

Investigation of local structures of organic compounds using pair-distribution function analysis

Prof. Dr. Martin U. Schmidt

Institut für Anorganische und Analytische Chemie

Goethe-Universität

Frankfurt am Main

m.schmidt@chemie.uni-frankfurt.de

Phone +49 69 798 29171

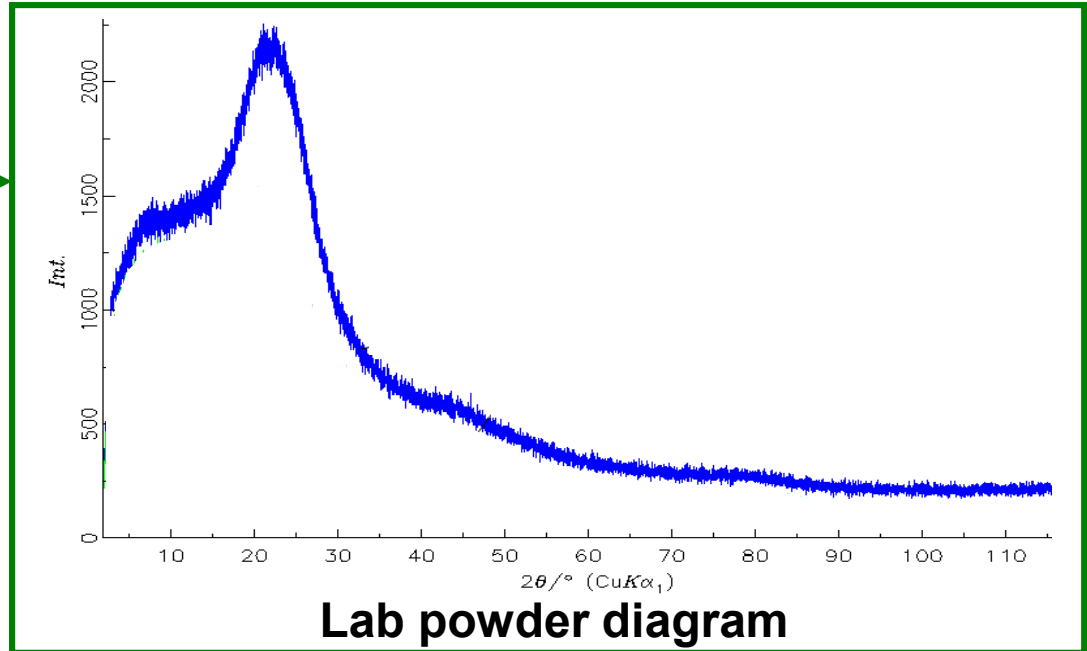
= Confidential. Not to be published =

Amorphous compounds: The dream of a crystallographer

**Amorphous
pharmaceutical
powder**

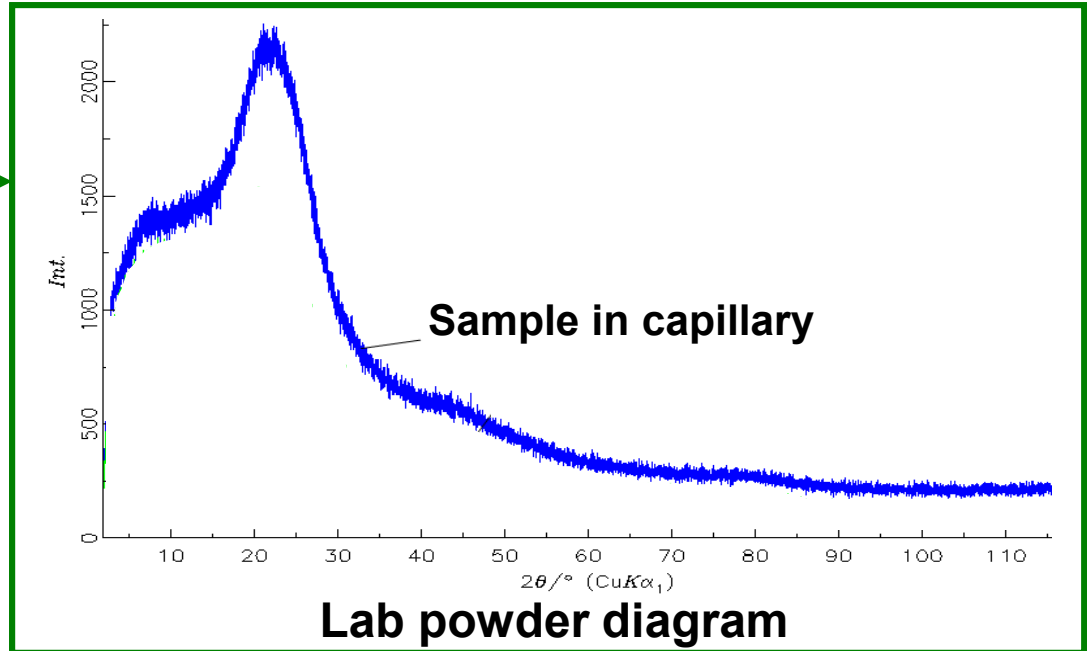
Amorphous compounds: The dream of a crystallographer

Amorphous
pharmaceutical
powder



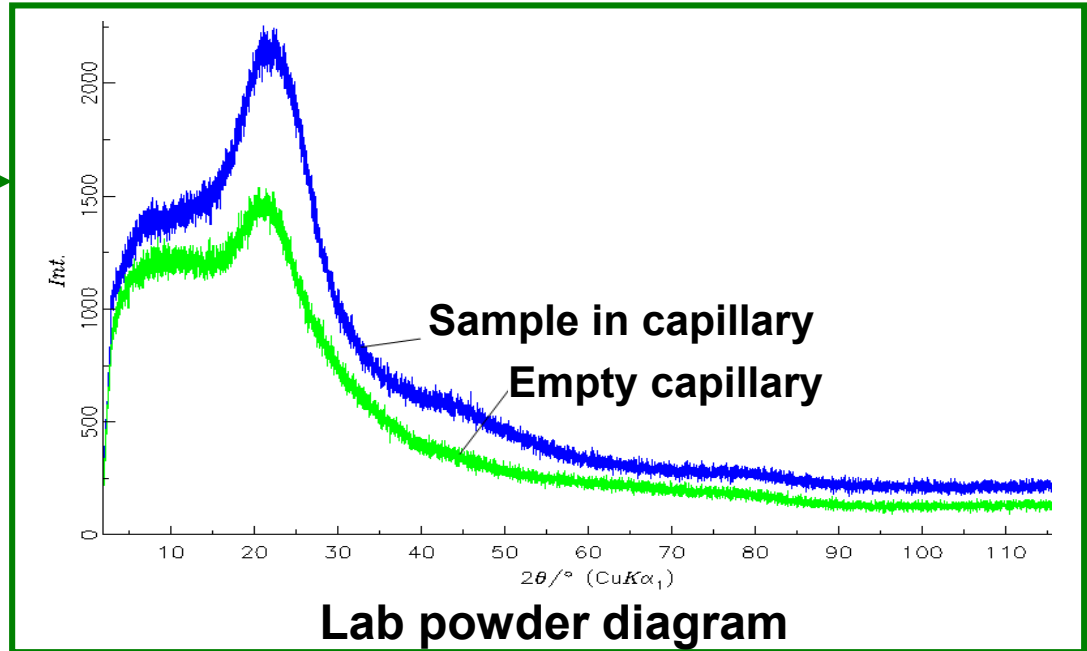
Amorphous compounds: The dream of a crystallographer

Amorphous
pharmaceutical
powder



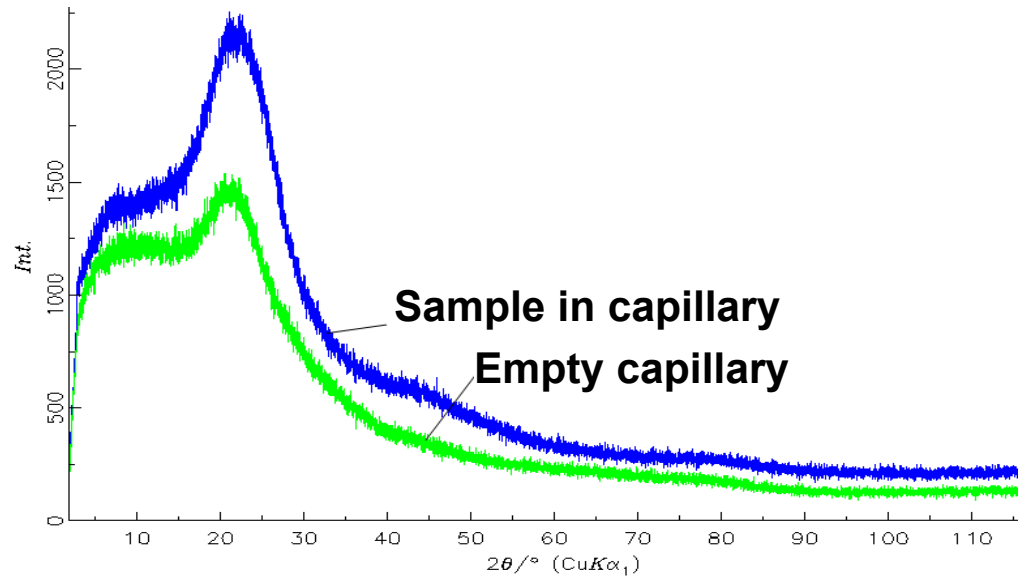
Amorphous compounds: The dream of a crystallographer

Amorphous
pharmaceutical
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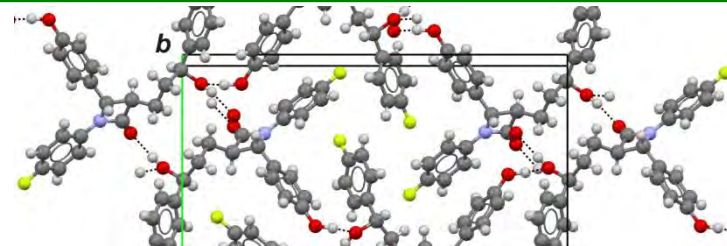


Amorphous compounds: The dream of a crystallographer

Amorphous
pharmaceutical
powder



Lab powder diagram

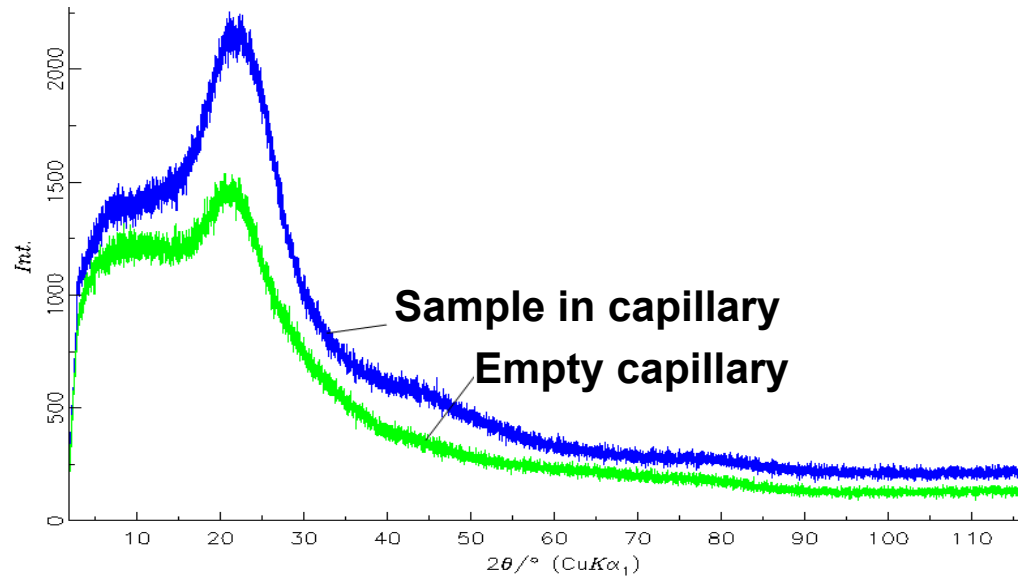


Crystal structure

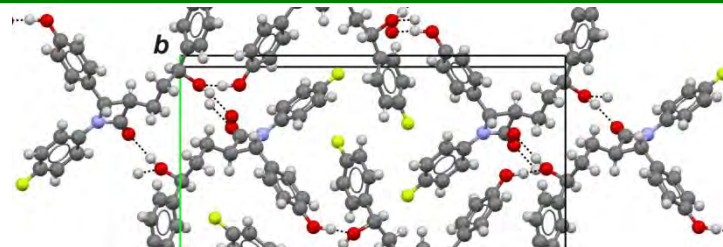
Amorphous compounds: The dream of a crystallographer

Amorphous
pharmaceutical
powder

*Here occurs
a miracle*



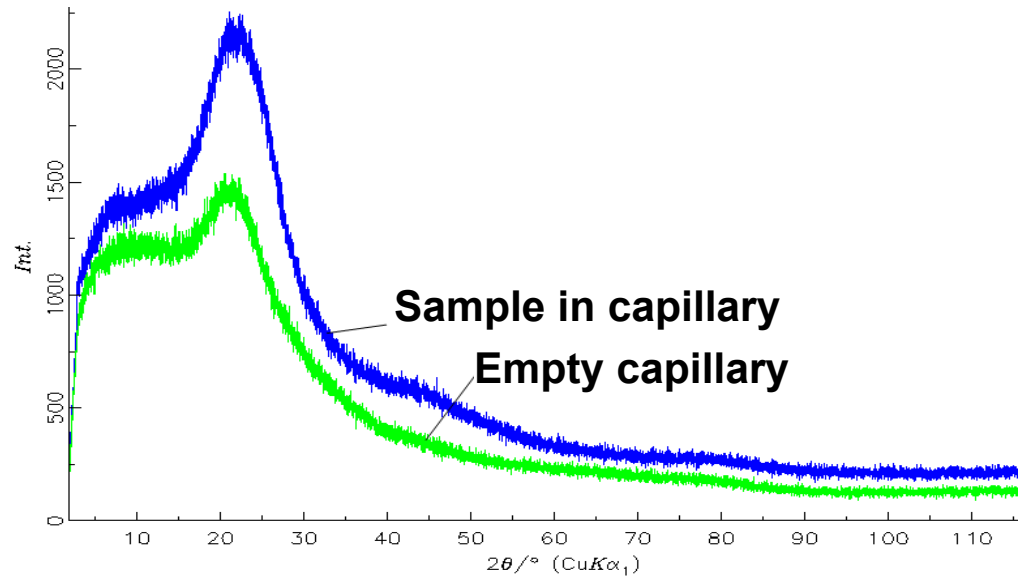
Lab powder diagram



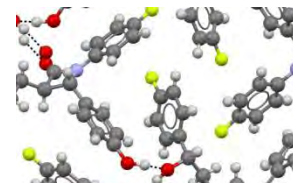
Crystal structure

Amorphous compounds: What is possible?

Amorphous
pharmaceutical
powder

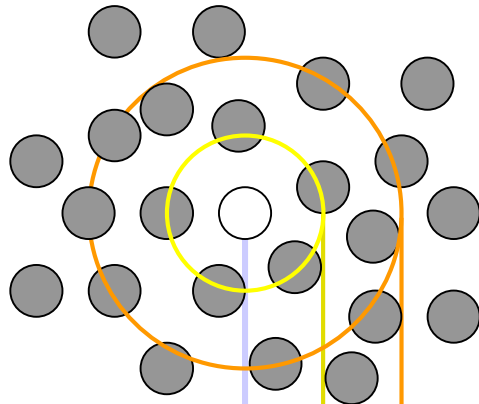


Pair-distribution
function analysis



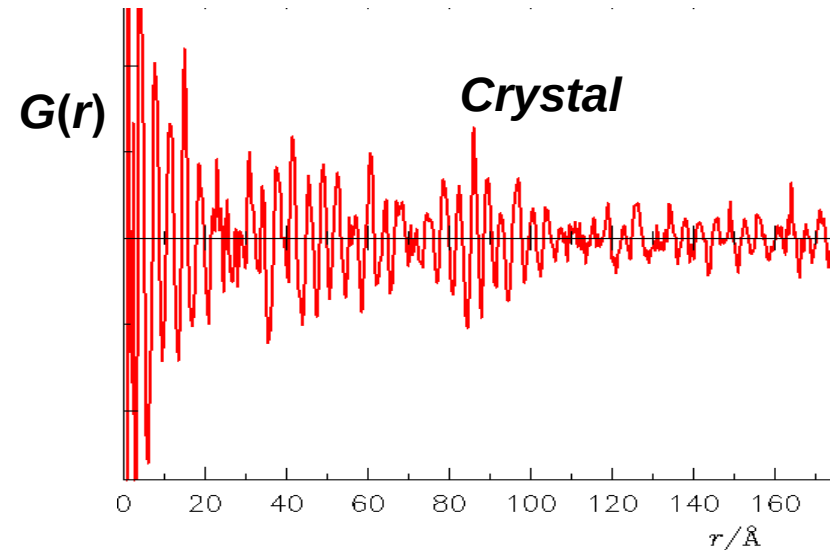
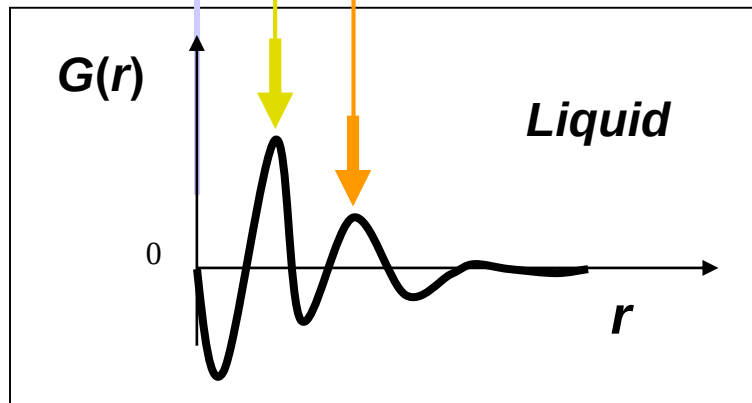
Information on the local structure

Pair Distribution Function (PDF)



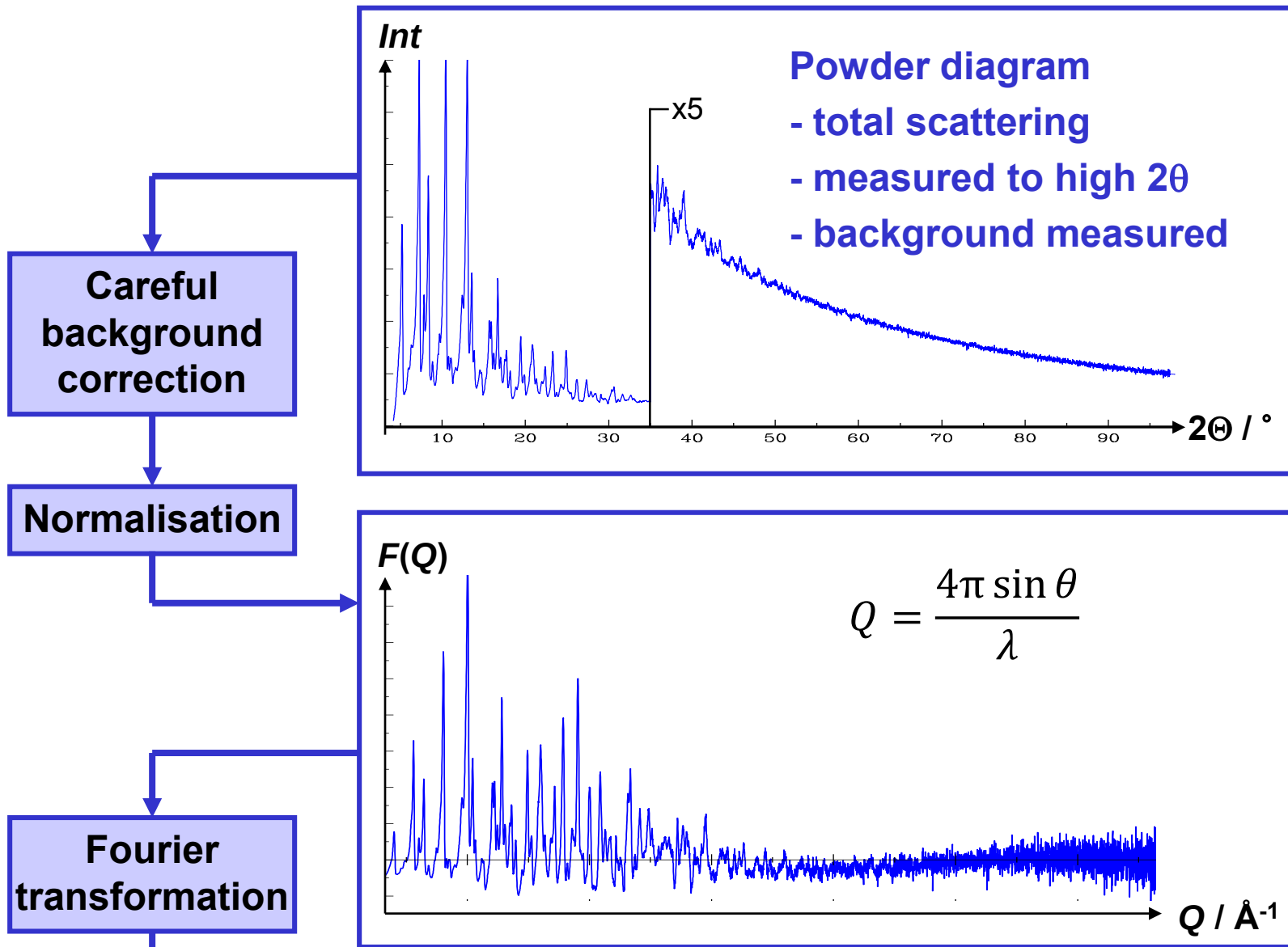
PDF, $G(r)$:

- Probability to find two atoms with a distance r**
- weighted with the scattering power**
- summed over all atoms**
- normalised to a homogeneous atom density**
- similar to the "radial distribution function"**



$$G(r) = 4\pi r[\rho(r) - \rho_0] = \frac{2}{\pi} \int_{\min}^{\max} Q[S(Q) - 1] \sin(Qr) dQ$$

How to get the PDF

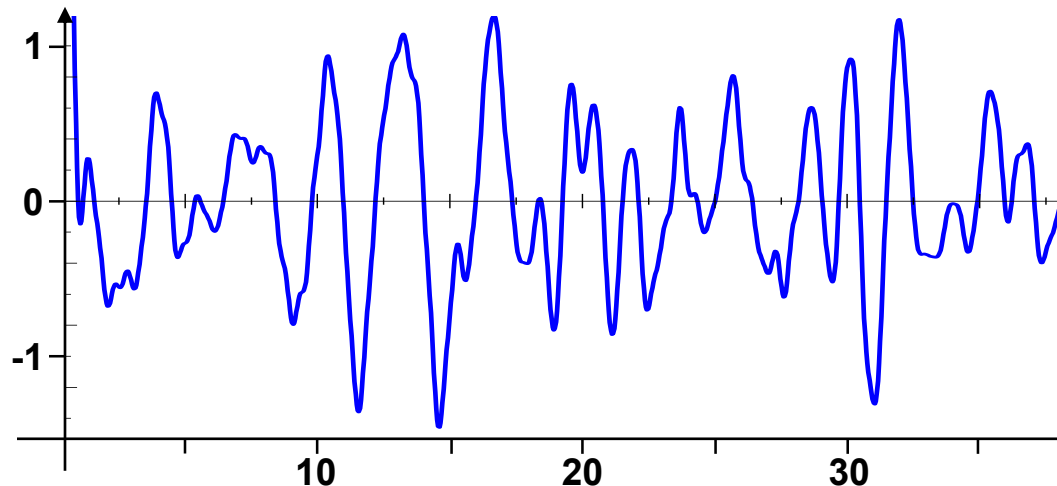


How to get the PDF

$$G(r) = 4\pi r[\rho(r) - \rho_0] = \frac{2}{\pi} \int_{\min}^{\max} Q[S(Q) - 1] \sin(Qr) dQ$$

Pair distribution function

Probability
 $G(r) / \text{\AA}^{-2}$



Pair Distribution Function

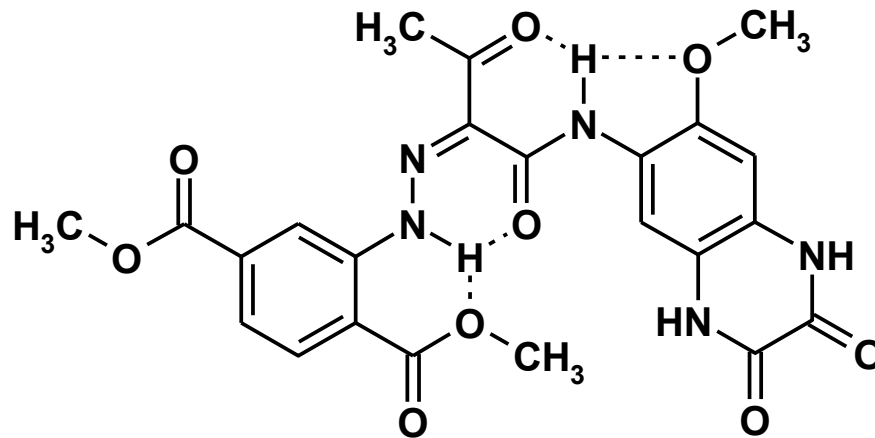
Classical applications:

