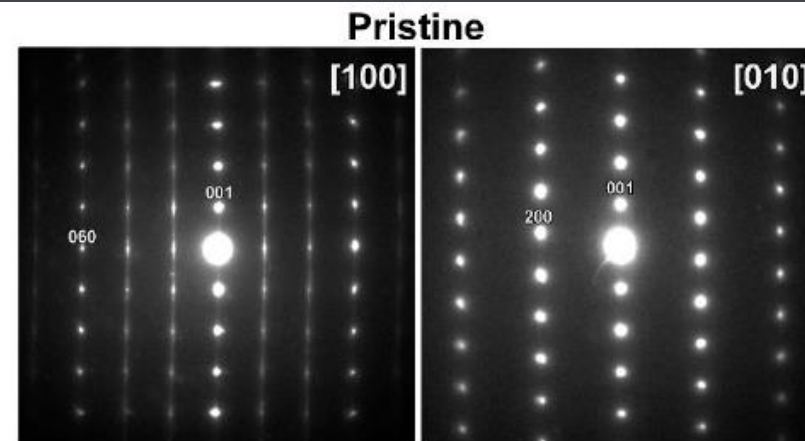
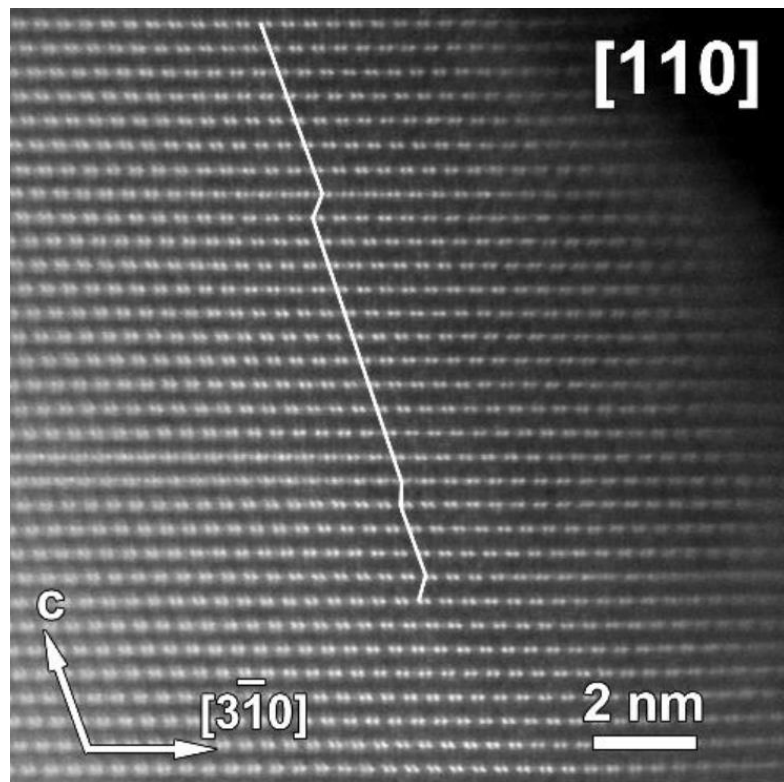




**Figure 2.** (a) Bright-field transmission electron microscopy of the LMNO FIB lamella sample before cycling; (b) phase map before cycling where dark red represents the platinum, in green the LMNO phase, and in blue the amorphous phase; insert bright red shows the electron diffraction pattern of spinel LMNO in [001] orientation; (c) Cyclic voltammogram measured in a holder with a voltage sweep rate of 0.1 mV/s in the voltage window of 4.1 V- 4.8 V vs. lithium using 1M lithium perchlorate in EC:DMC electrolyte; (d) phase map after cycling where dark red represents the platinum, in green the LMNO phase and in blue the amorphous phase; insert bright red shows the electron diffraction pattern of spinel LMNO in [001] orientation.

**Figure 4.** ASTAR orientation map of 220 nm x 220 nm (scan time 20 sec) by Medipix III detector, 20 fps, 12 bit, and step size: 11 nm. White arrows on gold nanoparticles correspond to individual ED patterns, Spacer 200 nm, 30 nm Si<sub>3</sub>N<sub>4</sub> thickness.