Local structures of

- glasses**
- liquids**
- nanocrystalline inorganic compounds**
- amorphous inorganic compounds**
- quasicrystals**
- disordered crystals**

New: PDF on nanocrystalline and amorphous organic compounds

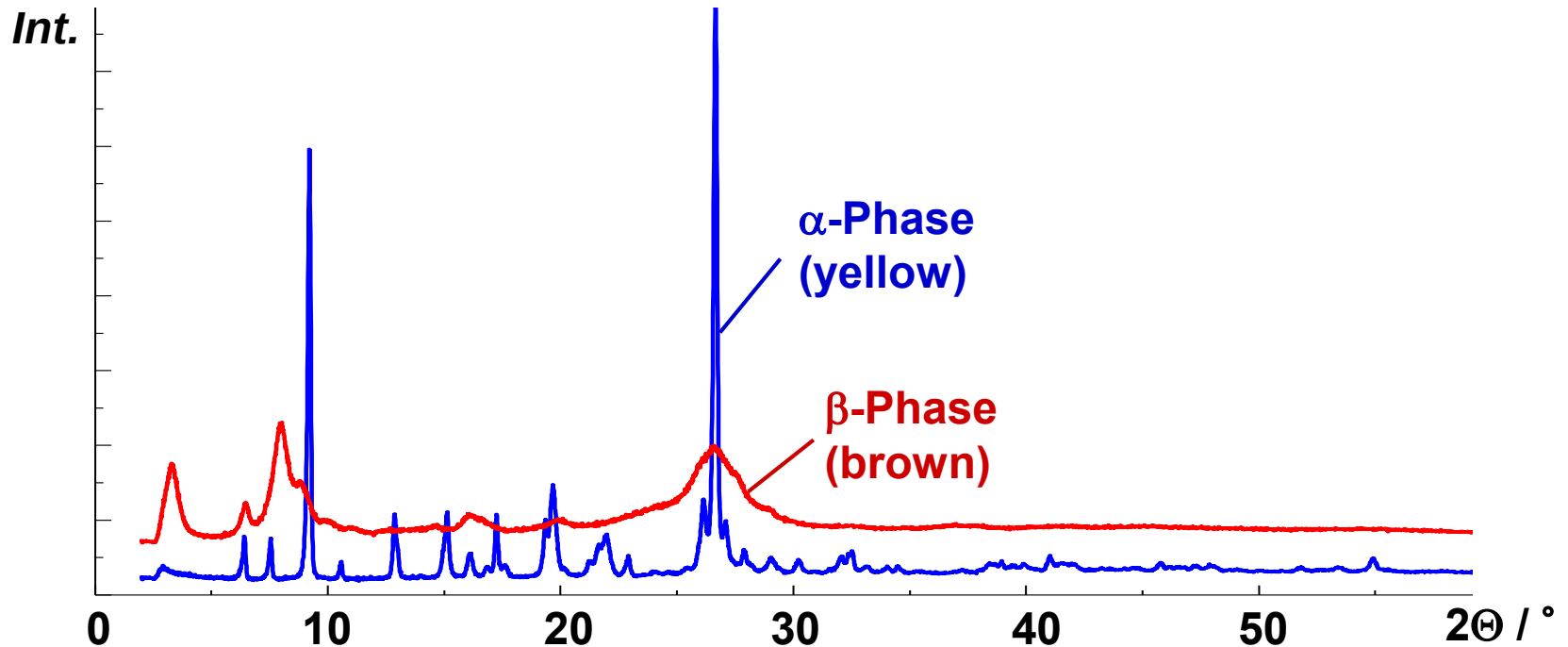
PDF on a nanocrystalline pigment



Pigment Yellow 213
(PY 213)



PDF on a nanocrystalline pigment



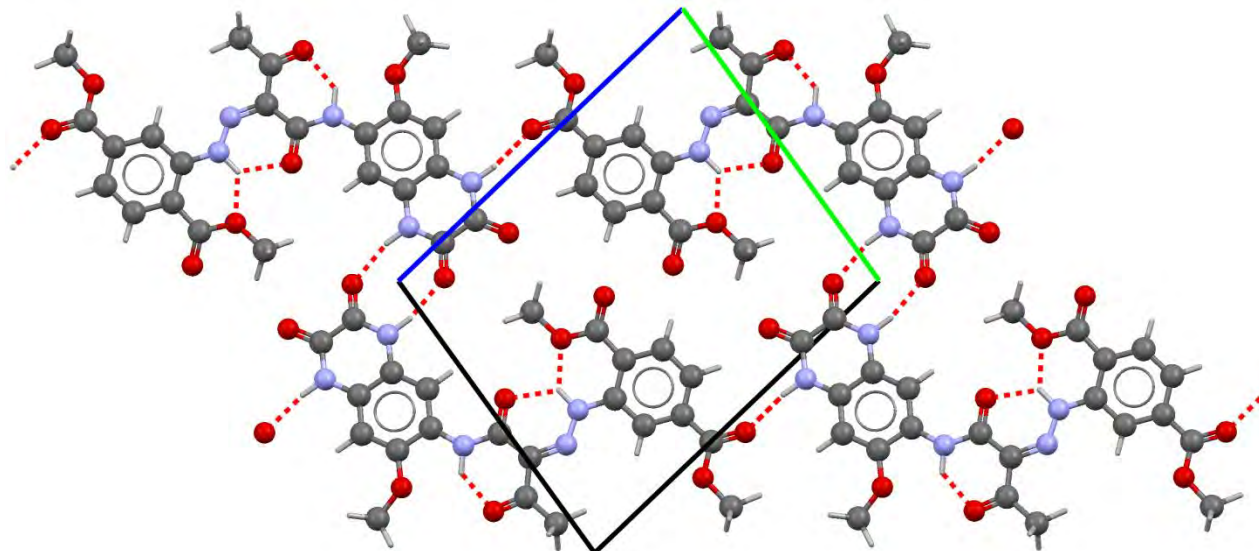
α -Phase:

Structure determined by combination of electron diffraction and X-ray powder diffraction

M. U. Schmidt, S. Brühne, A. K. Wolf, A. Rech, J. Brüning, E. Alig, L. Fink, C. Buchsbaum, J. Glinemann, J. van de Streek, F. Gozzo, M. Brunelli, F. Stowasser, T. Gorelik, E. Mugnaioli, U. Kolb, *Acta Cryst. B* **2009**, 65, 189-199.

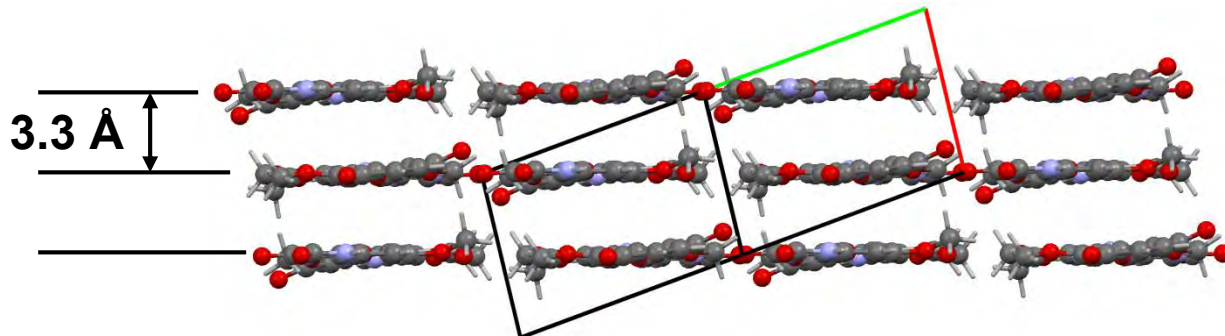
PDF on a nanocrystalline pigment

$a = 6.9006(3) \text{ \AA}$
 $b = 11.8347(6) \text{ \AA}$
 $c = 14.0592(7) \text{ \AA}$
 $\alpha = 81.811(4)^\circ$
 $\beta = 81.032(9)^\circ$
 $\gamma = 87.541(10)^\circ$
 $V = 1122.3(1) \text{ \AA}^3$
 $P\bar{1}, Z = 2$



View direction [100]

Layer
structure



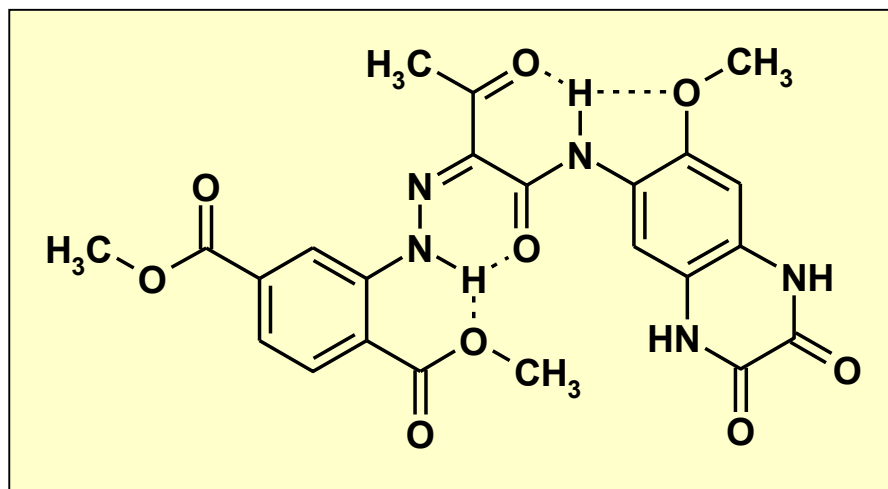
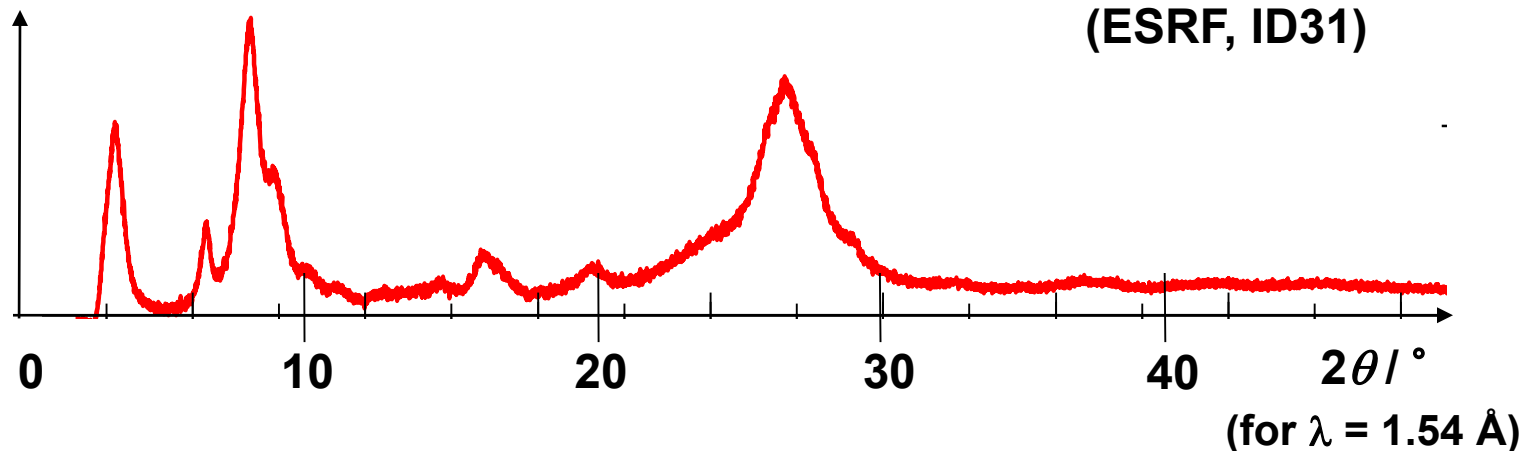
View direction [011]

M. U. Schmidt, S. Brühne, A. K. Wolf, A. Rech, J. Brüning, E. Alig, L. Fink, C. Buchsbaum, J. Glinemann, J. van de Streek, F. Gozzo, M. Brunelli, F. Stowasser, T. Gorelik, E. Mugnaioli, U. Kolb, *Acta Cryst. B* **2009**, 65, 189-199.

PDF on a nanocrystalline pigment

β -Phase: Best sample ever obtained

Synchrotron data
(ESRF, ID31)

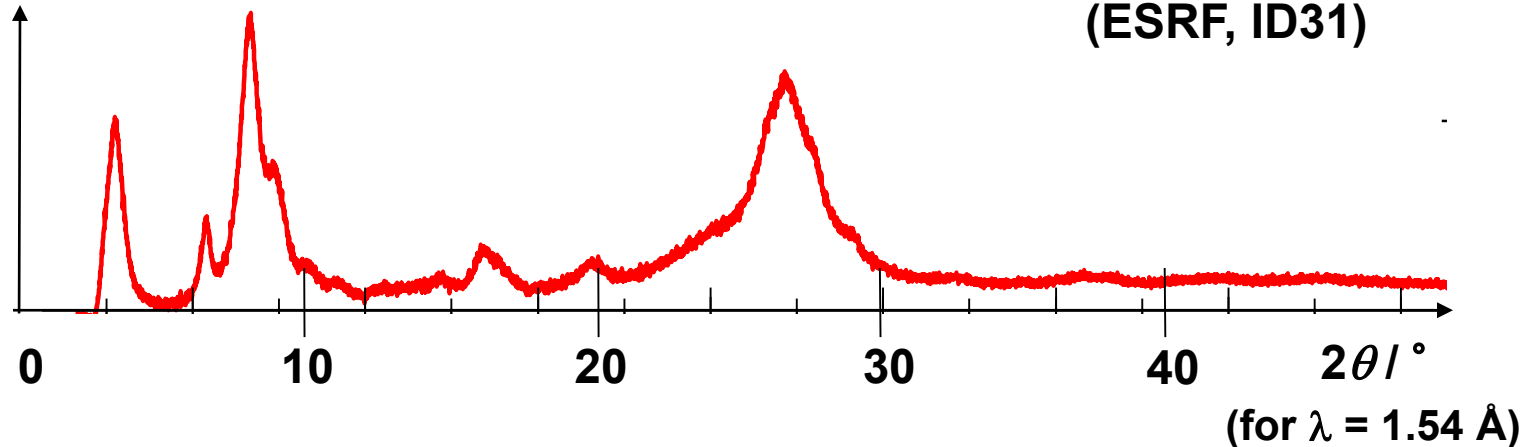


M. U. Schmidt, S. Brühne, A. K. Wolf, A. Rech, J. Brüning, E. Alig, L. Fink, C. Buchsbaum, J. Glinemann, J. van de Streek, F. Gozzo, M. Brunelli, F. Stowasser, T. Gorelik, E. Mugnaioli, U. Kolb, *Acta Cryst. B* **2009**, 65, 189-199.

PDF on a nanocrystalline pigment

β -Phase: Best sample ever obtained

Synchrotron data
(ESRF, ID31)



Does this diagram contain any usefull information?

Yes!

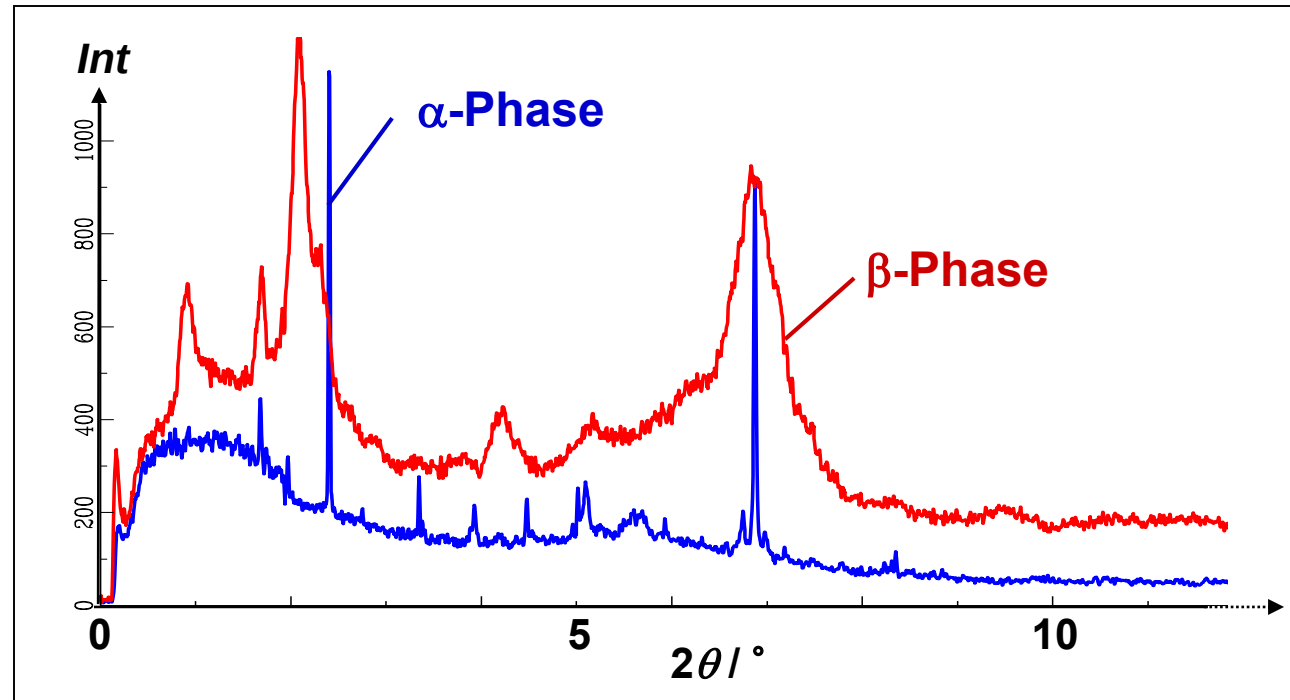
Use Pair Distribution Function Analysis (PDF)!

M. U. Schmidt, S. Brühne, A. K. Wolf, A. Rech, J. Brüning, E. Alig, L. Fink, C. Buchsbaum, J. Glinemann, J. van de Streek, F. Gozzo, M. Brunelli, F. Stowasser, T. Gorelik, E. Mugnaioli, U. Kolb, *Acta Cryst. B* **2009**, 65, 189-199.

PDF on a nanocrystalline pigment

Synchrotron data
(ESRF, ID31):

- $\lambda = 0.40 \text{ \AA}$
- $2\theta(\text{max}) = 120^\circ$
- longer counting time
at high 2θ

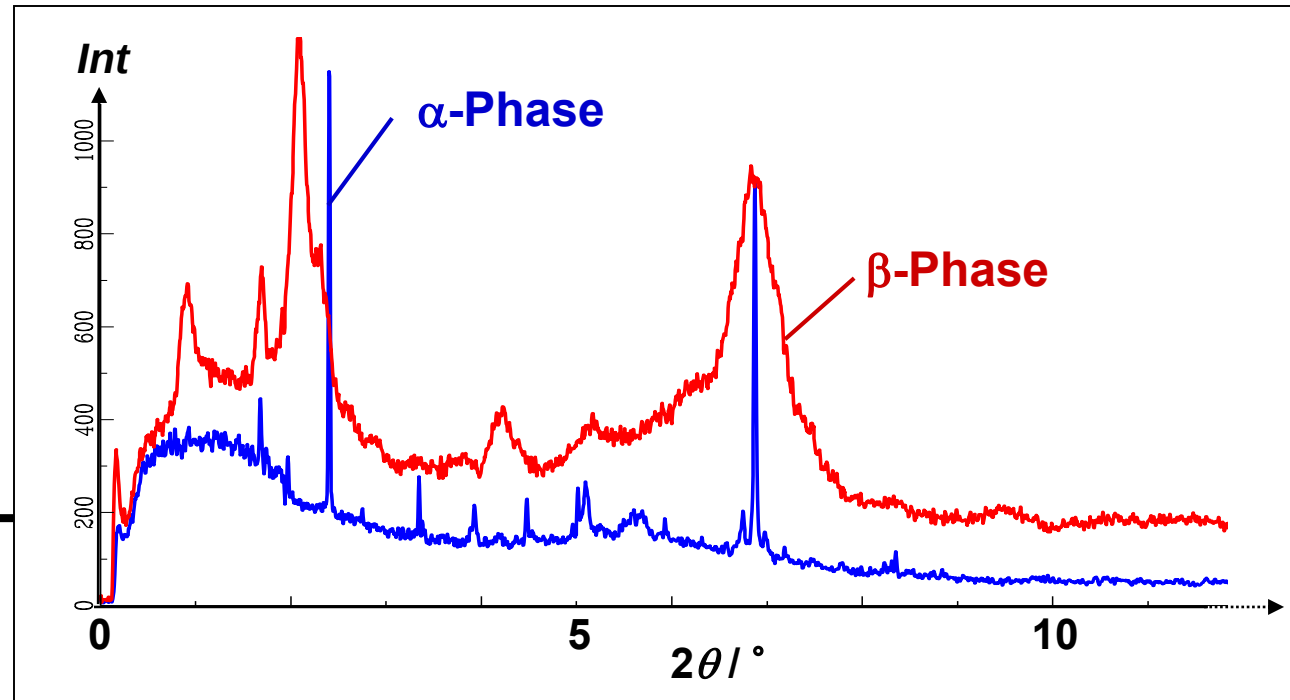


M. U. Schmidt, S. Brühne, A. K. Wolf, A. Rech, J. Brüning, E. Alig, L. Fink, C. Buchsbaum, J. Glinemann, J. van de Streek, F. Gozzo, M. Brunelli, F. Stowasser, T. Gorelik, E. Mugnaioli, U. Kolb, *Acta Cryst. B* **2009**, 65, 189-199.

PDF on a nanocrystalline pigment

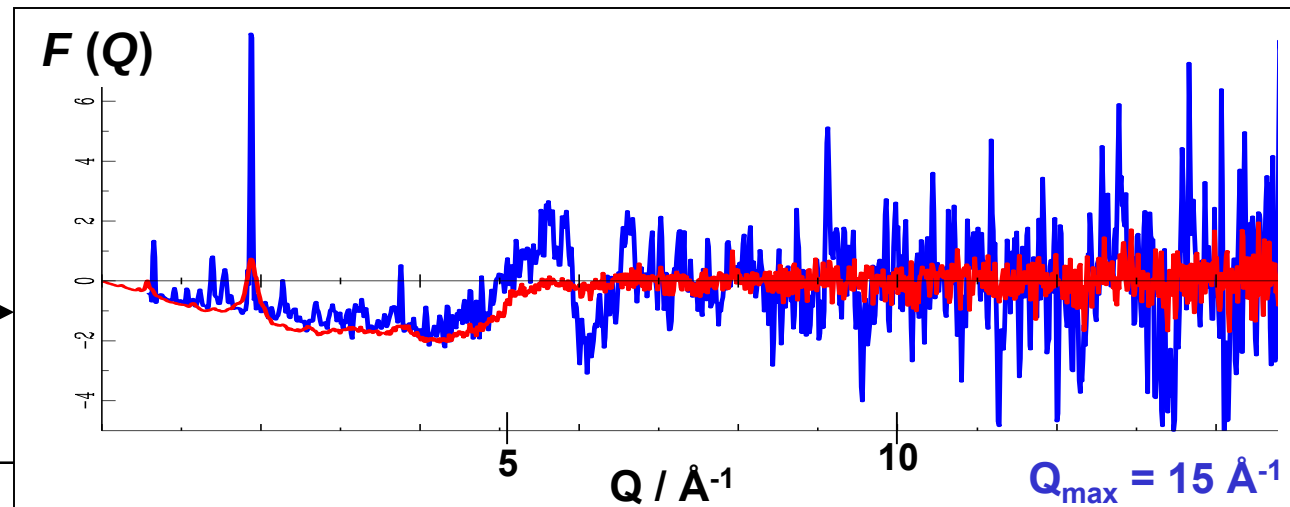
Synchrotron data
(ESRF, ID31):

- $\lambda = 0.40 \text{ \AA}$
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at high 2θ

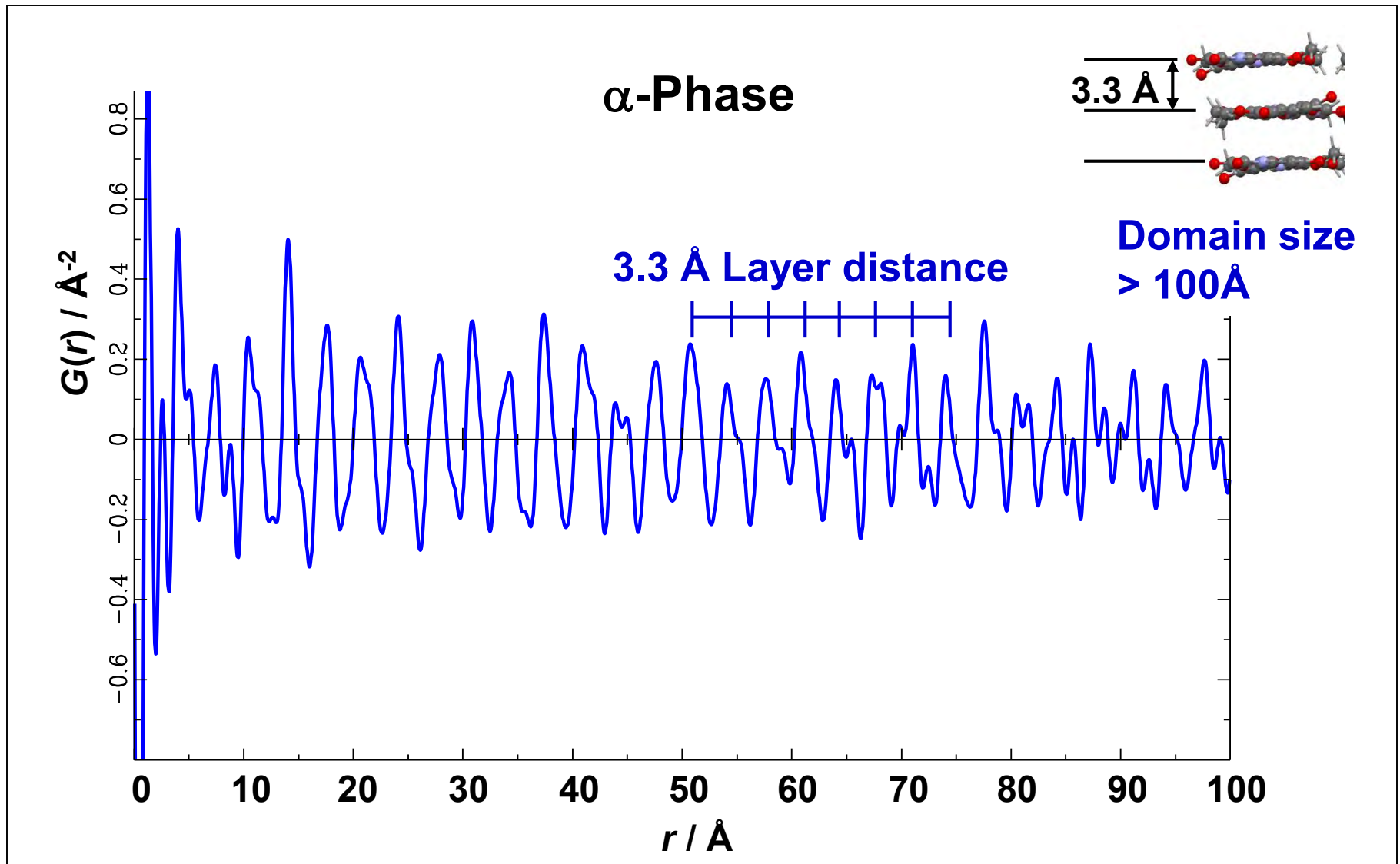


Carefull
background
correction

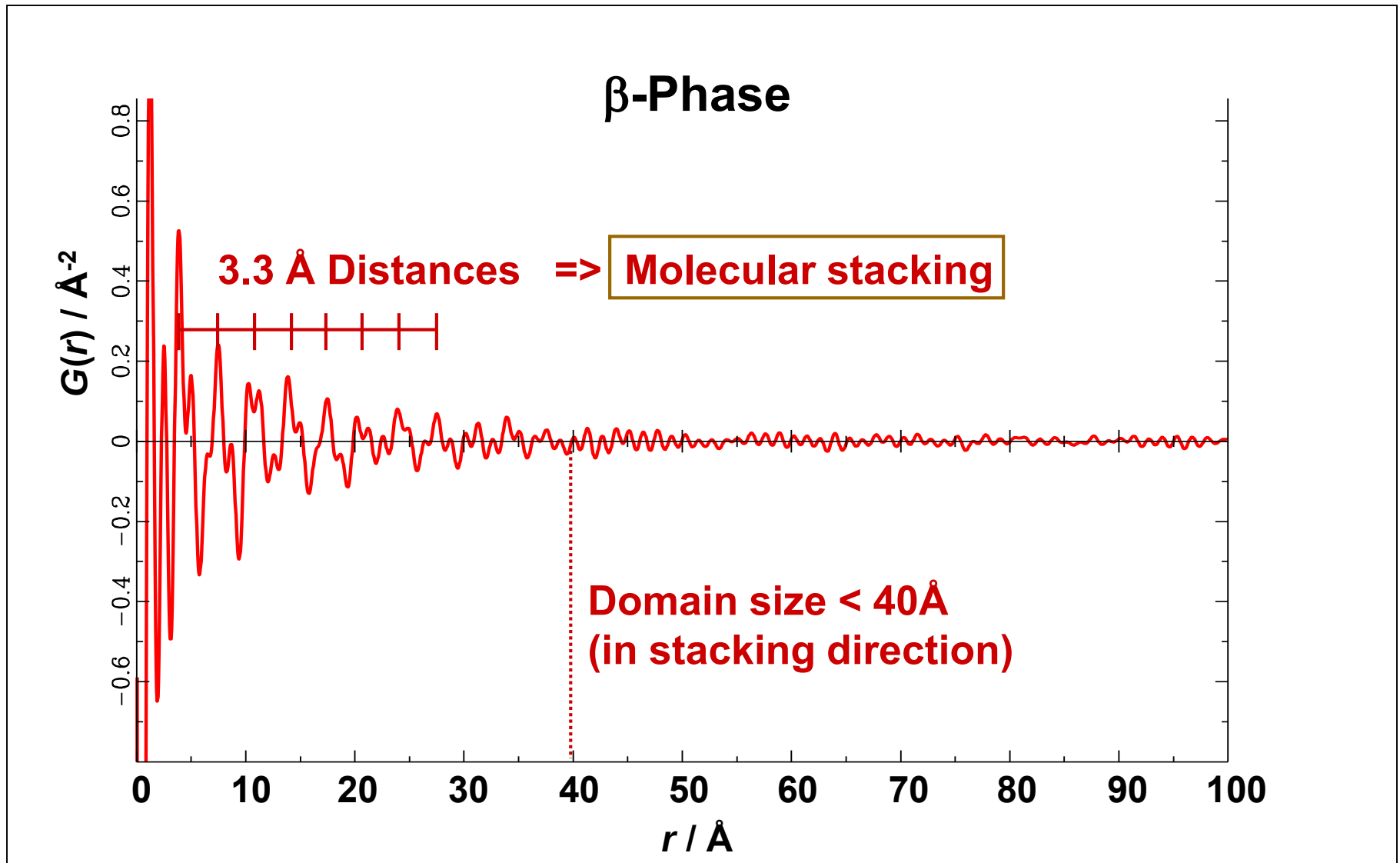
Normalisation



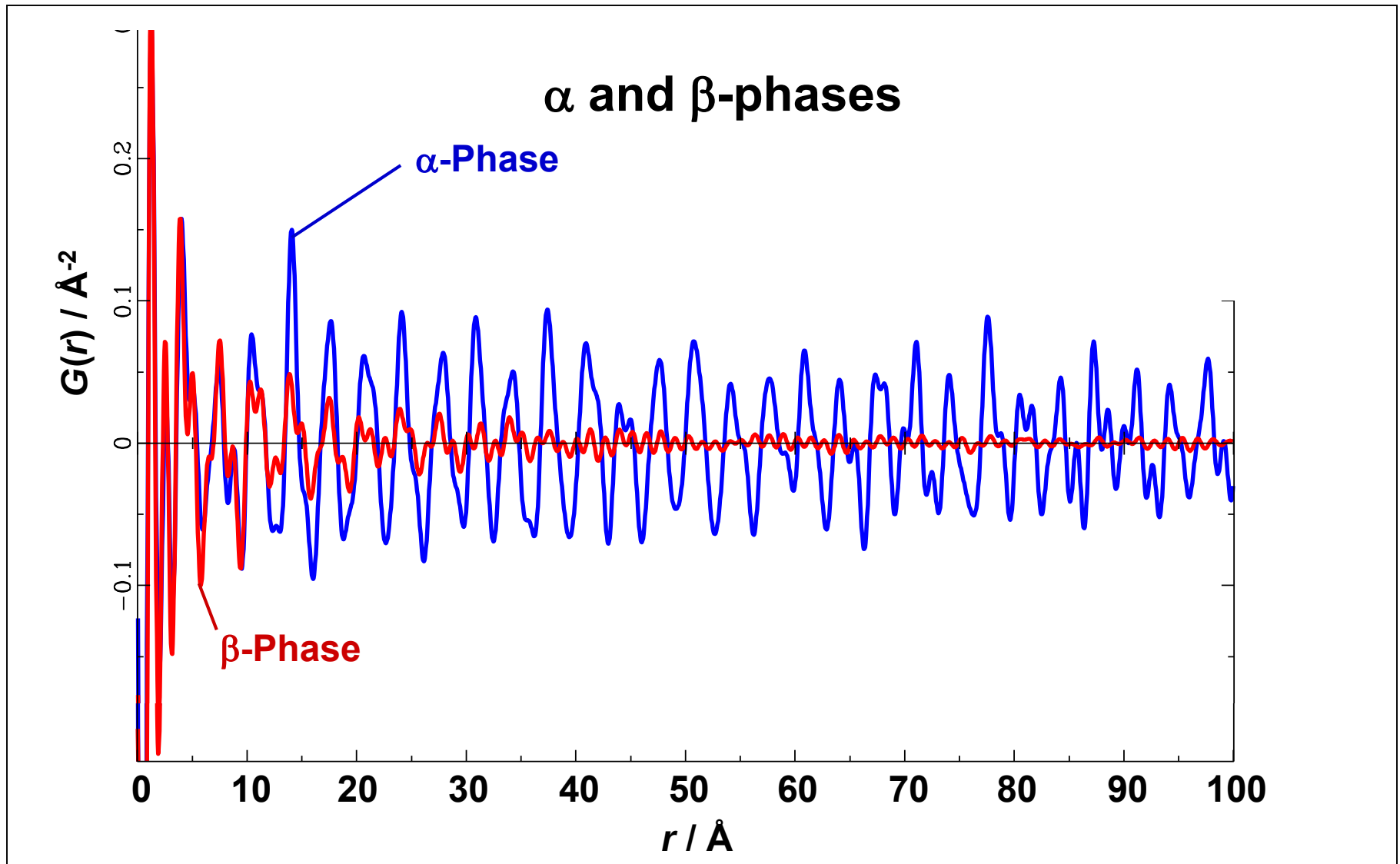
PDF on a nanocrystalline pigment



PDF on a nanocrystalline pigment

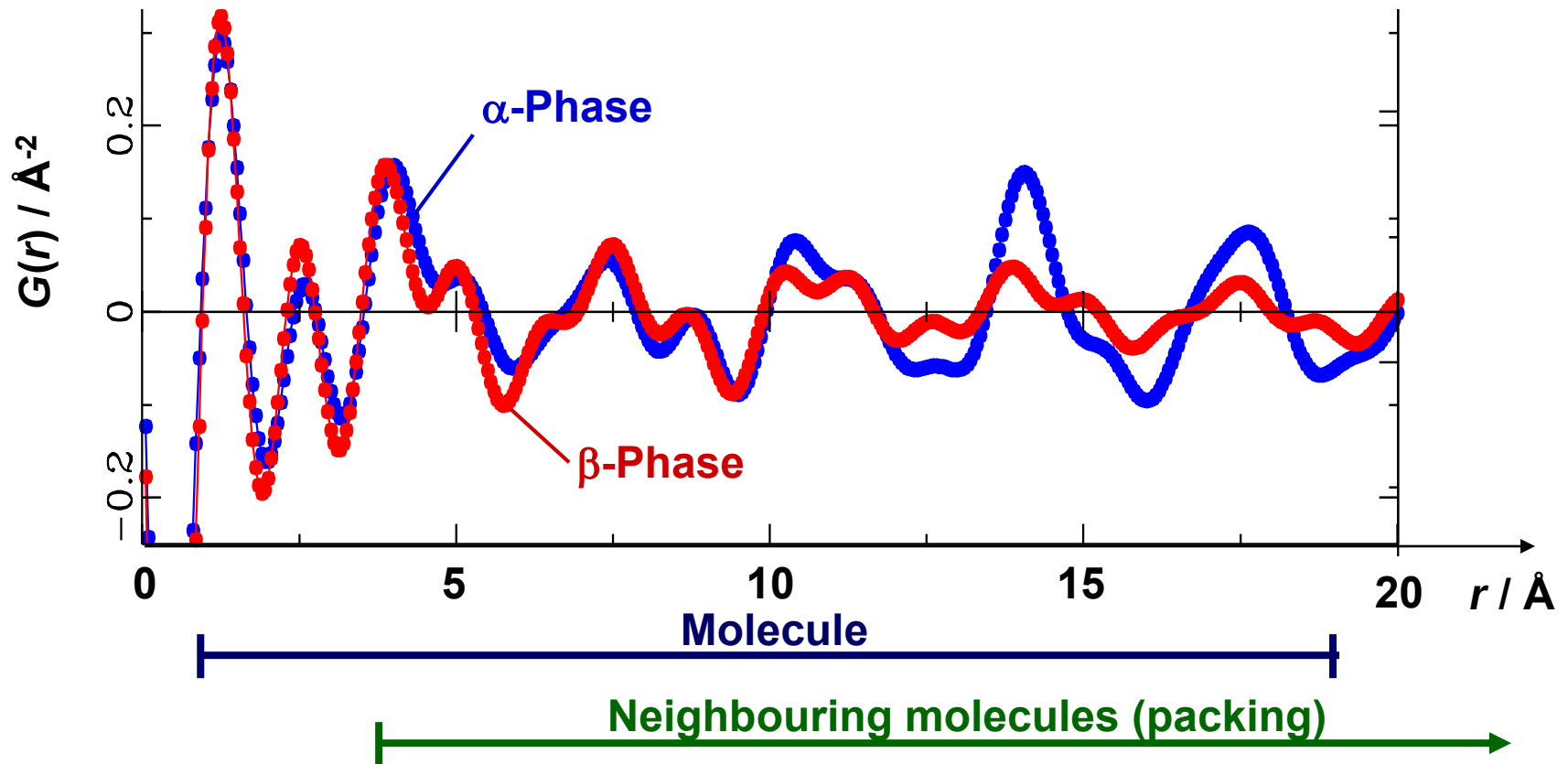


PDF on a nanocrystalline pigment



PDF on a nanocrystalline pigment

α - and β -phases



=> Local structures of α - and β -phases similar

General remarks on PDF of organic compounds

Problems with organic compounds:

1) Only light atoms

- Weak scattering
- Bad signal-to-noise ratio at high 2θ angles
- Longer measurement times required, esp. at high 2θ
- Nevertheless (esp. with lab data):
Frequently too much noise at high 2θ
=> Truncation at high 2θ necessary => Limited $Q_{\max}$

2) Sample in capillary or between polymer films

- Measurement of an empty capillary or empty sample holder
- Always careful background correction required
- Scaling of background vs. sample ambiguous

General remarks on PDF of organic compounds

Problems with organic compounds (continued):

3) Radiation damage

- May cause amorphisation or decomposition

The PDF of organic compounds strongly depends on:

- The sample
- The measurement
- The program used for calculating the PDF
- Background correction
- Parameters in the Fourier transformation,
e.g. $Q_{\min}$, $Q_{\max}$, various corrections, ...

The PDF strongly depends on the user and his/her experience

General remarks on PDF of organic compounds

Caution with artefacts!

Origin of artefacts, e.g.:

- noise**
- data corrections**
- Fourier transformation**

Artefacts are frequent at:

- $r = 0 - 2 \text{ \AA}$ (suggesting unusual atom-atom distances)**
- high r , e.g. sine waves in the PDF**

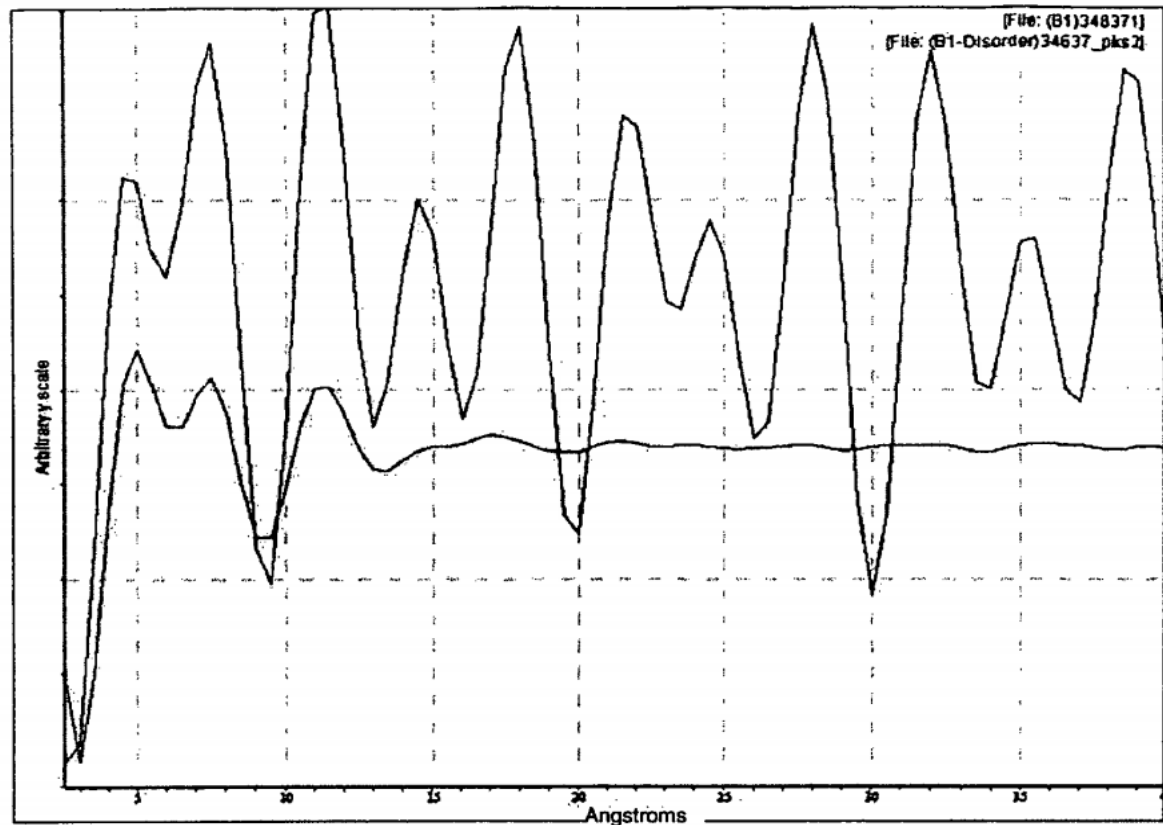
General remarks on PDF of organic compounds

Caution with artefacts!

The literature contains many examples with

- unreliable data

PDF from
Cu radiation
with $2\theta_{\max} = 40^\circ$
[Patent, 2005]

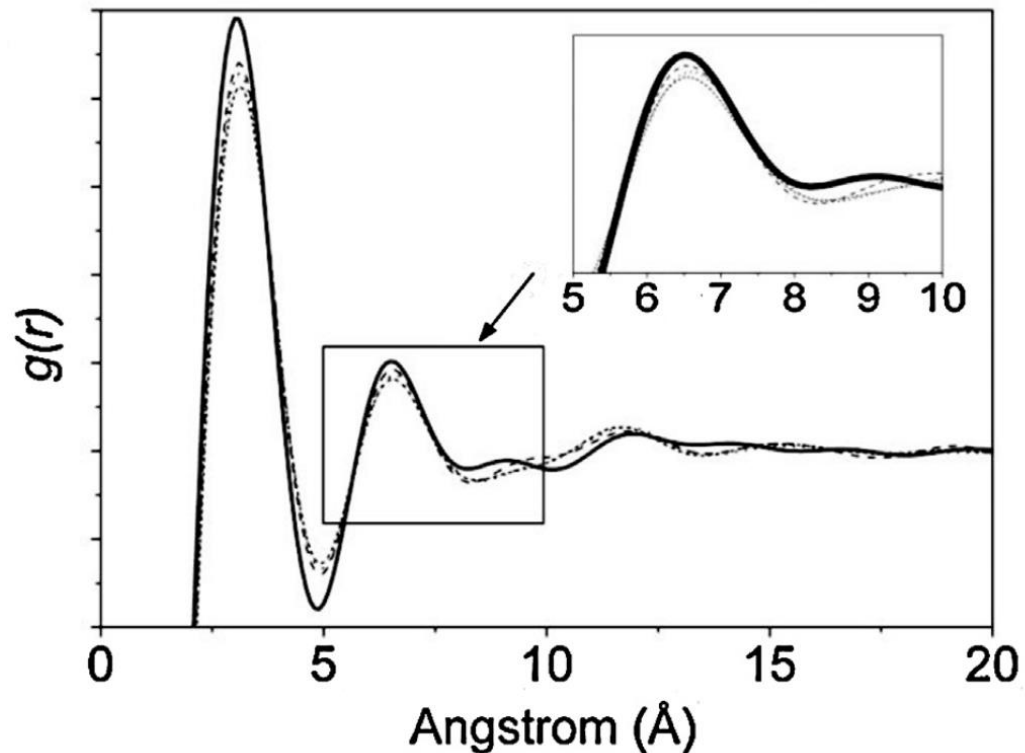


General remarks on PDF of organic compounds

Caution with artefacts!

The literature contains many examples with

- unreliable data



PDF from
Cu radiation
with $2\theta_{\text{max}} = 35^\circ$

[*Int. J. Pharm.*, 2013]

General remarks on PDF of organic compounds

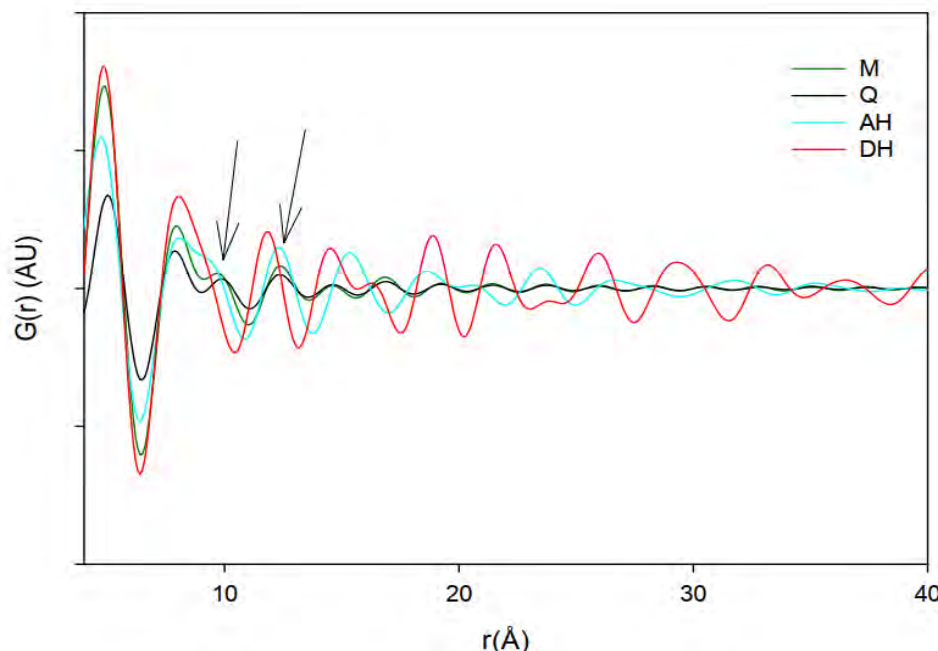
Caution with artefacts!

The literature contains many examples with

- unreliable data
- unreliable data processing
(e.g. no background correction,
generation of artefacts, ...)

**Interpretation of PDFs
from Cu radiation
with $2\theta_{\max} = 40^\circ$**

[Pharmaceutics, 2012]



General remarks on PDF of organic compounds

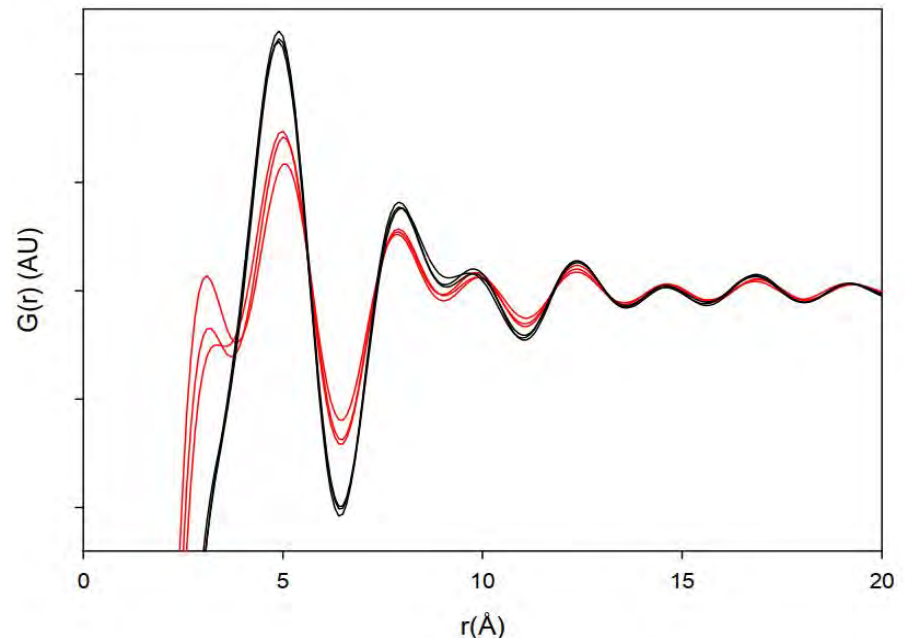
Caution with artefacts!