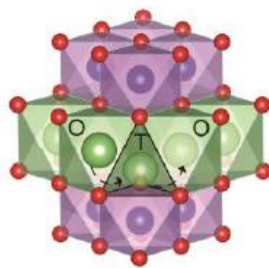
# LR-NMC



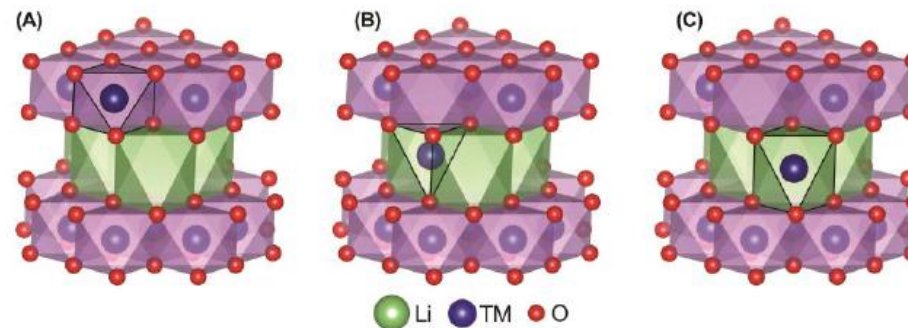


# LR-NMC

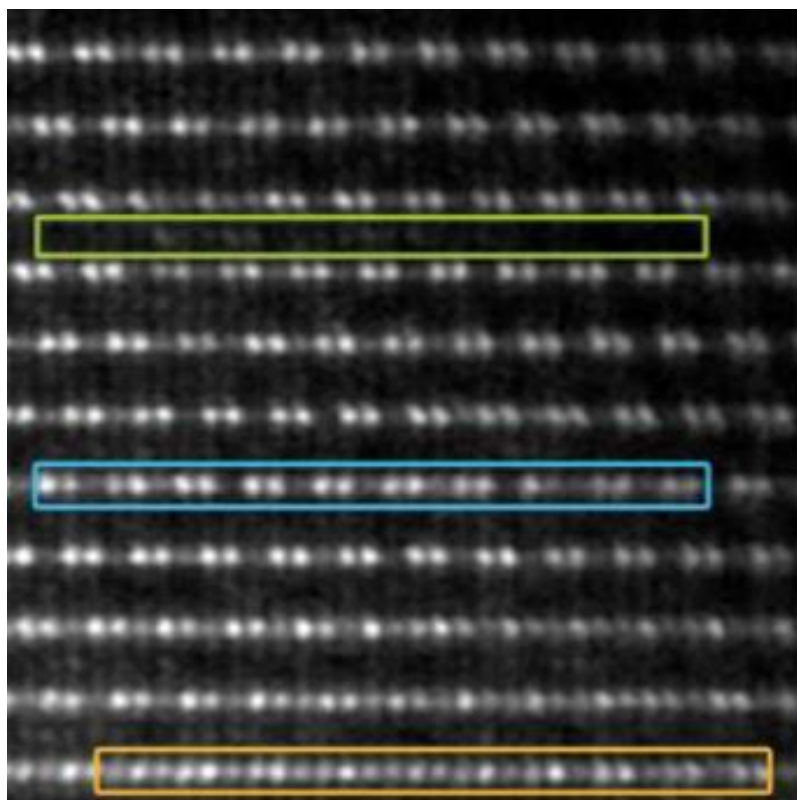
O-T-O diffusion path



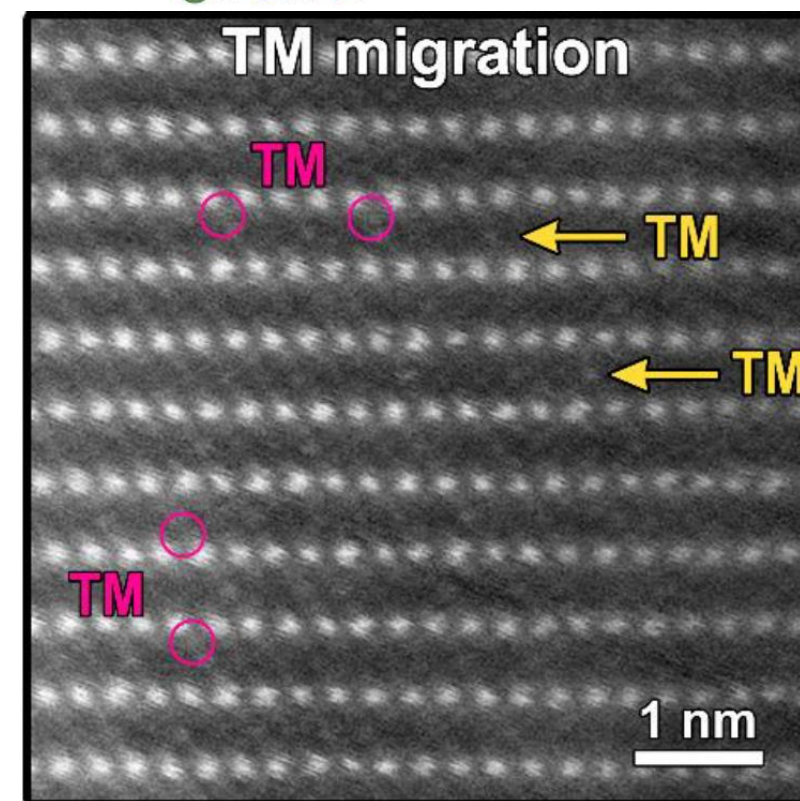
Transition metal migration



Kleiner et al. Chem. Mater. **30**, 3656 (2018)



Pristine



Charged



# ESR 3

- optimize the setup for in situ 3D ED in an electrochemical cell
- tackle contamination issues
- tackle decomposition of the electrolyte
- validate the electrochemistry in the thinner cell