The literature contains many examples with

- unreliable data
- unreliable data processing
(e.g. no background correction,
generation of artefacts, ...)
- Interpretation of artefacts
(**very frequent**)

Interpretation of PDFs
from Cu radiation
with $2\theta_{\text{max}} = 40^\circ$

[Pharmaceutics, 2012]



General remarks on PDF of organic compounds

My recommendations for PDF of organic compounds:

- **Sample holder: capillary or between films: both possible**
- **2θ as high as possible**
- **Background measurement**
(of course measured under the same conditions as the sample)
- **Longer counting time at high 2θ (especially on lab instruments)**
- **Good signal-to-noise ratio**
On lab diffractometers: long measurement times
- **Careful corrections and Fourier transform by experienced people**
- **No interpretation of distances less than ca. 2Å**
- **Lab data are sufficient, if PDF is used as "finger print",**
(e.g. phase analysis, domain sizes, crystallinity,
comparison of local structures, ...)
- **Structure refinement by PDF fit: Synchrotron data required**

Nanocrystalline and amorphous API

3 samples:

- Crystalline, unmilled**
- Nanomilled**
- Amorphous**

Nanocrystalline and amorphous API

3 samples:

- Crystalline, unmilled**
- Nanomilled**
- Amorphous**

All measurements:

STOE Stadi-P diffractometer

Ge(111) monochromator

Linear PSD

Cu- $K\alpha_1$ radiation

Capillary

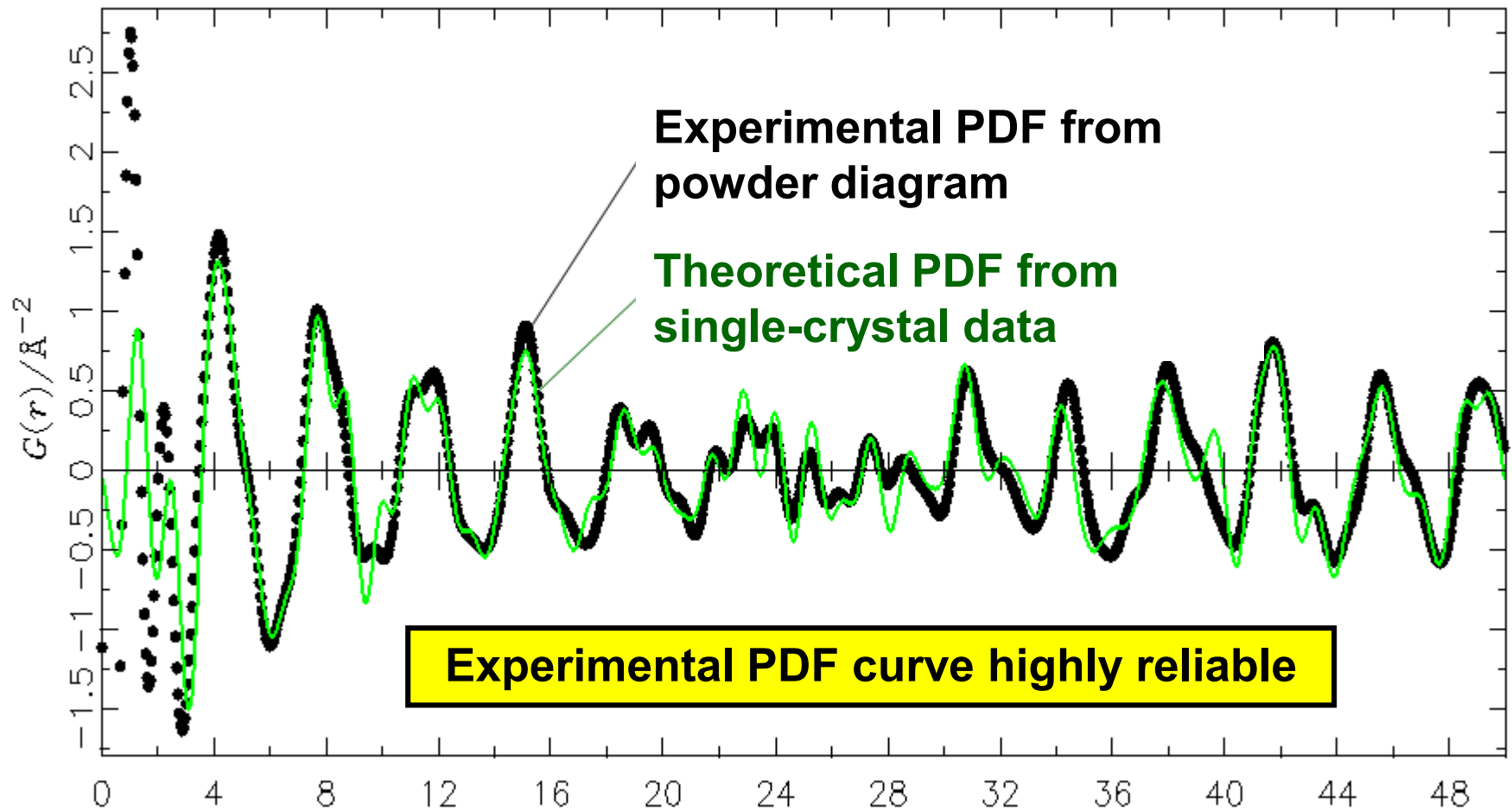
Transmission mode

$2\theta = 2 - 120^\circ$

Room temperature

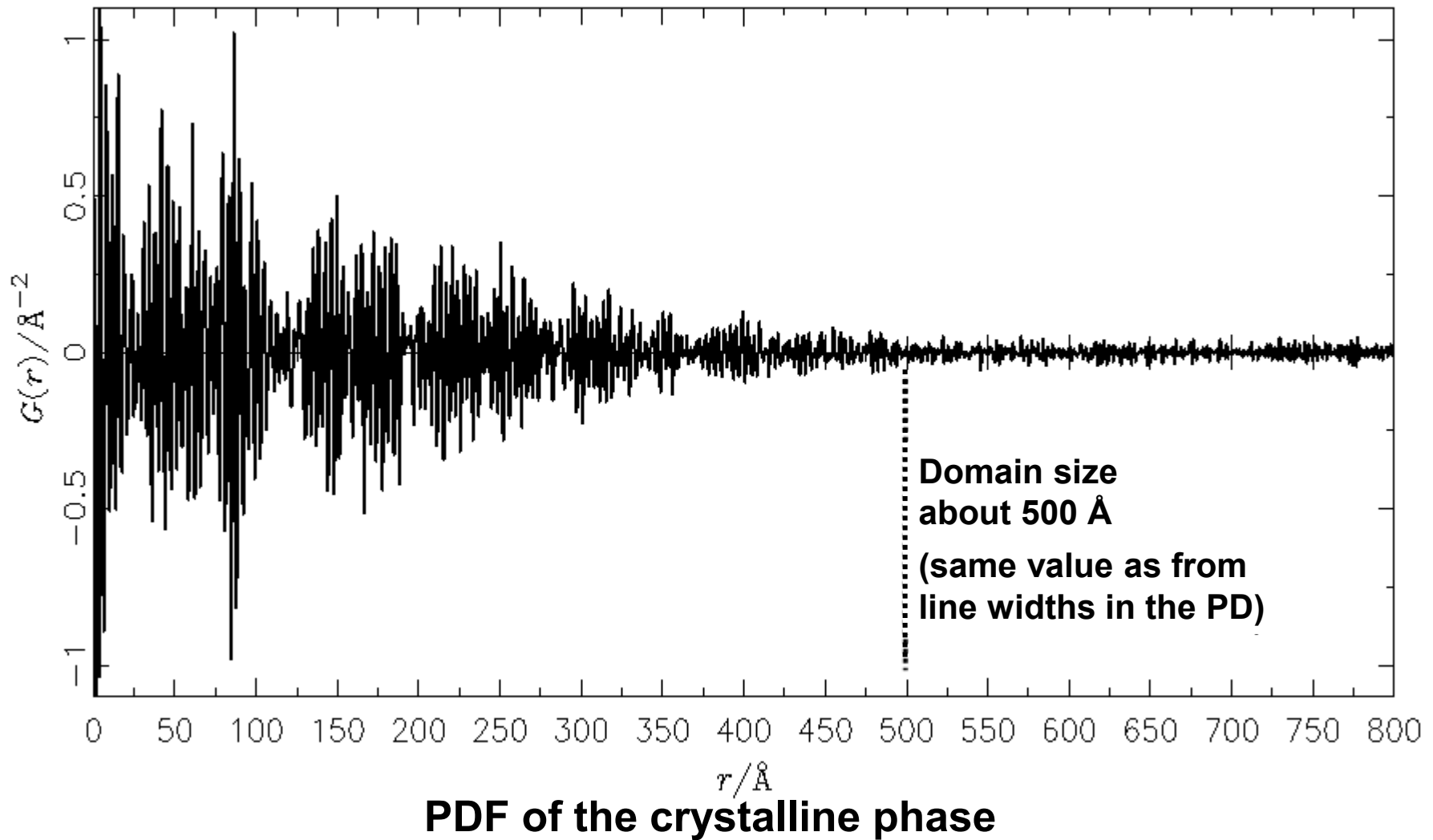
**Background correction:
empty capillary**

Nanocrystalline and amorphous API

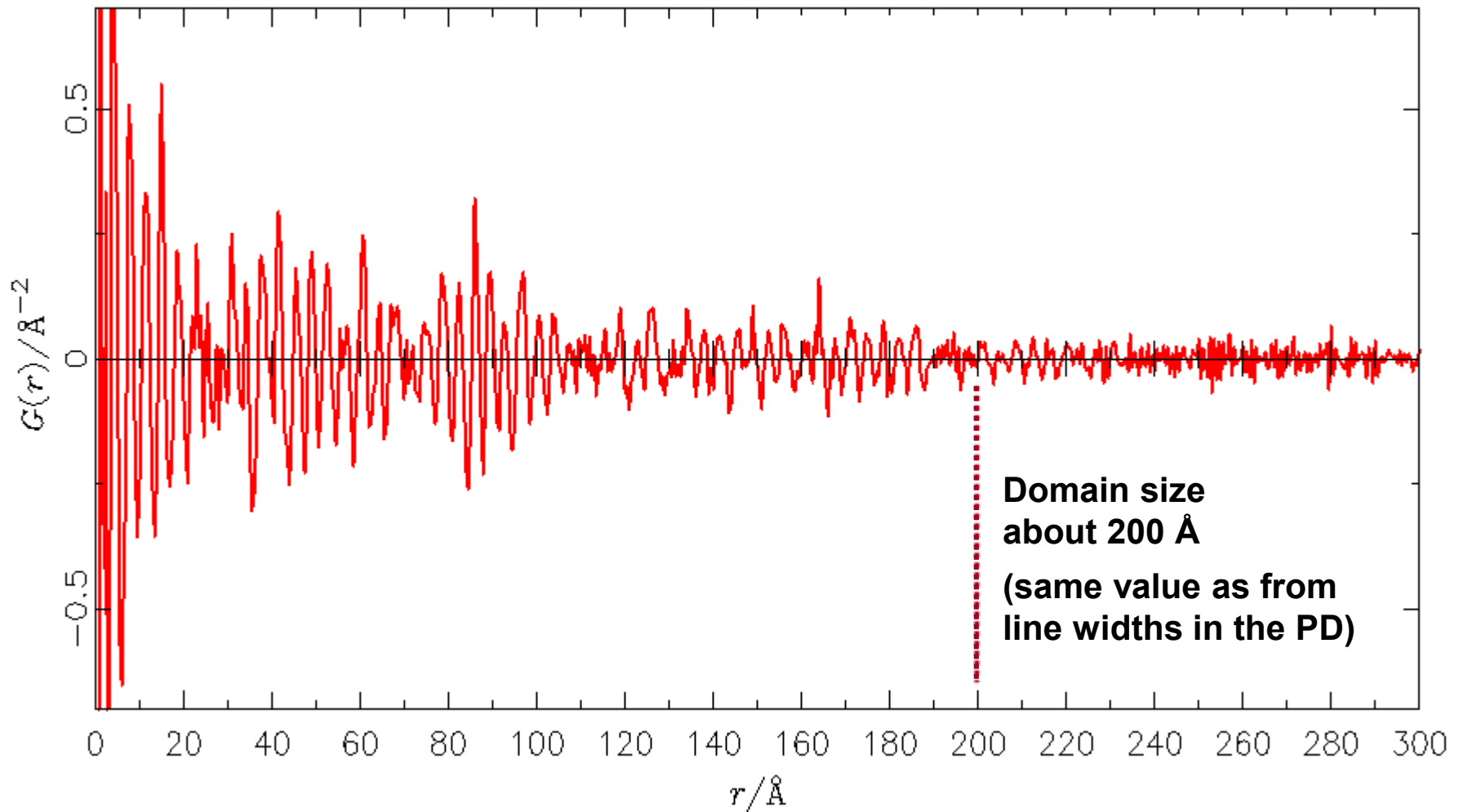


Comparison of experimental and theoretical PDF

Nanocrystalline and amorphous API

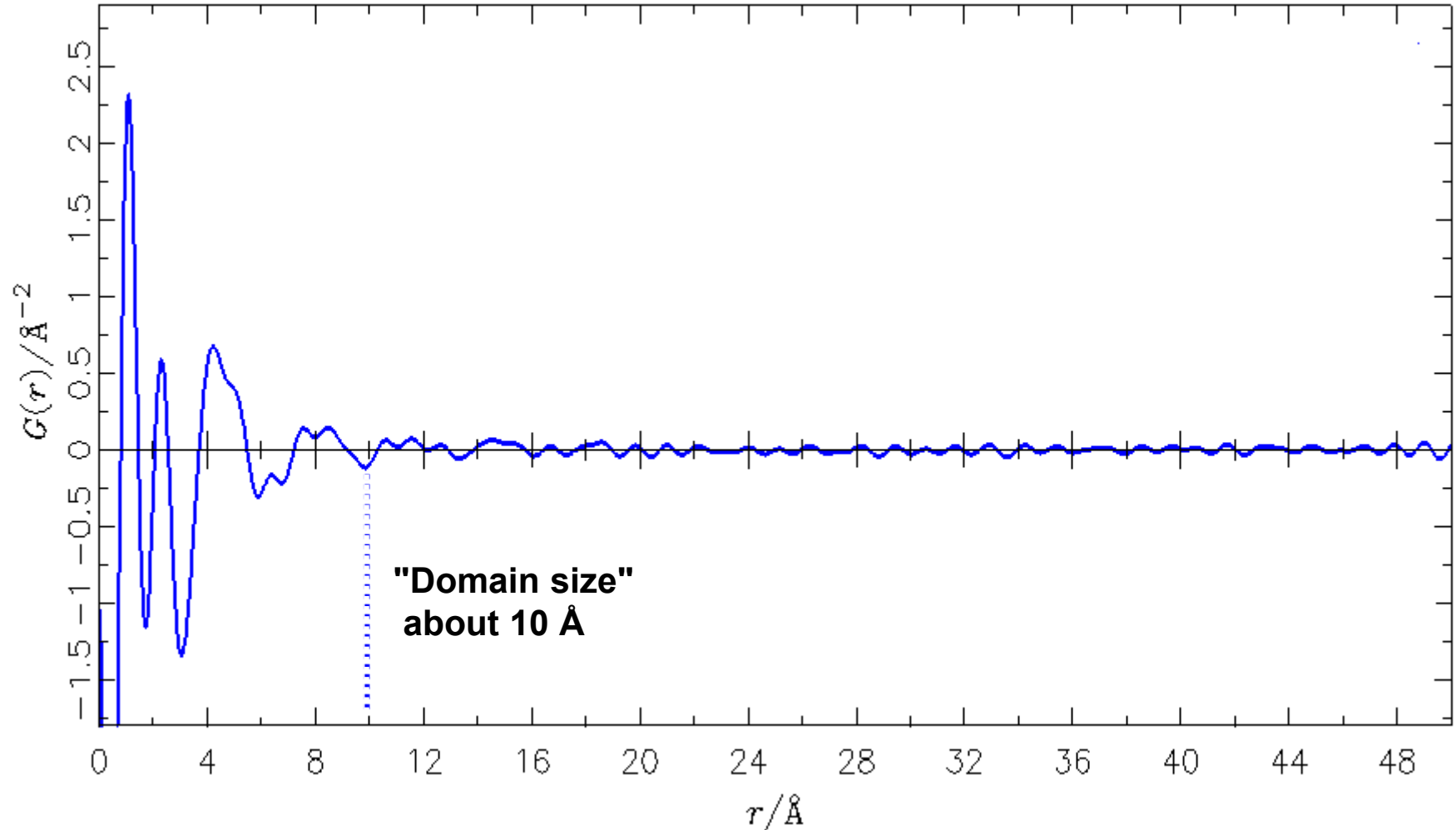


Nanocrystalline and amorphous API



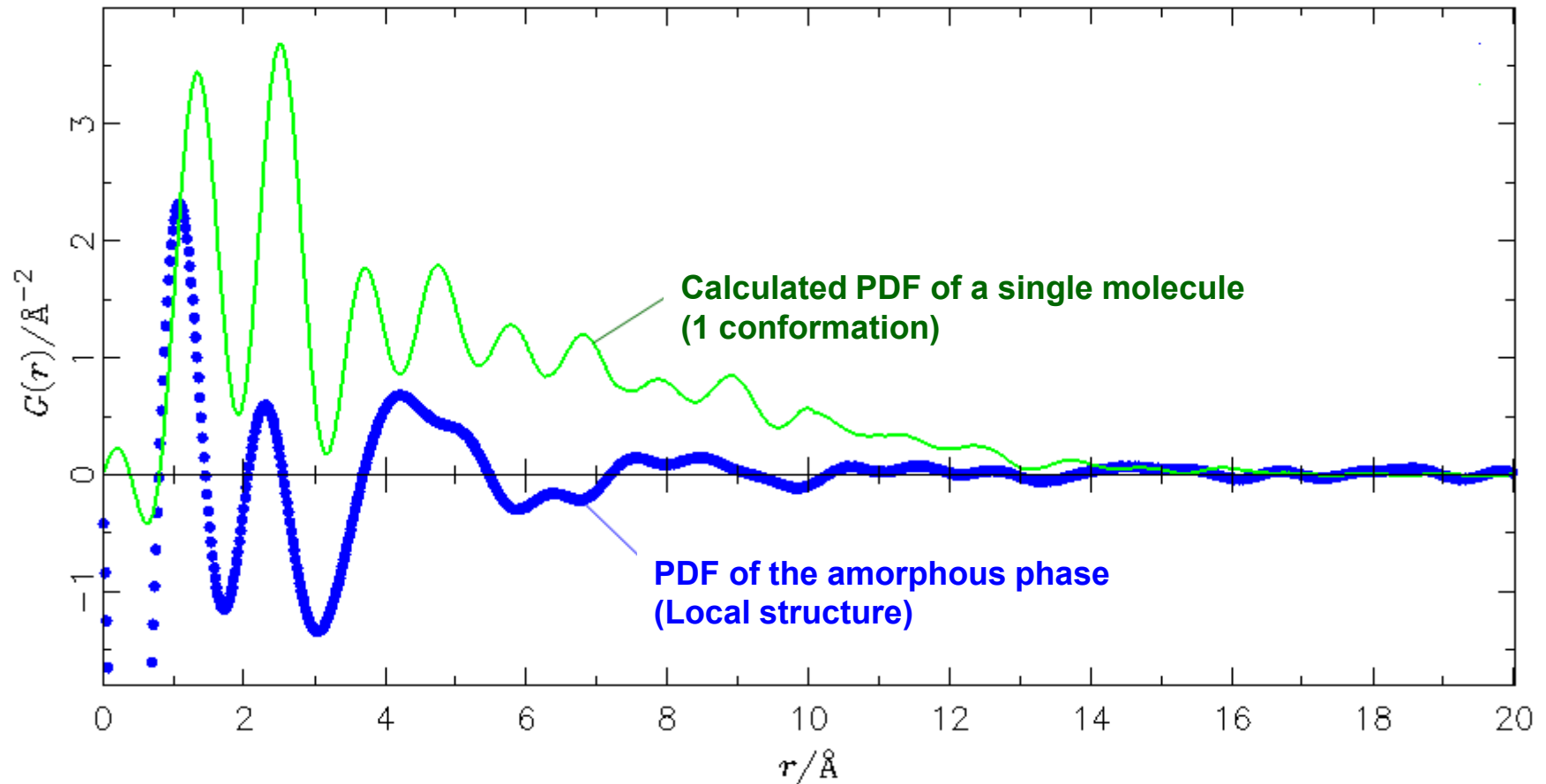
PDF of the nanomilled phase

Nanocrystalline and amorphous API



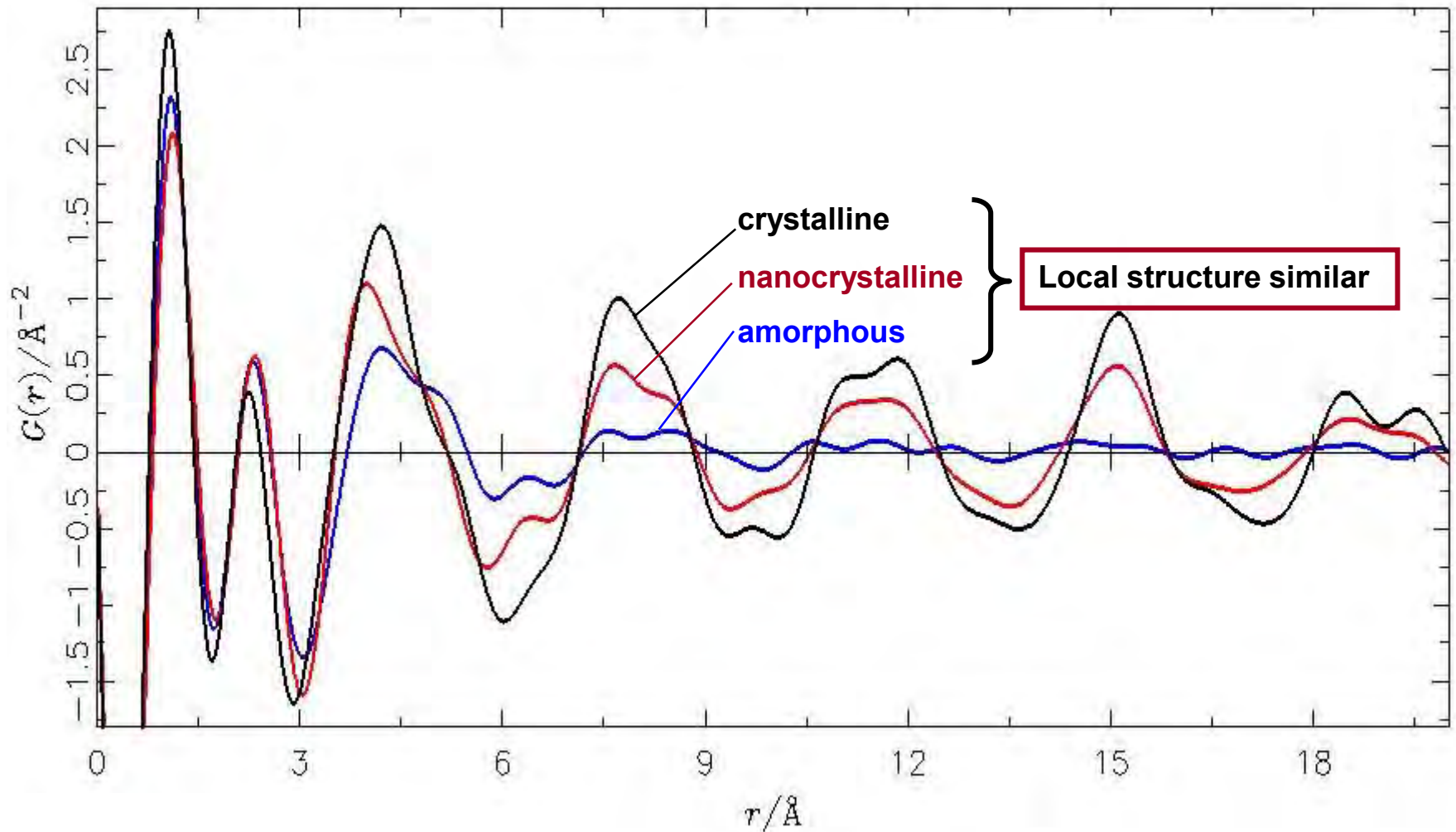
PDF of the amorphous phase

Nanocrystalline and amorphous API



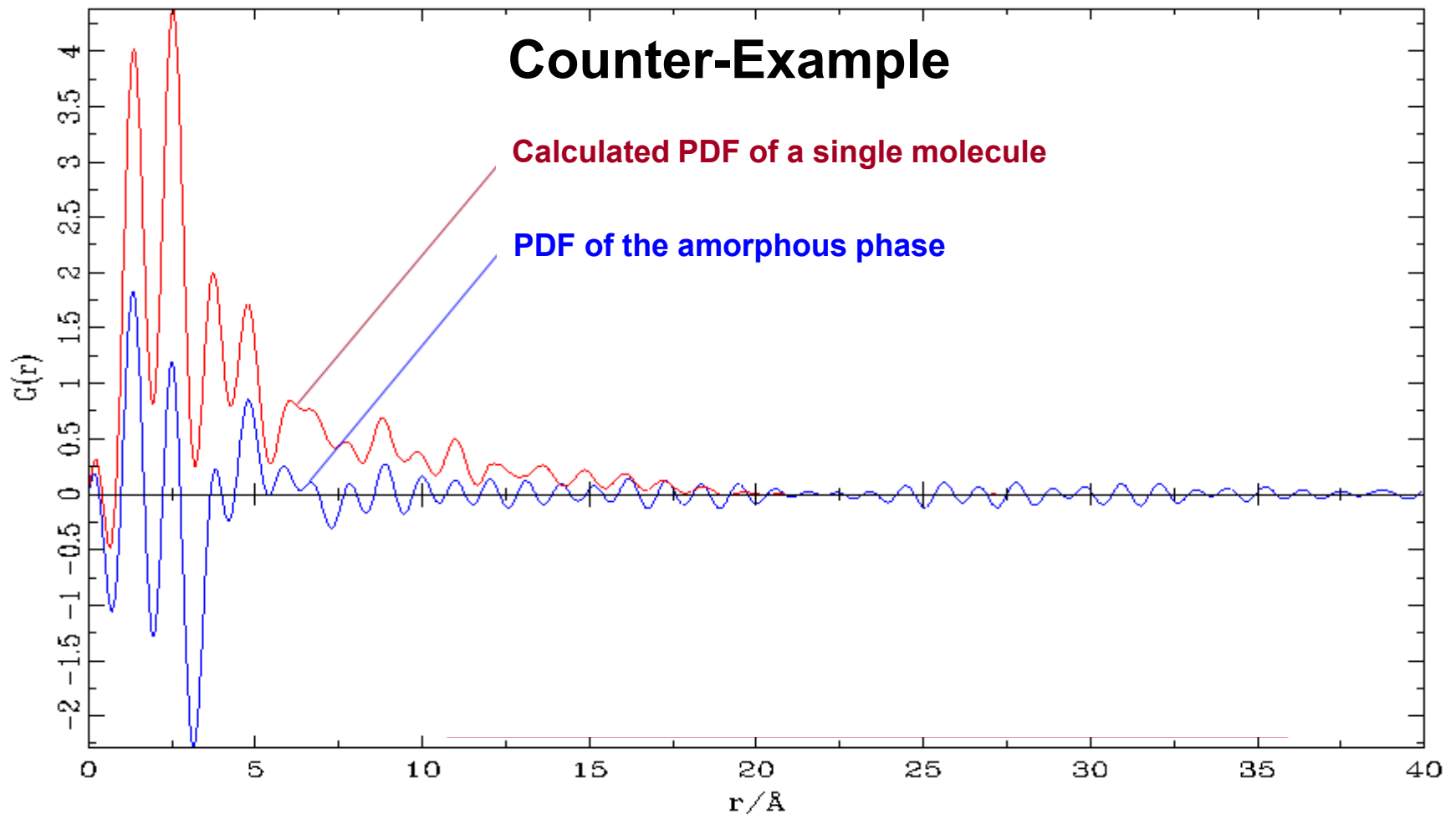
PDF of the amorphous phase

Nanocrystalline and amorphous API



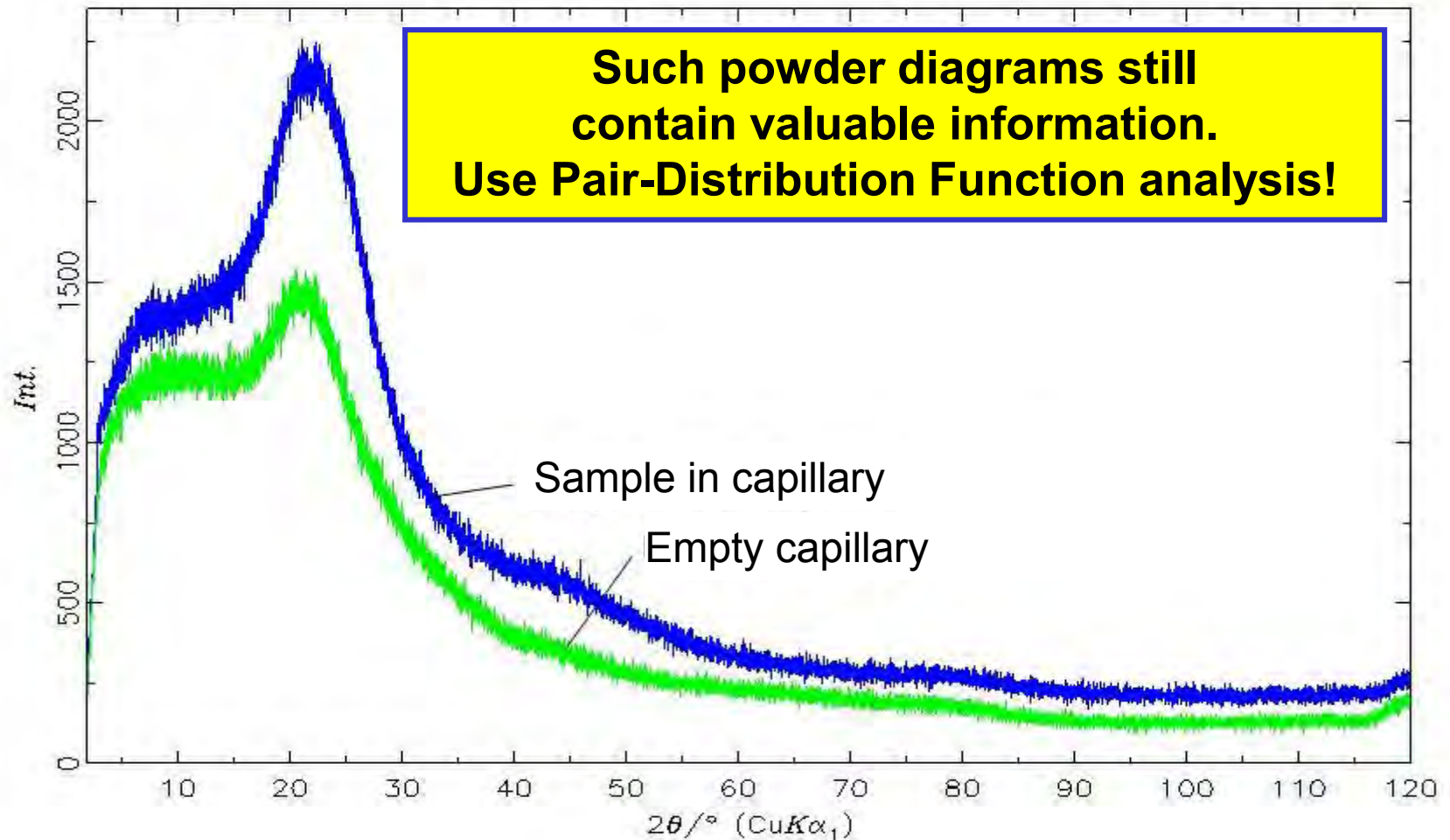
PDF: comparison (0 - 20 Å)

Nanocrystalline and amorphous API



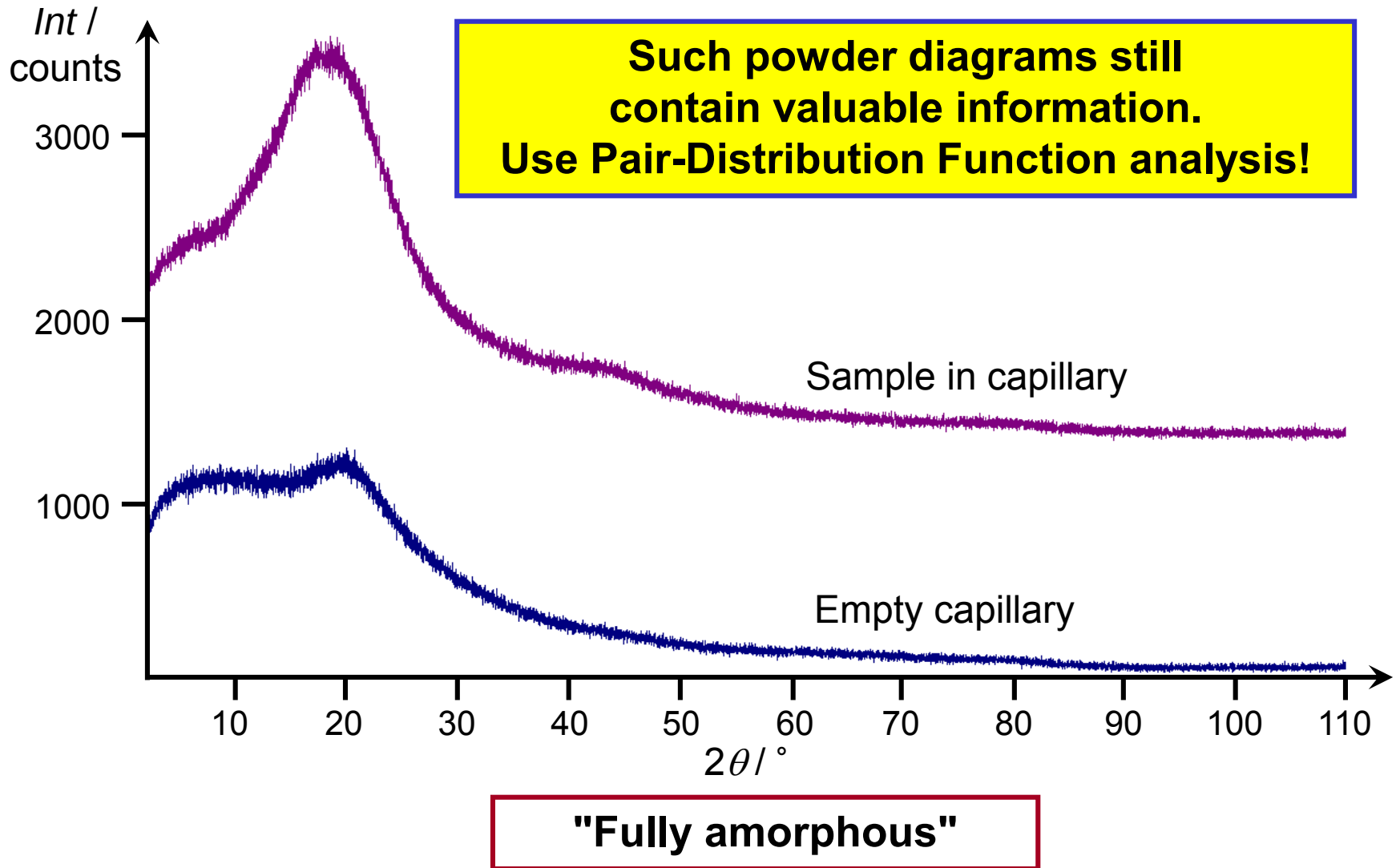
**"Really amorphous".
No visible ordering of neighbouring molecules.**

Nanocrystalline and amorphous API



Amorphous with local structure

Nanocrystalline and amorphous API



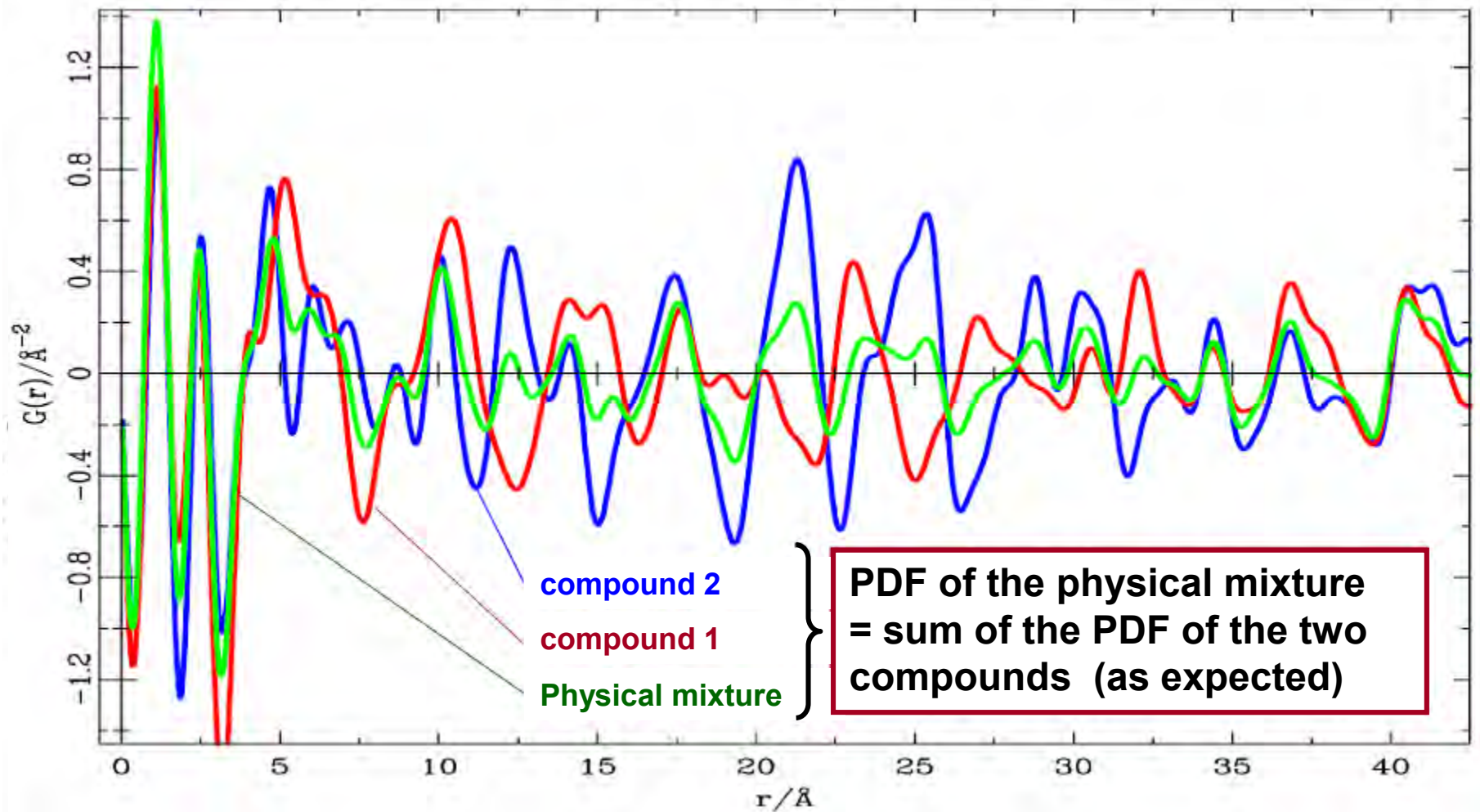
Cocrystal

Cocrystal of 2 organic compounds

Questions:

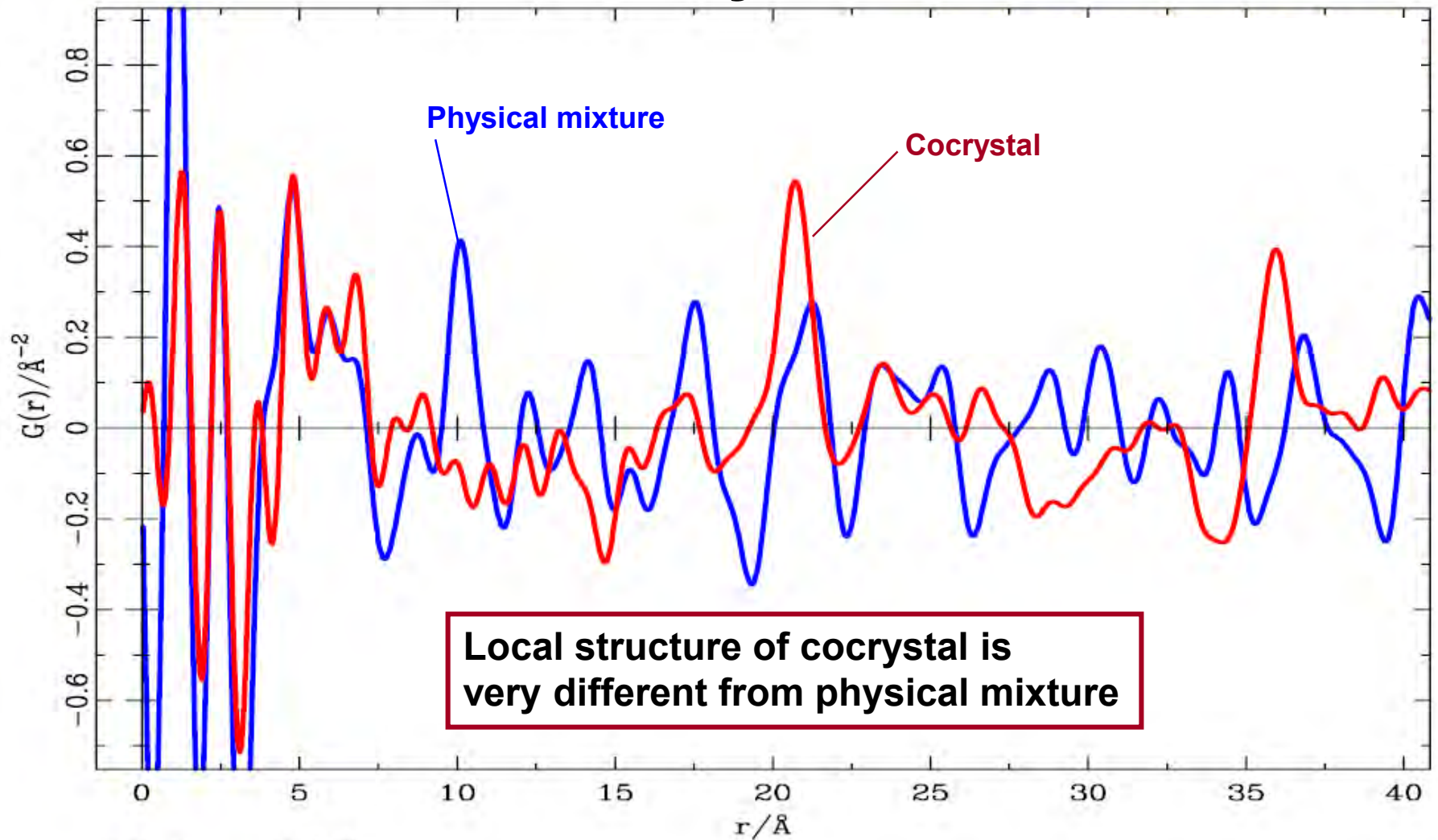
- Cocrystal or physical mixture?**
- Is it still a cocrystal after grinding or micronising?**

Physical mixture



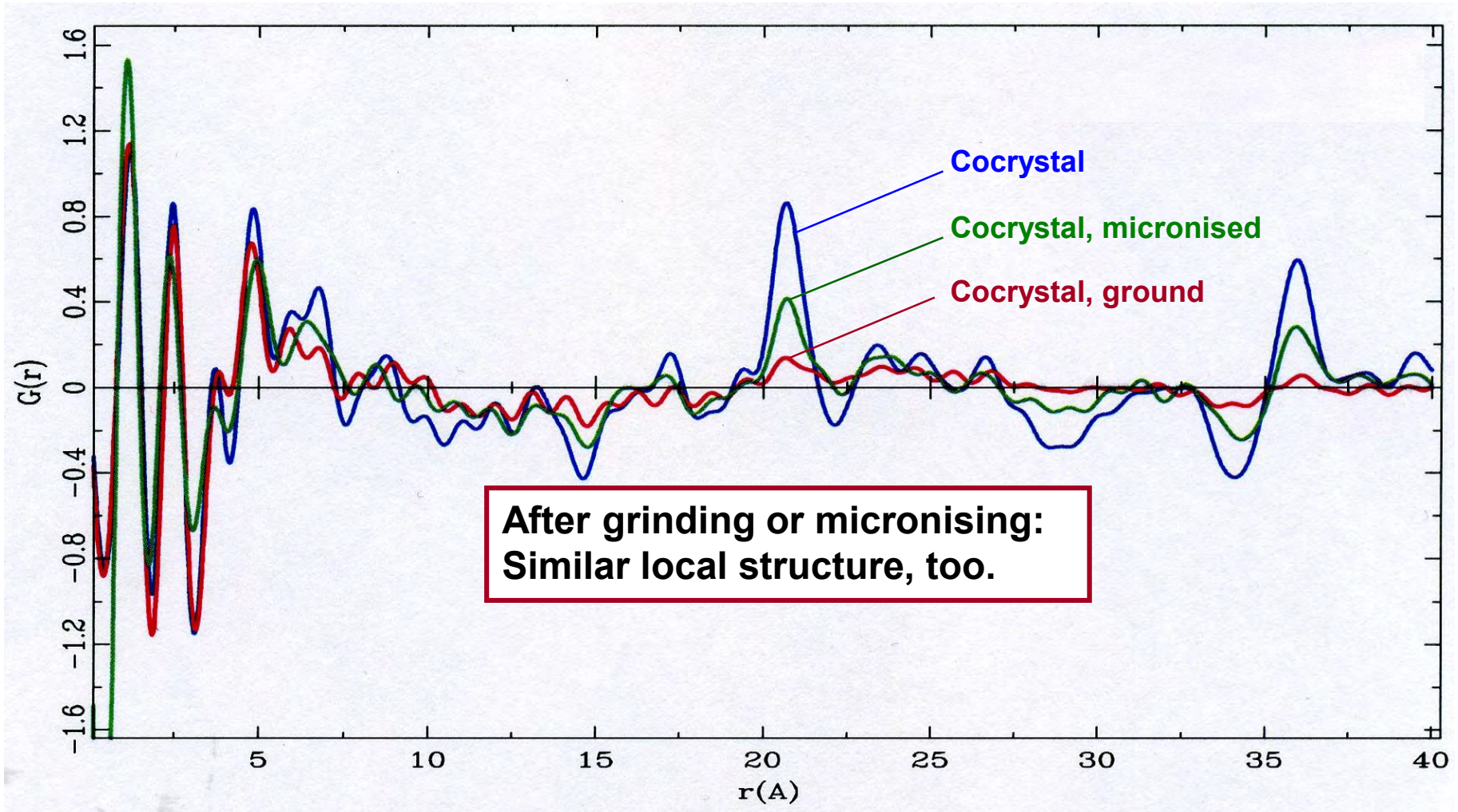
PDF of the physical mixture

Cocrystal



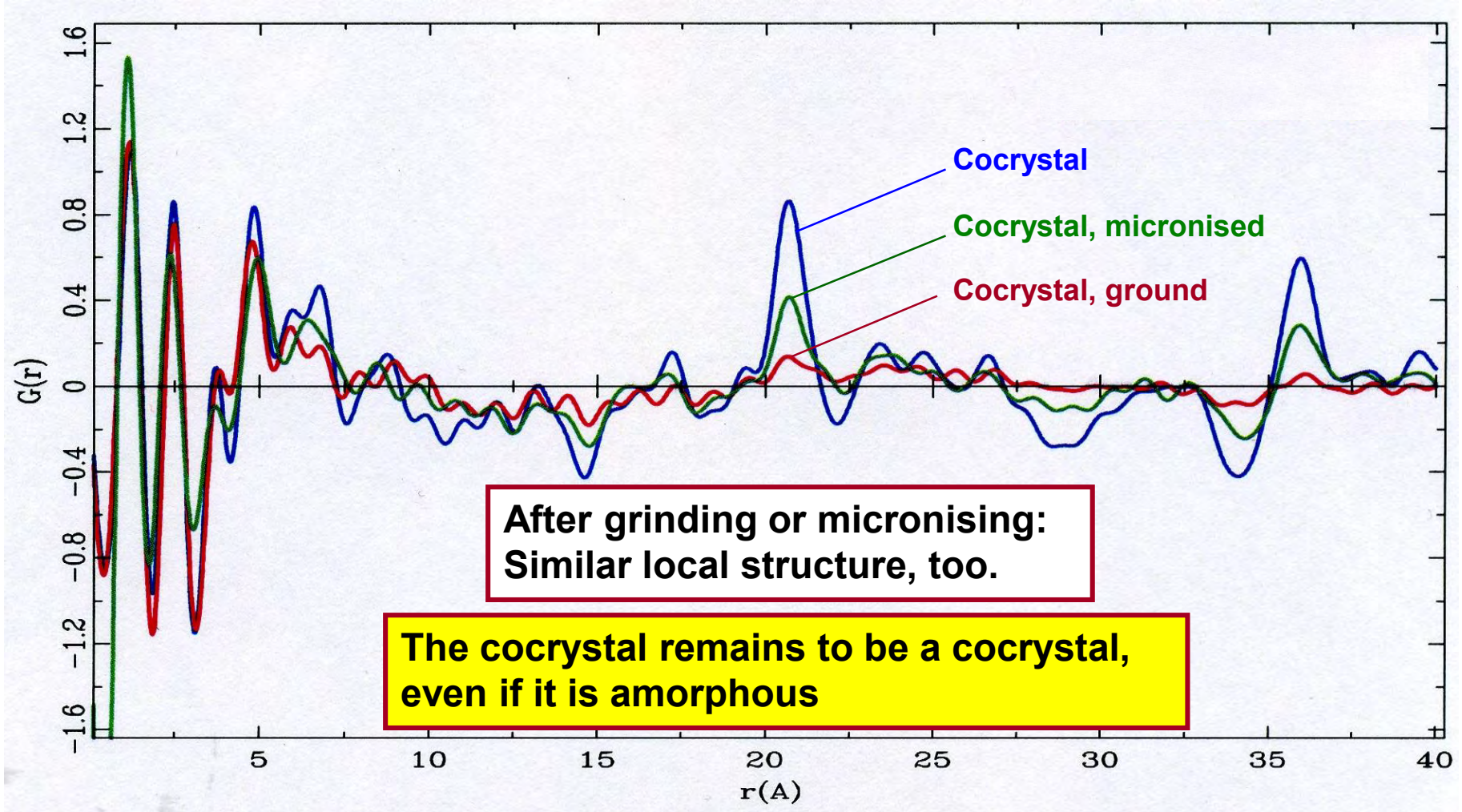
PDF of cocrystal and physical mixture

Cocrystal



Cocrystal: PDF

Cocrystal



Cocrystal: PDF

PDF of a pharmaceutical formulation

Formulation:

- Felodipine (well crystalline API)
- in two different types of "Eudragit" polymers (polyacrylates)
- Melt extrusion

Experimental observation:

10 % API
90 % polymer 1 (Eudragit E)
Poor dissolution behaviour

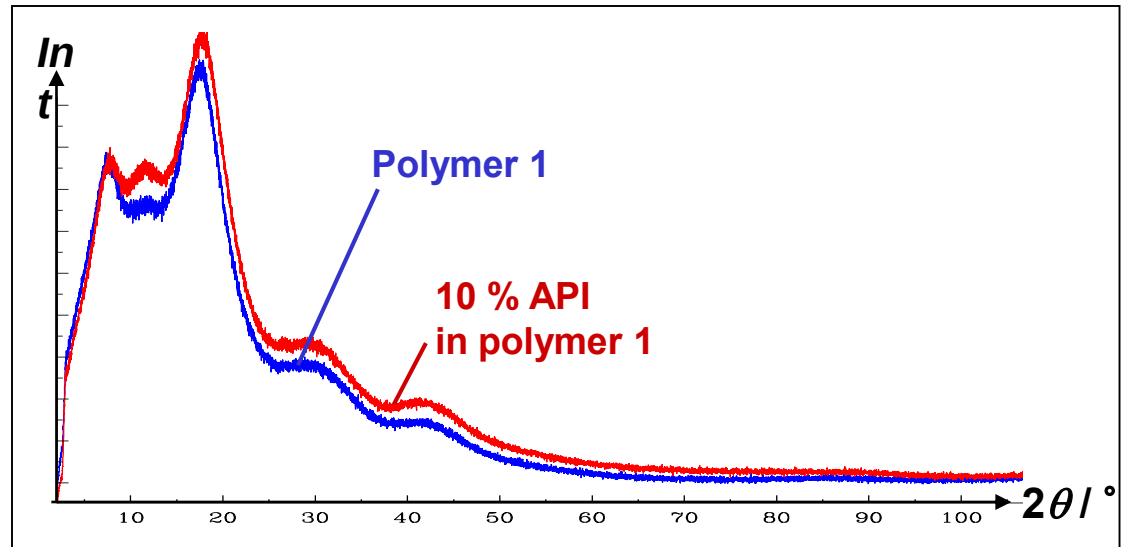
10 % API
90 % polymer 2 (Eudragit NE)
Poor dissolution behaviour

10 % API
85 % polymer 1
5 % polymer 2
Much better dissolution behaviour

Why?