- demonstrate on commercially available lithium battery electrode compounds
- analysis: multiphased, diffuse, defects, twinning



# In situ solid - gas reactions





# Example of a successful experiment: $\text{SrFeO}_{2.5+x}$



Brownmillerite  $\text{Pbcm}$



Orthorhombic  $\text{Cmmm}$

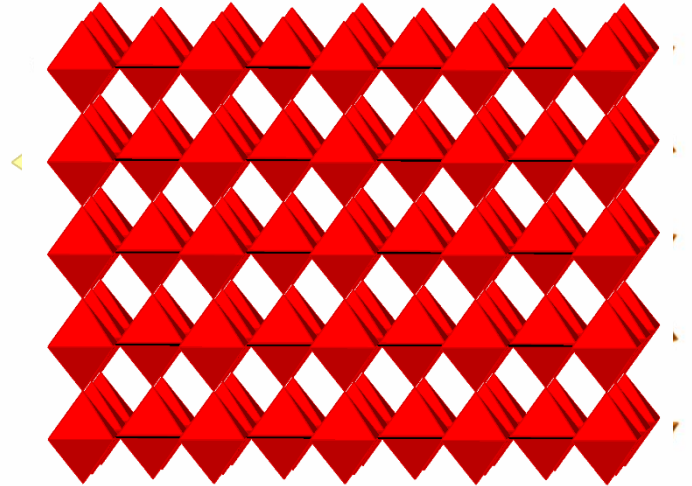


Tetragonal  $\text{I4/mmm}$



Perovskite-type  $\text{Pm}\bar{3}\text{m}$

Oxidizing gas ( $\text{O}_2$ ,  $\text{H}_2\text{O}$  damp, air,...)