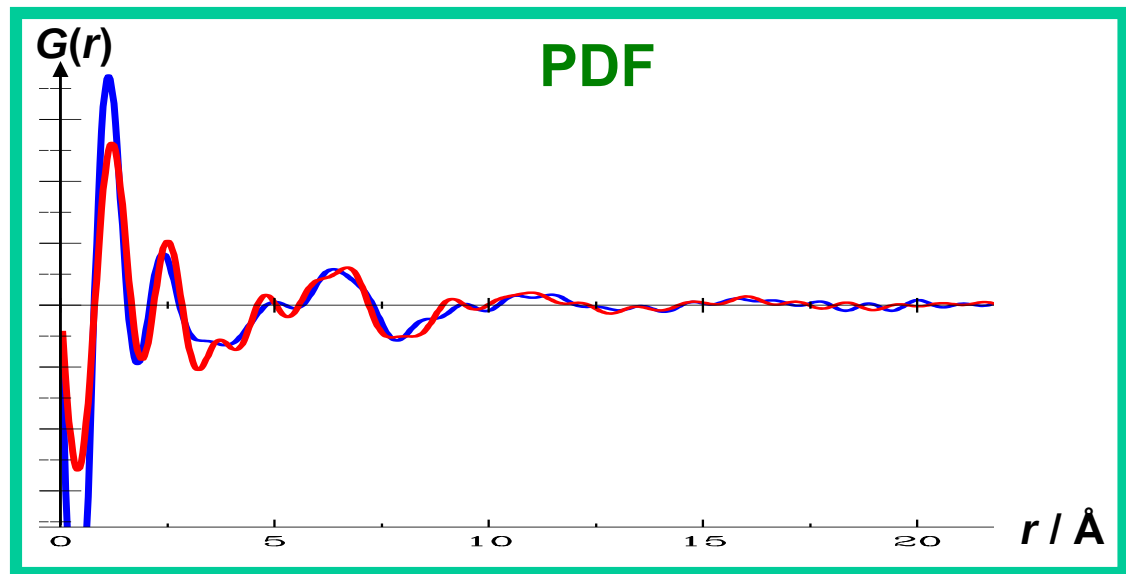
PDF of a pharmaceutical formulation

**10 % API
in polymer 1**



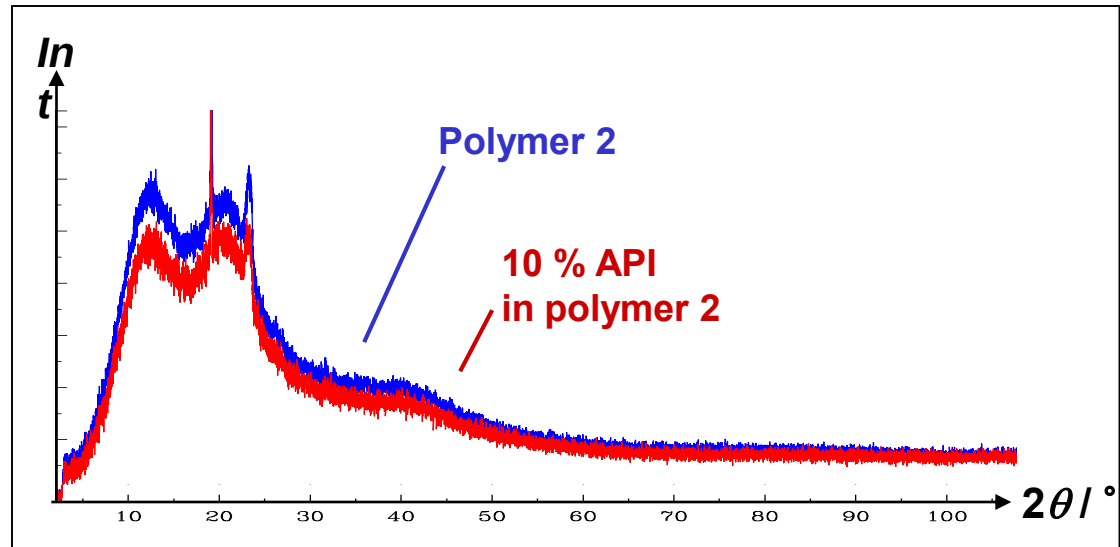
=> API is dissolved

**No structural changes
of polymer 1**



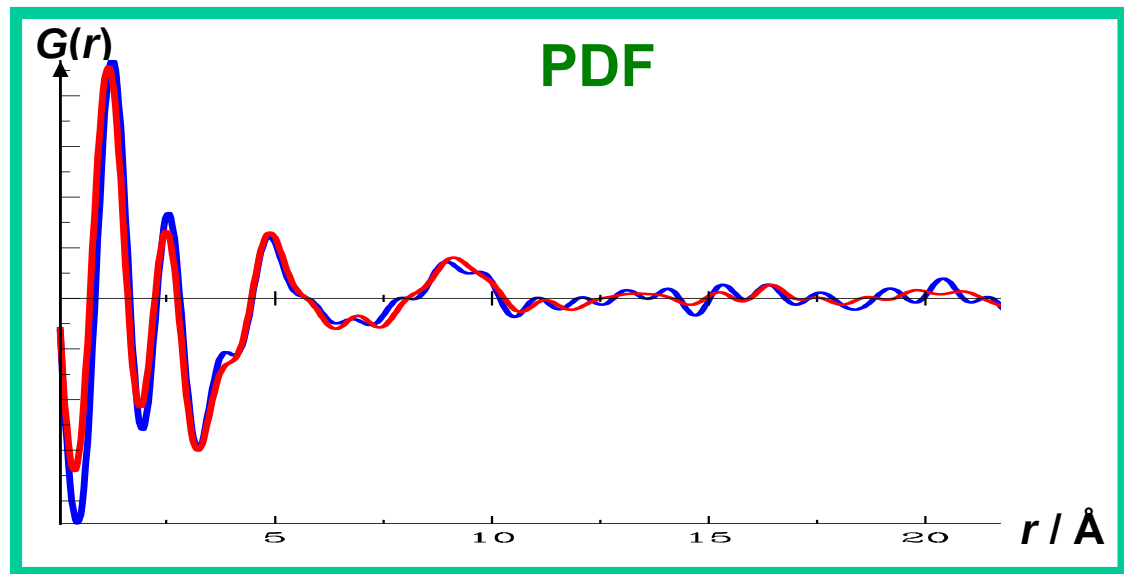
PDF of a pharmaceutical formulation

**10 % API
in polymer 2**



=> API is dissolved

**No structural changes
of polymer 2**



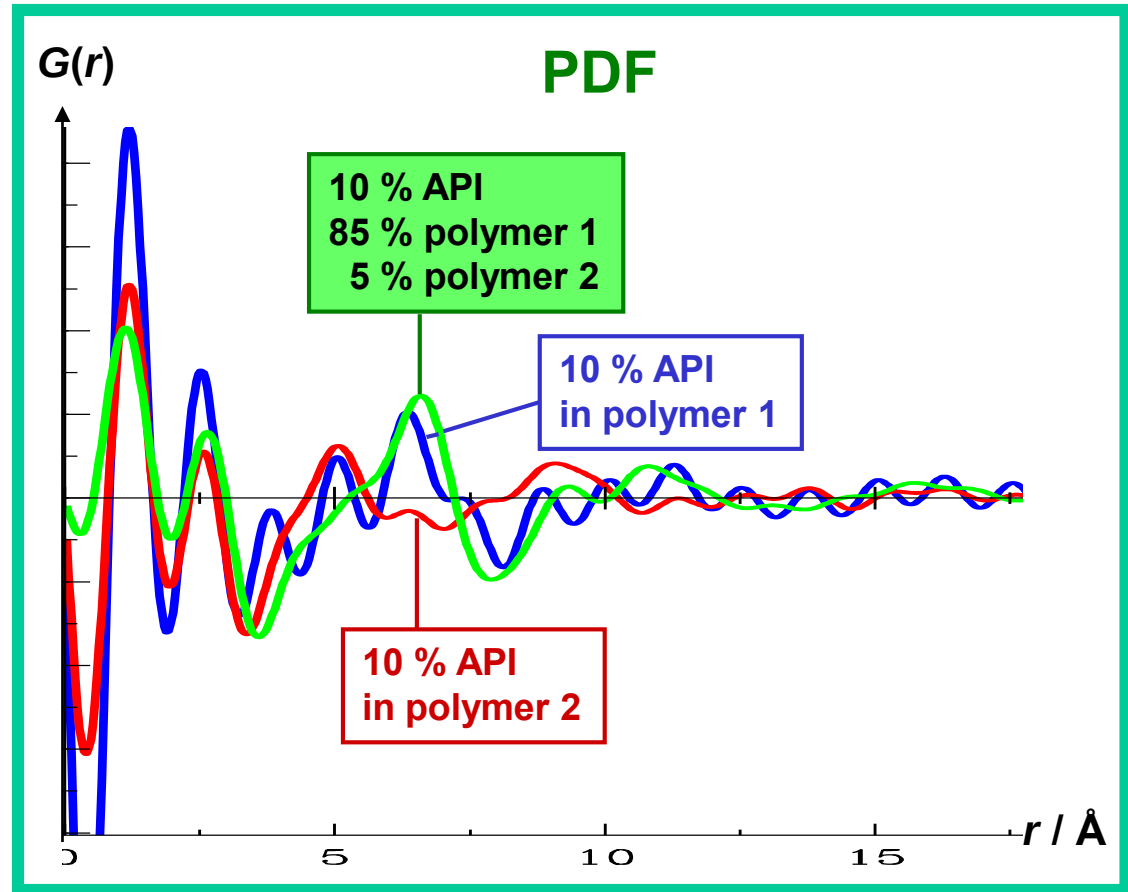
PDF of a pharmaceutical formulation

10 % API

85 % polymer 1

5 % polymer 2

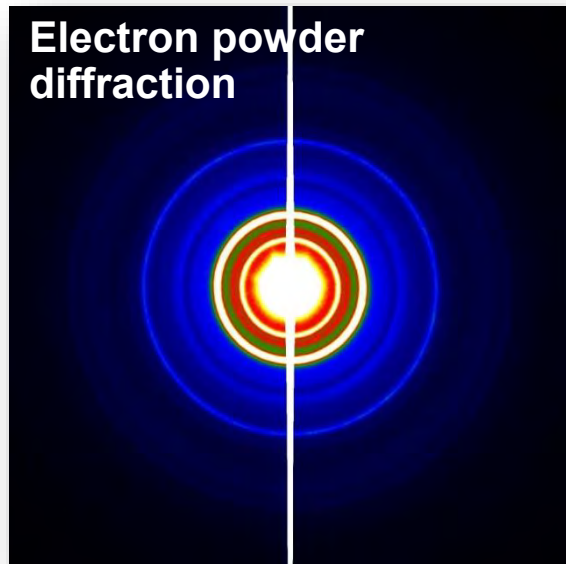
Structure of the mixture changes considerably.



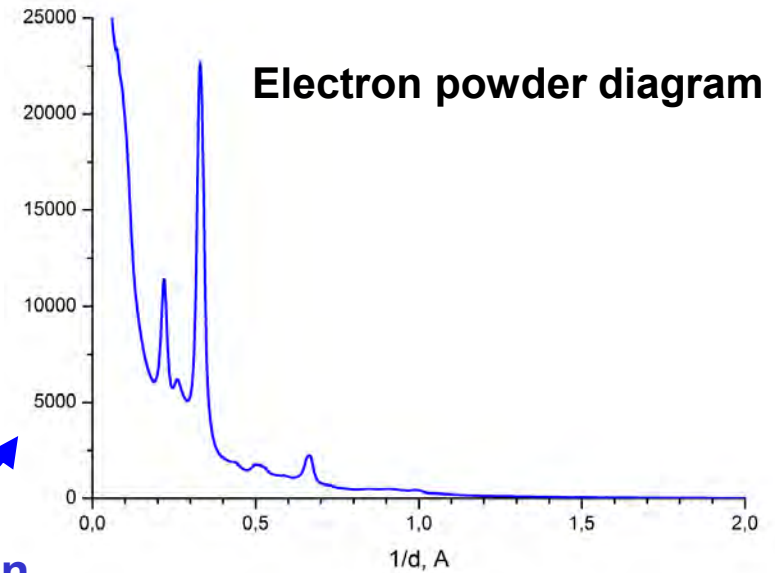
**This explains the observed differences
in the dissolution behaviour.**

PDF from electron diffraction data (ePDF)

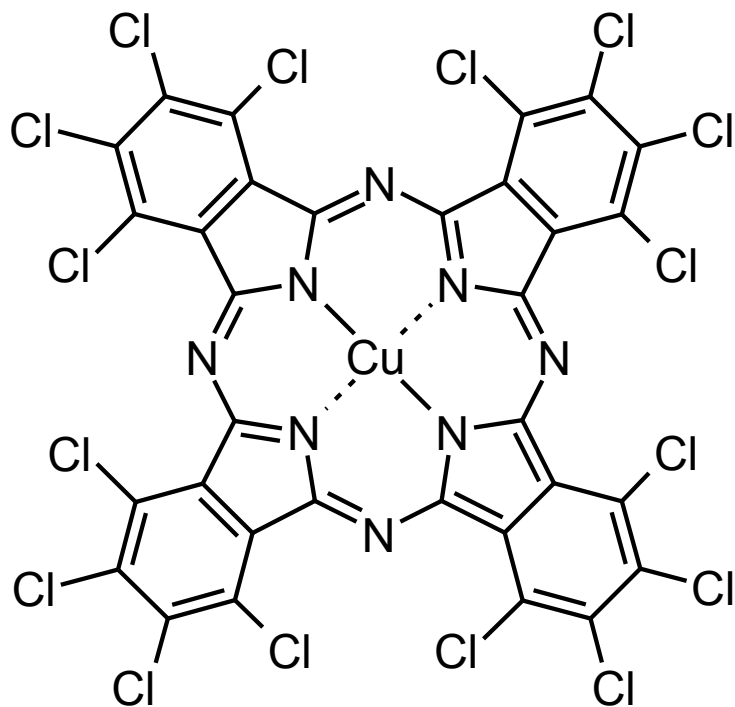
- Polycrystalline thin films
- Amorphous thin films



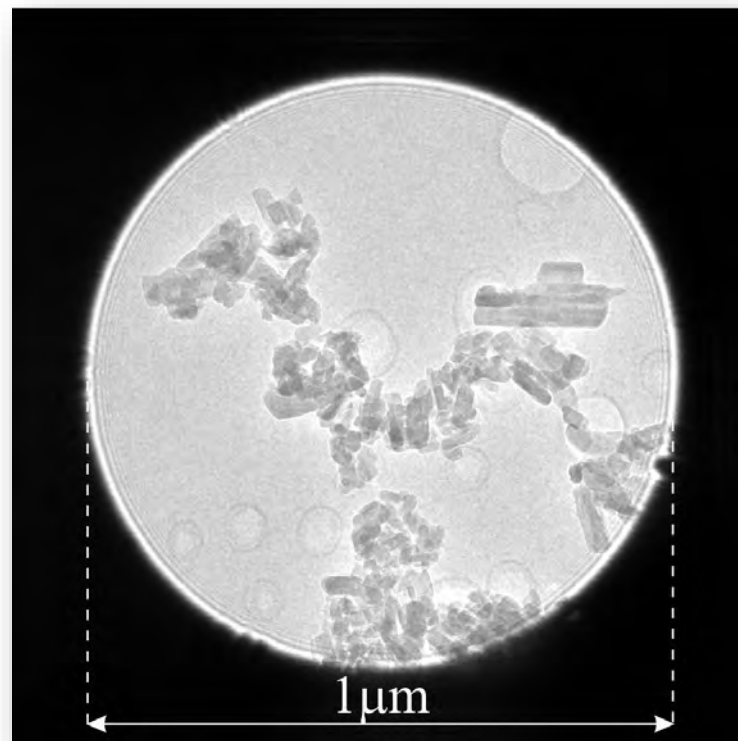
Integration
into 1D profile



PDF from electron diffraction data (ePDF)



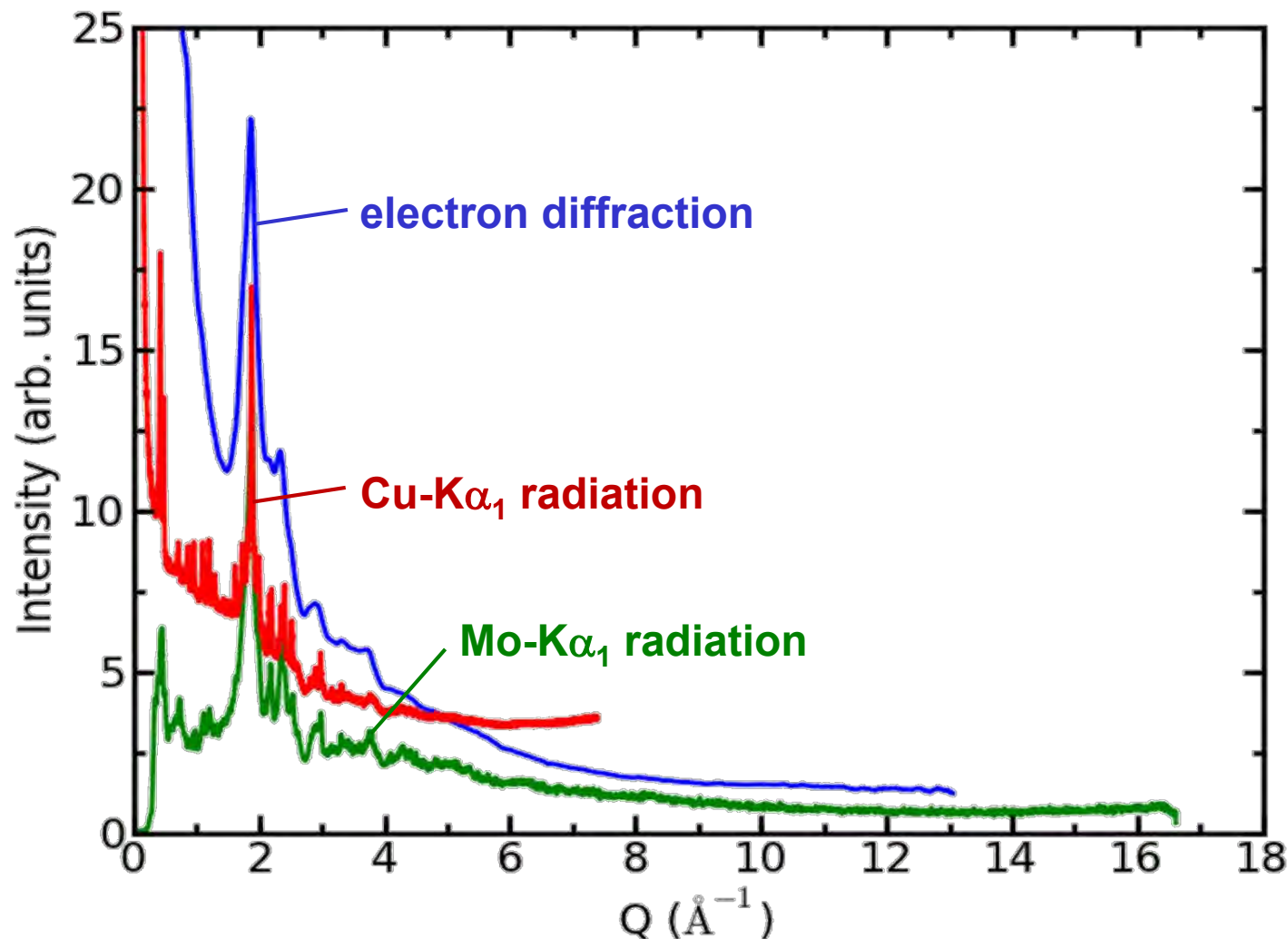
CuPcCl₁₆
(Chlorinated Cu-phthalocyanine)
Pigment Green 7



Data collection:

- good counting statistics
- good orientation statistics

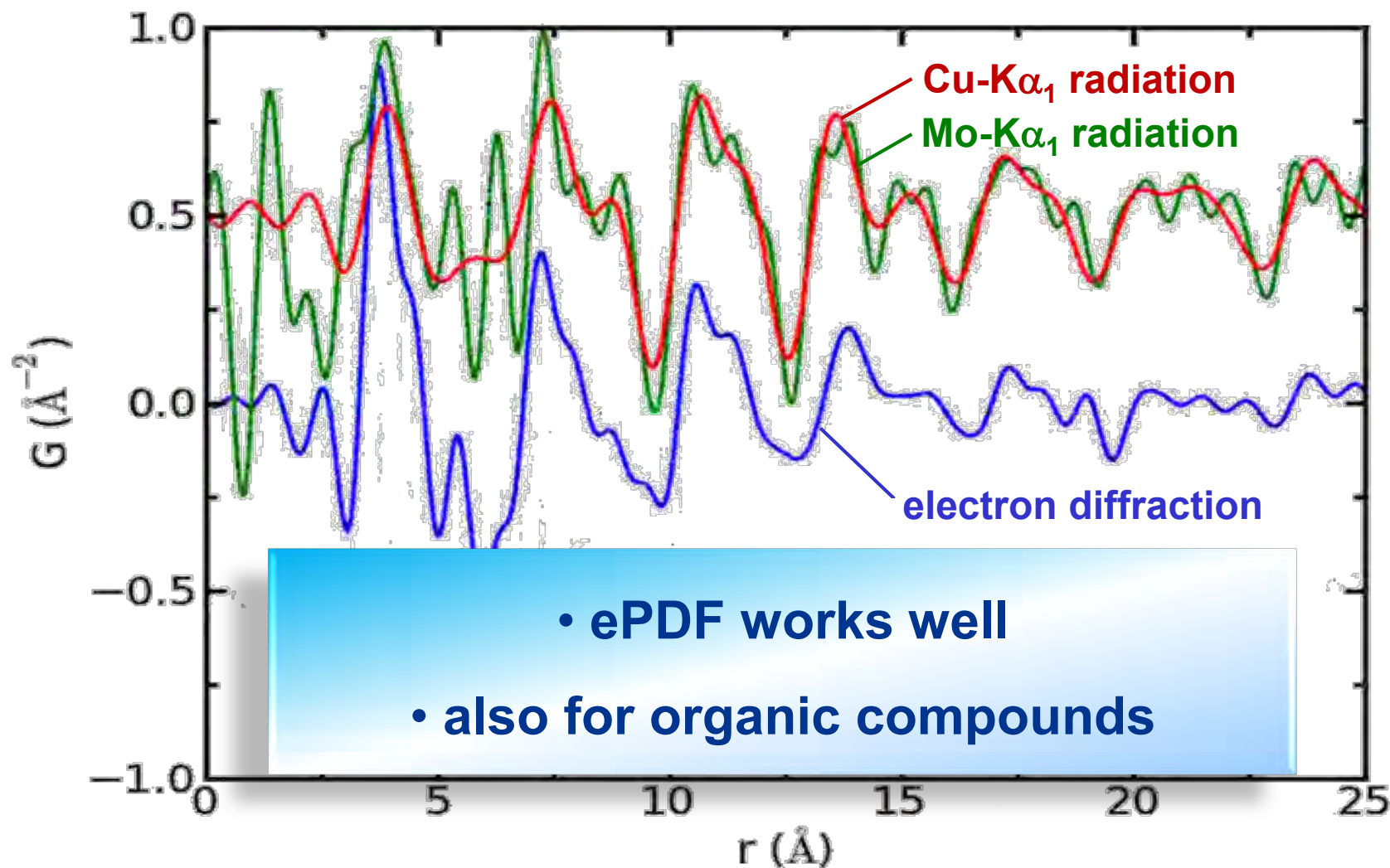
PDF from electron diffraction data (ePDF)



Powder diagrams of CuPcCl_{16}

[S. Billinge, C. Farrow, M. Kanatzidis, T.E. Gorelik, M.U. Schmidt, *Patent* US 8921783 B2, **2014**.
T.E. Gorelik, M.U. Schmidt, U. Kolb, S.J.L. Billinge, *Microsc. Microanal.*, **2015**, 21, 459-471]

PDF from electron diffraction data (ePDF)



PDF of CuPcCl_{16}

[S. Billinge, C. Farrow, M. Kanatzidis, T.E. Gorelik, M.U. Schmidt, *Patent* US 8921783 B2, **2014**.
T.E. Gorelik, M.U. Schmidt, U. Kolb, S.J.L. Billinge, *Microsc. Microanal.*, **2015**, 21, 459-471]

Applications of the PDF

- Investigation of local structures and packing motifs in
 - nanocrystalline organic compounds. *(Done)*
 - amorphous organic compounds. *(Done)*
 - disordered organic and organometallic compounds. *(Done)*
- Comparing the local structures in hydrates, solvates, and polymorphs. *(Done)*
Hydrate → Anhydrate (amorphous): Similar local structure?
- Detecting the crystallisation of amorphous compounds (PDF: more sensitive than X-ray powder pattern). *(Done)*
- Pharmaceutical cocrystals:
Proving that a micronized cocrystal remains to be a cocrystal, even if it is amorphous. *(Done)*
- Do different amorphous states exist? *(Done)*

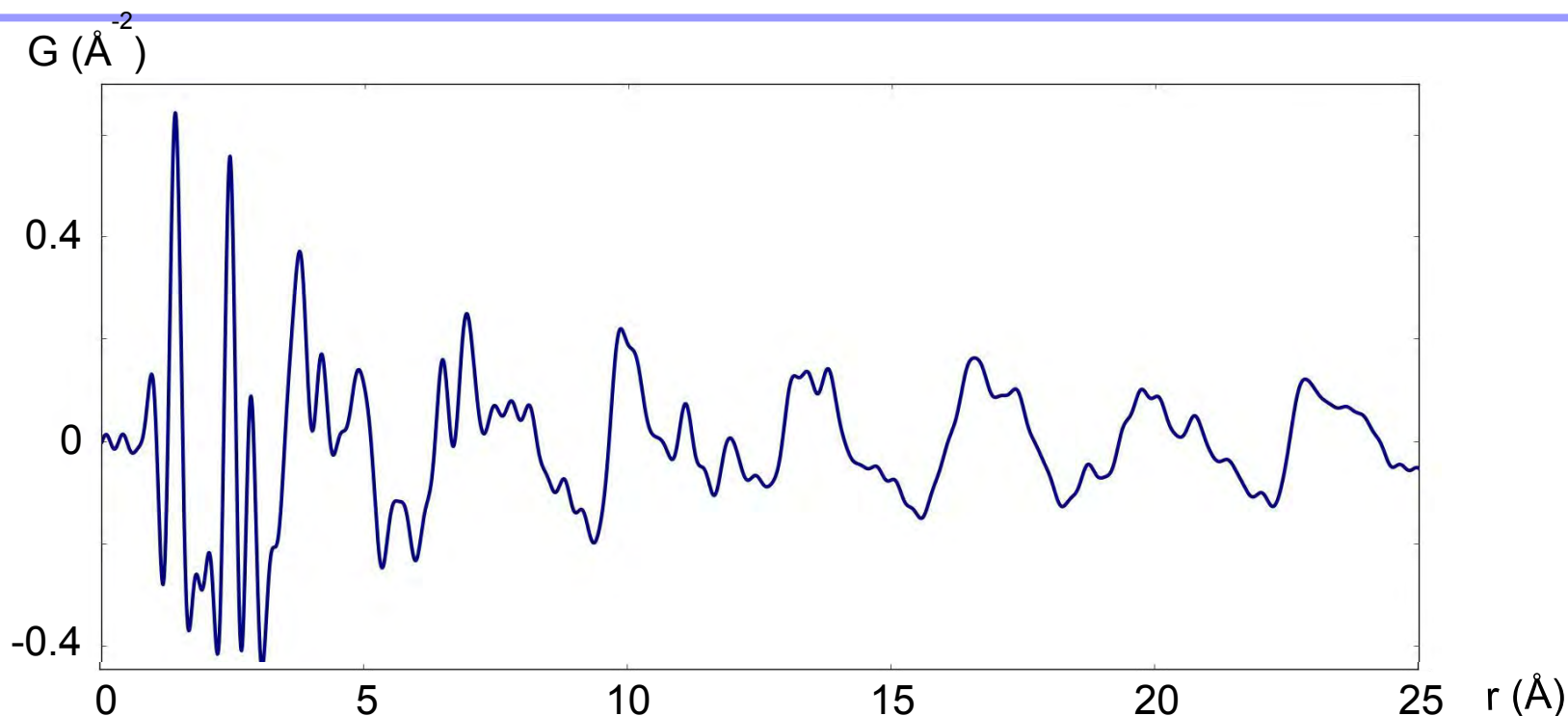
Applications of the PDF

- **Determination of crystal structures of nanocrystalline organic compounds by fit to the PDF curve (*Done*).**

[D. Prill, P. Juhás, S.J.L. Billinge, M.U. Schmidt, Acta Cryst. **2016**, A72, 62-72]

Structure determination by fit to the PDF curve

PDF of organic compounds



Intramolecular atom-atom distances

- covalent bonds (hard)
- small vibrational amplitudes
- small variation of distances

Sharp signals

B_{intra}

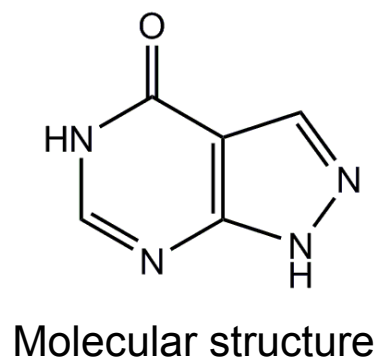
Intermolecular atom-atom distances

- van der Waals bonds (soft)
- large vibrational amplitudes
- large variation of distances

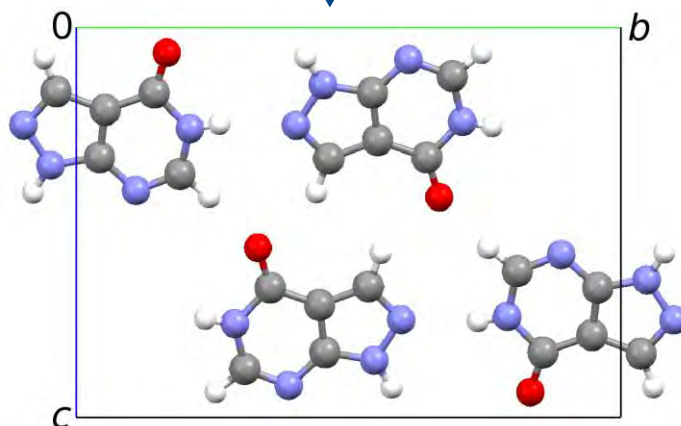
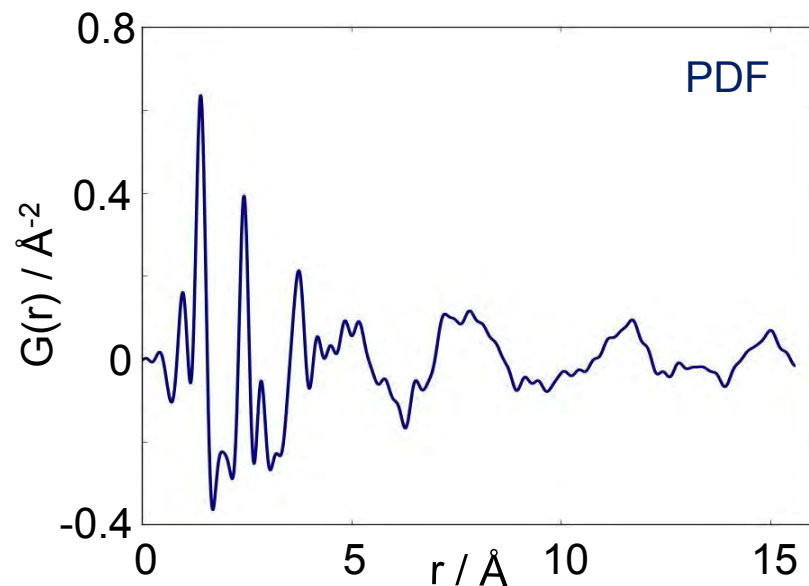
Broad signals

B_{inter}

Structure determination by fit to the PDF data



+



Structure determination by fit to the PDF data

Fit of a structural model to the PDF:

- Inorganic compounds: move individual atoms
- Organic compounds: move whole molecules

Structure determination by fit to the PDF data

Given:

- PDF data (reliable, high resolution) => Synchrotron data
- Molecular geometry (with flexible torsions)

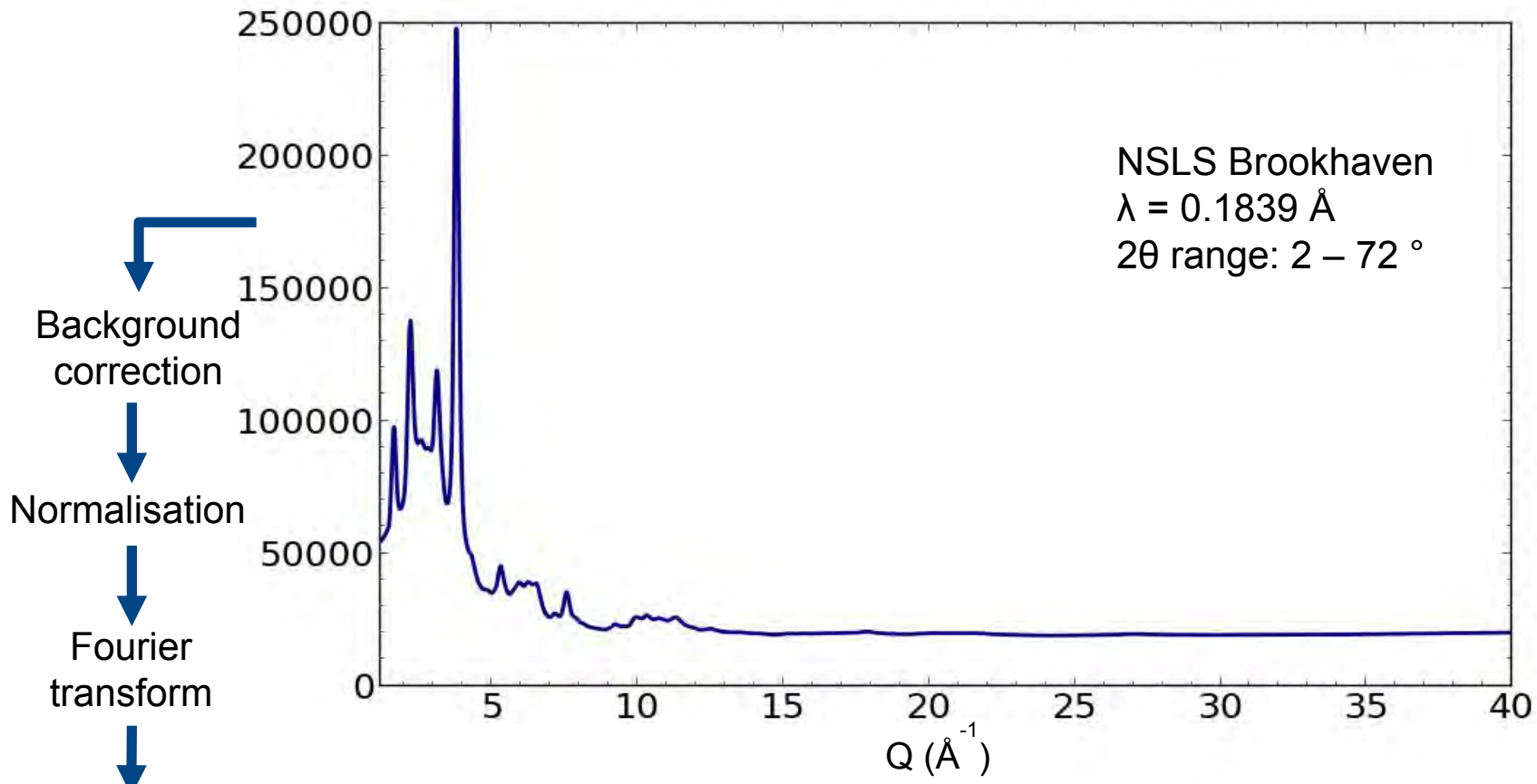
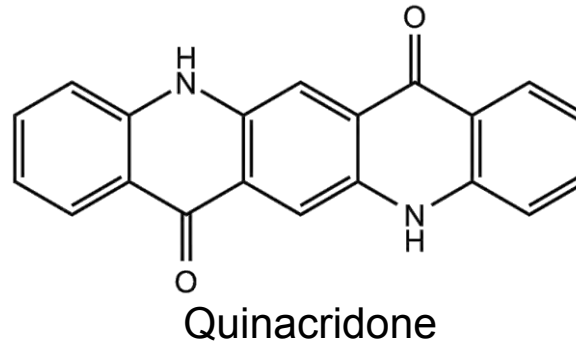
To be determined:

- Space group
- Lattice parameters
- Position of the molecule
- Orientation of the molecule
- Intramolecular degrees of freedom
- B_{intra} , B_{inter} , Scale factor

Structure determination by fit to the PDF data

Example 1:

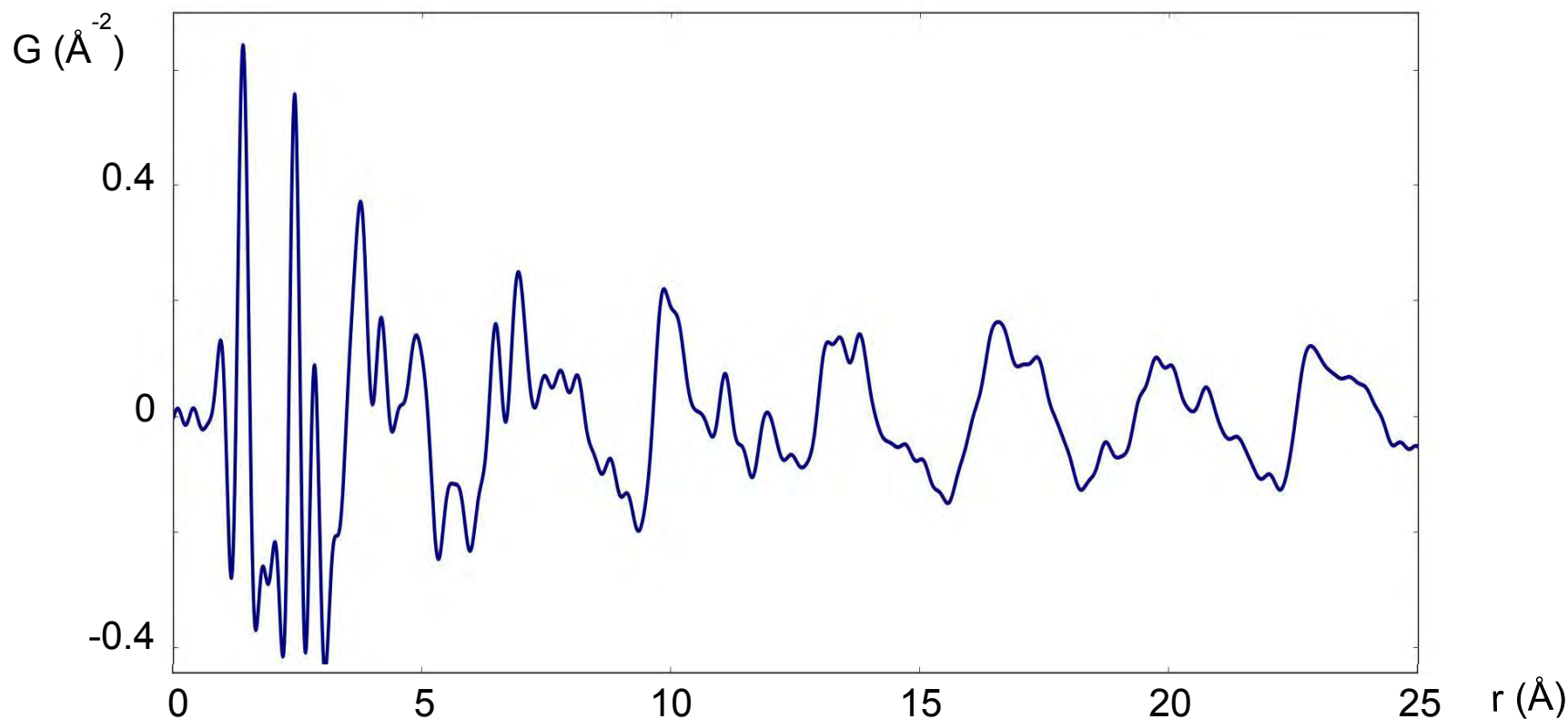
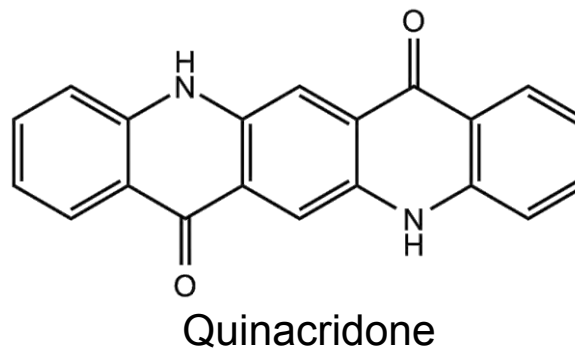
- lattice parameters given
- space group given



Structure determination by fit to the PDF data

Example 1:

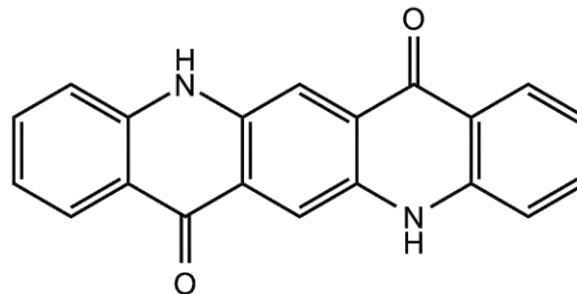
- lattice parameters given
- space group given



Structure determination by fit to the PDF data

Example 1:

- lattice parameters given
- space group given



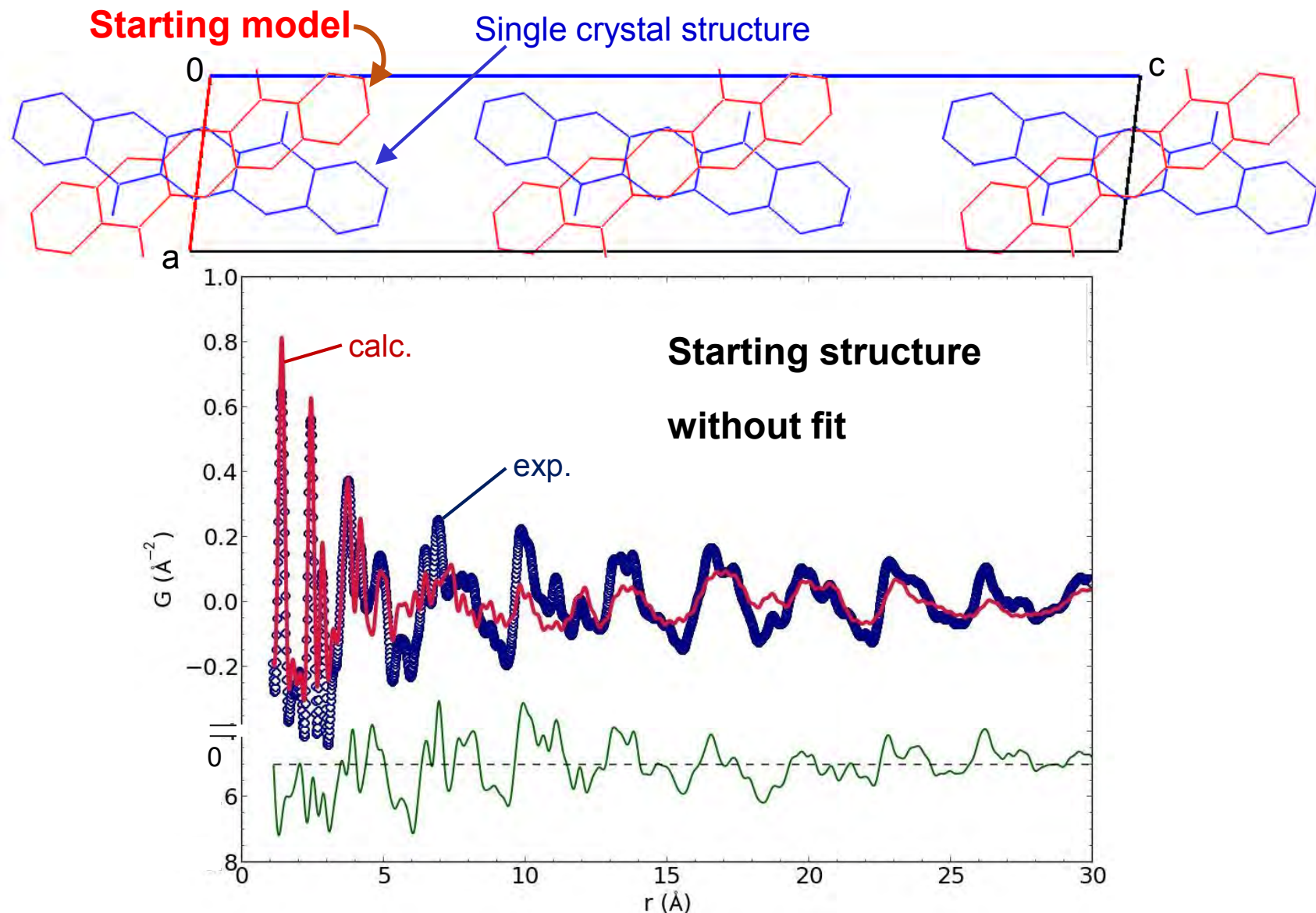
To be determined:

- position of the molecule (starting from random values)
- orientation of the molecule (starting from random values)
- B_{intra} , B_{inter}
- scale factor
- lattice parameters re-refined