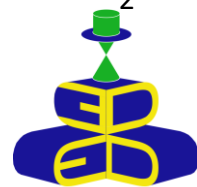
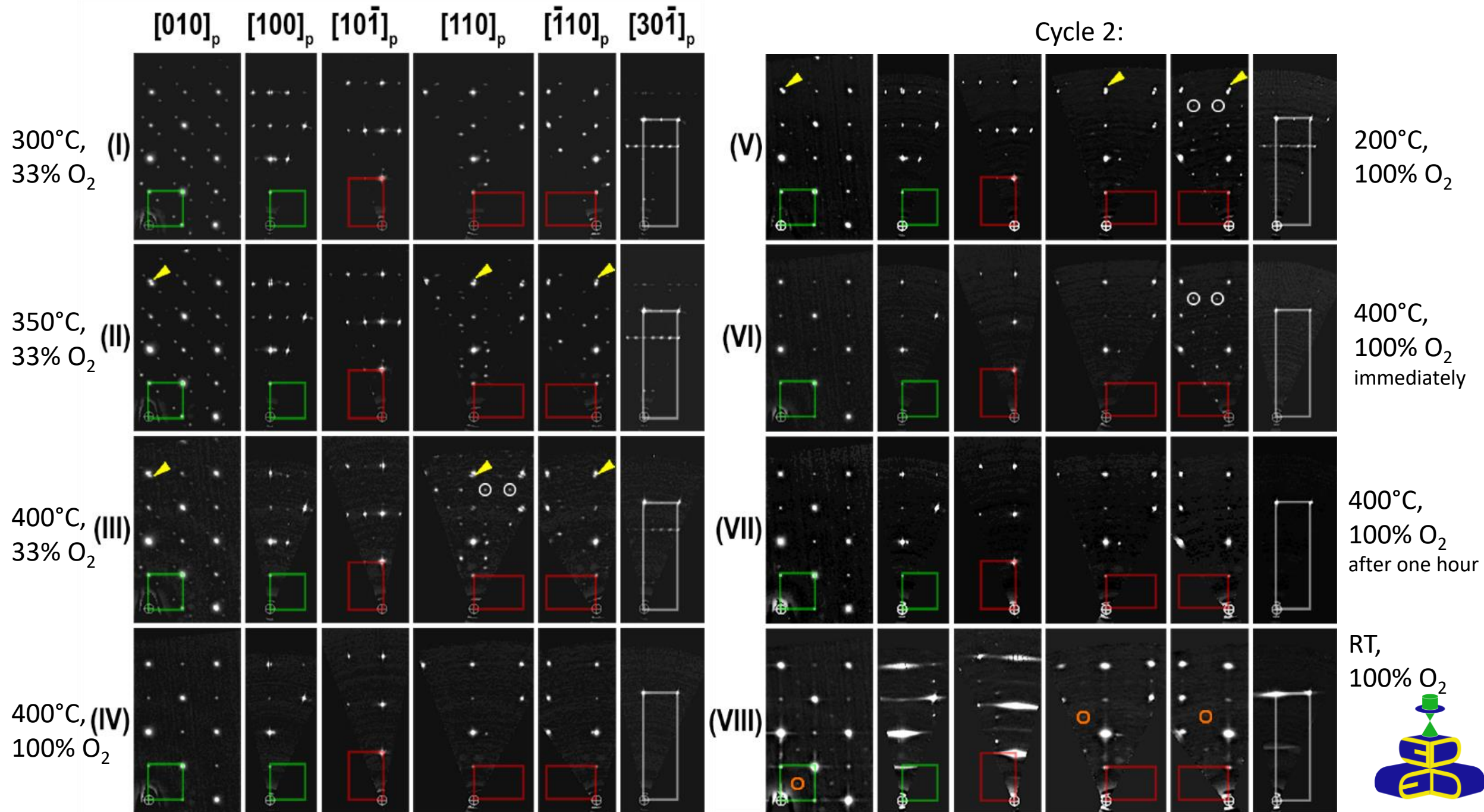


Crystal structures from

Hodges et al. JSSC 2000, 151, 190–209.

Maity et al. J. Phys. D: Appl. Phys. 2015, 48, 504004.





# ESR 3

- optimize the reaction uniformity among crystals for in situ 3D ED in a gas environment
- increase reproducibility related to temperature and gas pressures
- make merging possible
- optimize the practical experiment to allow detection of the best acquisition moment to have pure phase patterns

@ (1) reference sample, (2)  $\text{SrCoO}_{2.5}$  thin films



# Expected results

- (1) A working methodology for future acquisition of in situ crystallographic data.
- (2) Characterized structure evolution of relevant complex inorganic compounds during in situ reactions





# University of Antwerp



**EMAT**  
**Electron Microscopy for**  
**Materials Research**





# Electron Microscopy lab

- Atomic resolution imaging and spectroscopy (HRTEM, HAADF/ABF-STEM, iDPC, EELS, EDX)
- Electron tomography and electron diffraction tomography (3DED)
- 6 TEMs, including 3 state-of-the-art aberration corrected Titan instruments
- specialized holders for various in situ experiments
- dedicated sample preparation facilities



# Electron Crystallography group

- Focus on crystal structure determination
- 3DED
- Diffuse scattering, defects, SRO
- In situ structure evolution in gas-solid reactions
- Atomic resolution imaging and spectroscopy