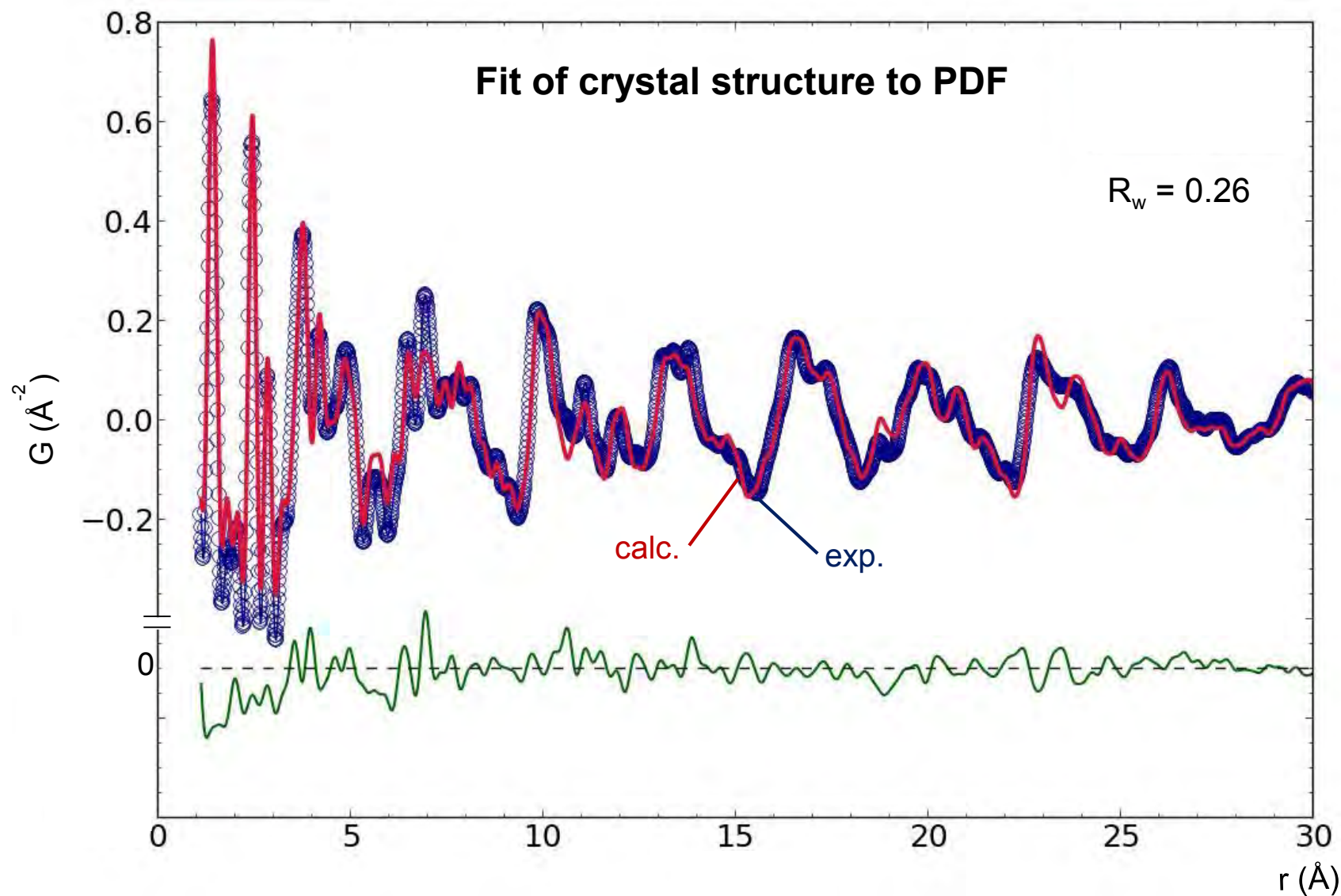
Programs:

- DiffPy-CMI (Group of Simon Billinge)
- TOPAS-6 (Alan Coelho / Bruker)

Structure determination by fit to the PDF data

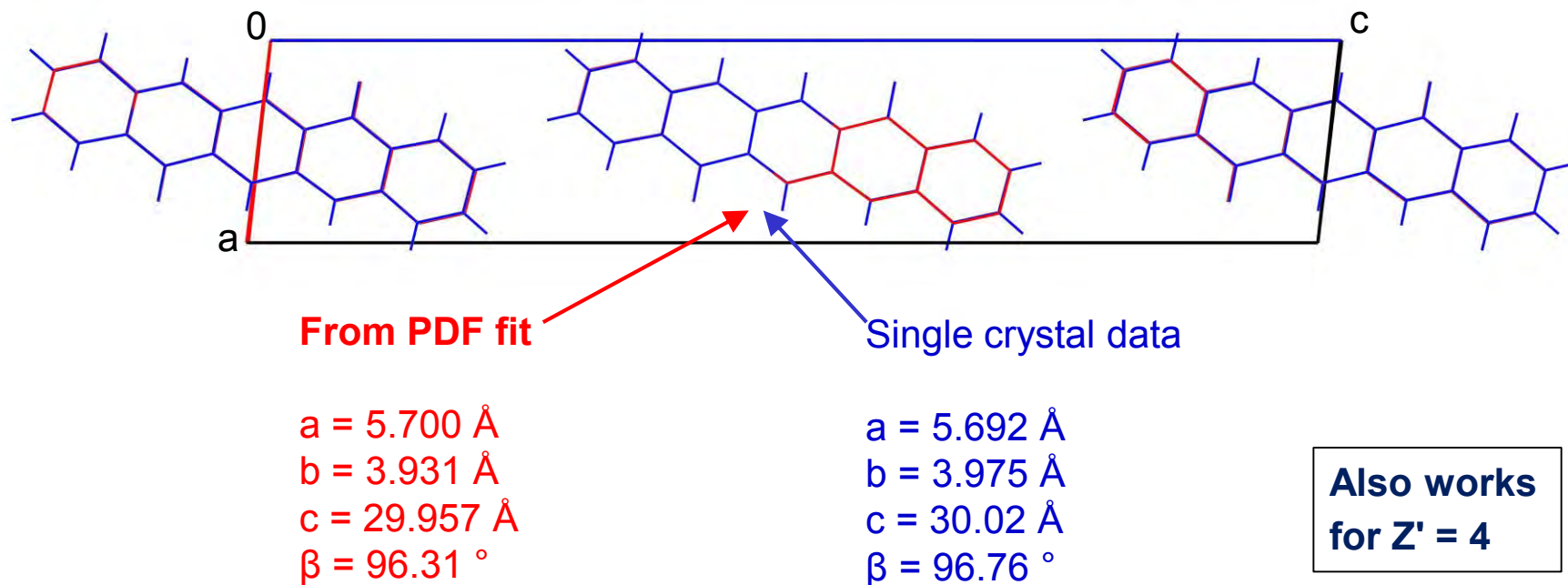


Structure determination by fit to the PDF data



Structure determination by fit to the PDF data

Fit of crystal structure to PDF



Structure solved by fit to the PDF

(Lattice parameters and space group used as input)

Structure determination from scratch by fit to the PDF data

Structure determination from scratch

Input:

- Molecular geometry
- PDF curve

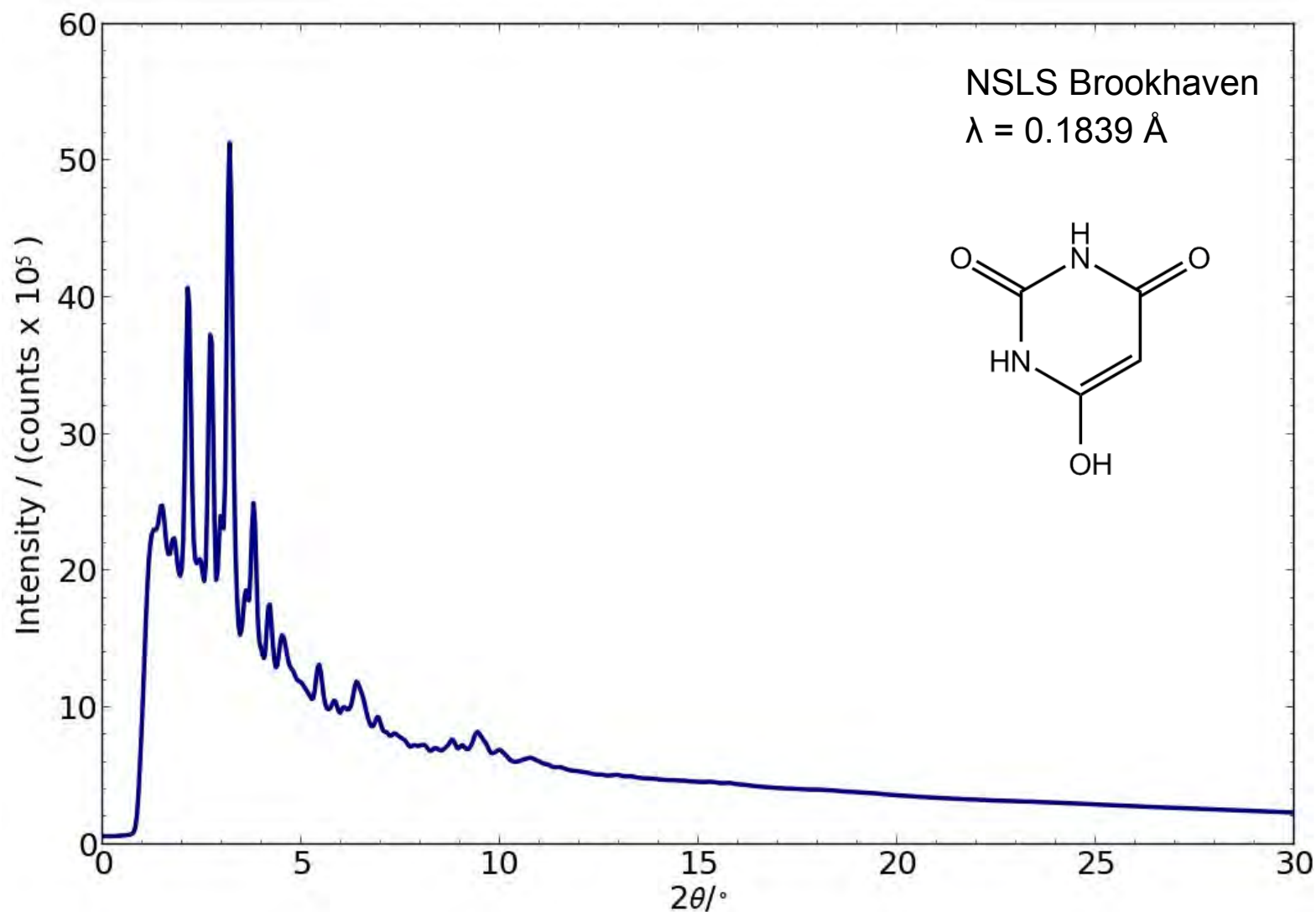
To be determined:

- Space group (Statistically frequent space groups)
 - Lattice parameters
 - Position of the molecule
 - Orientation of the molecule
 - B_{intra} , B_{inter}
 - Scale factor
- } Starting from random values
(About 1 Mio trials)

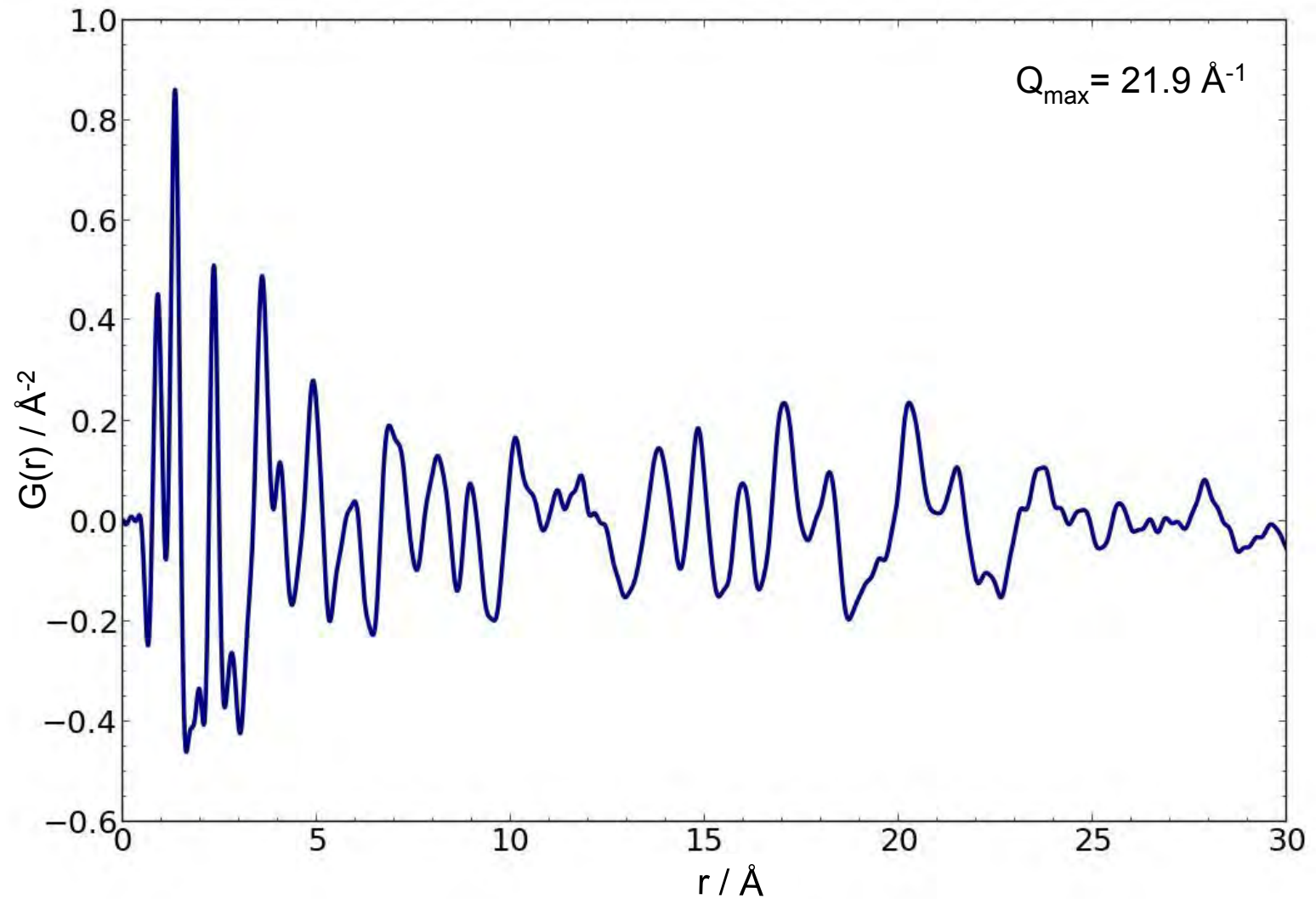
Programs:

- Set up and structure generation: self-developed software FIDEL
- PDF fit: TOPAS-6

Structure determination from scratch by fit to the PDF data



Structure determination from scratch by fit to the PDF data



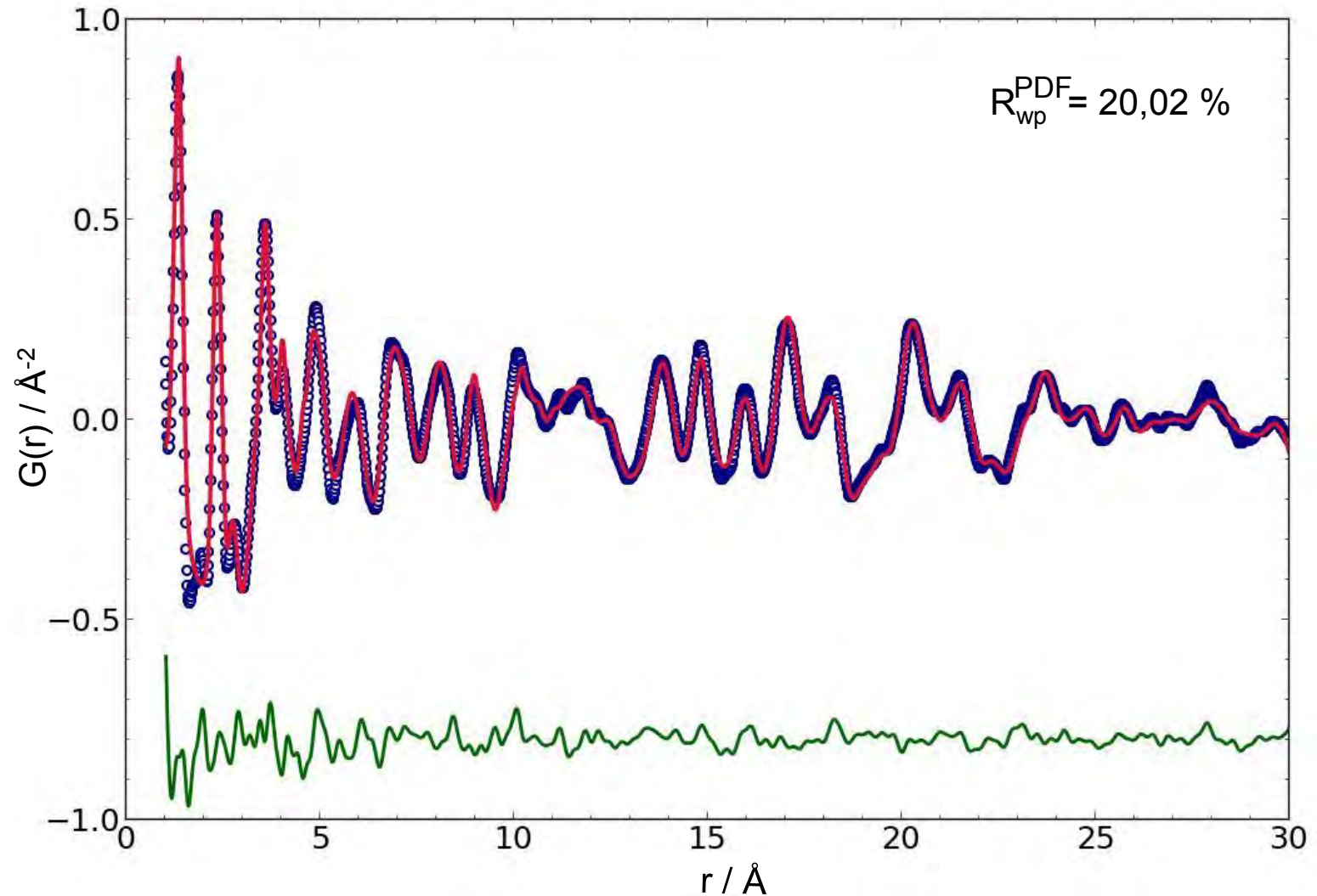
PDF derived by PDFgetX3^[*]

[*] P. Juhas et al., J. Appl. Cryst. (2013), 46, 560-566.

[Schlesinger, Habermehl, Prill, J. Appl. Cryst. 2021, 54, 776–786]

Structure determination from scratch by fit to the PDF data

Best solution in $P2_1/n$, $Z = 4$ (Correct space group)



Structure determination from scratch by fit to the PDF data

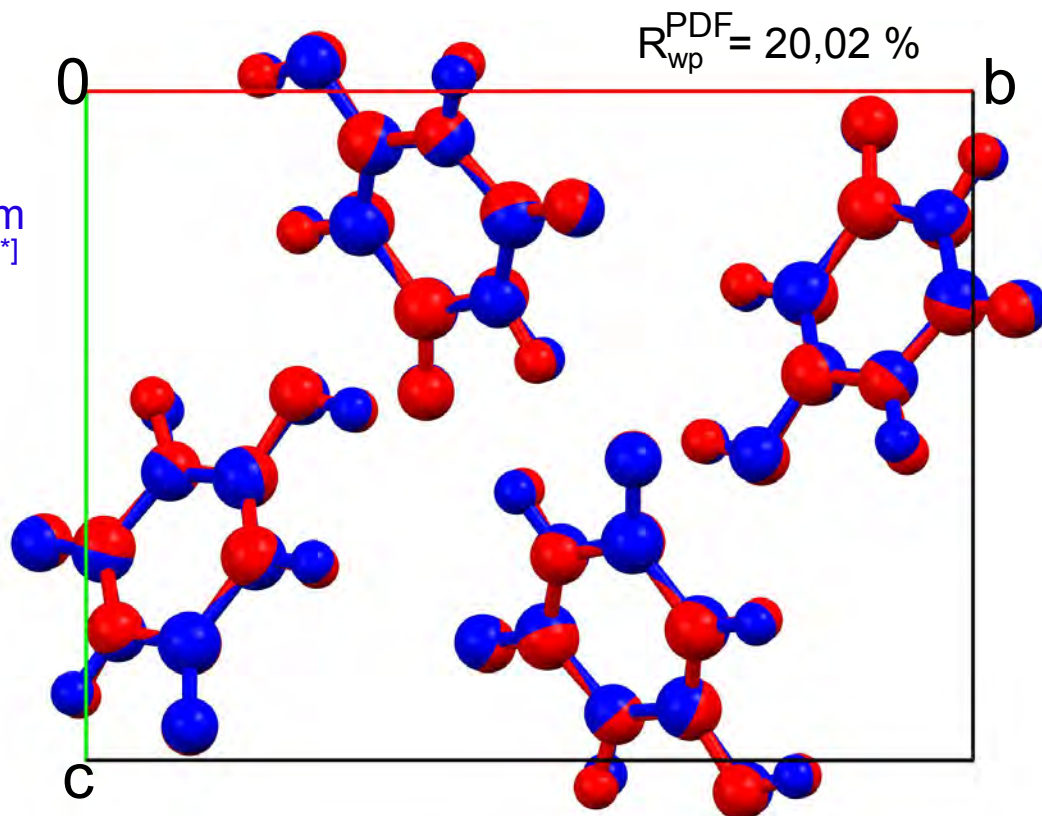
Best solution in $P2_1/n$, $Z = 4$ (Correct space group)

PDF structure

$a = 11.872 \text{ \AA}$
 $b = 8.937 \text{ \AA}$
 $c = 4.825 \text{ \AA}$
 $\beta = 95.06^\circ$

Crystal structure from
Rietveld refinement^[*]

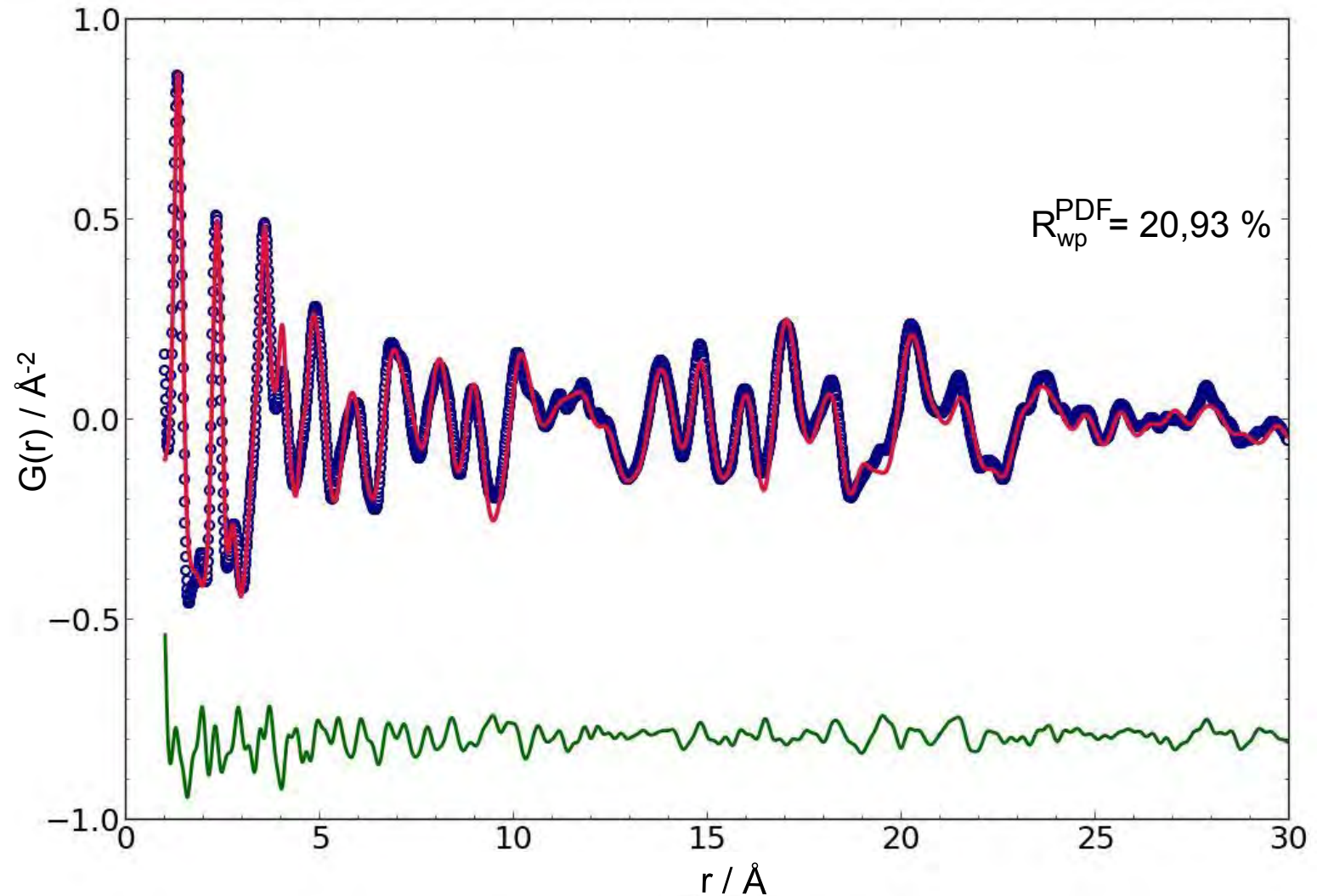
$a = 11.876 \text{ \AA}$
 $b = 8.915 \text{ \AA}$
 $c = 4.834 \text{ \AA}$
 $\beta = 95.08^\circ$



[*] Schmidt et al., Angew. Chem. (2011), 50, 7924.

Structure determination from scratch by fit to the PDF data

Test with $Z' = 2$ ($P \bar{1}$, $Z = 4$)



Structure determination from scratch by fit to the PDF data

Test with $Z' = 2$ ($P \bar{1}$, $Z = 4$)

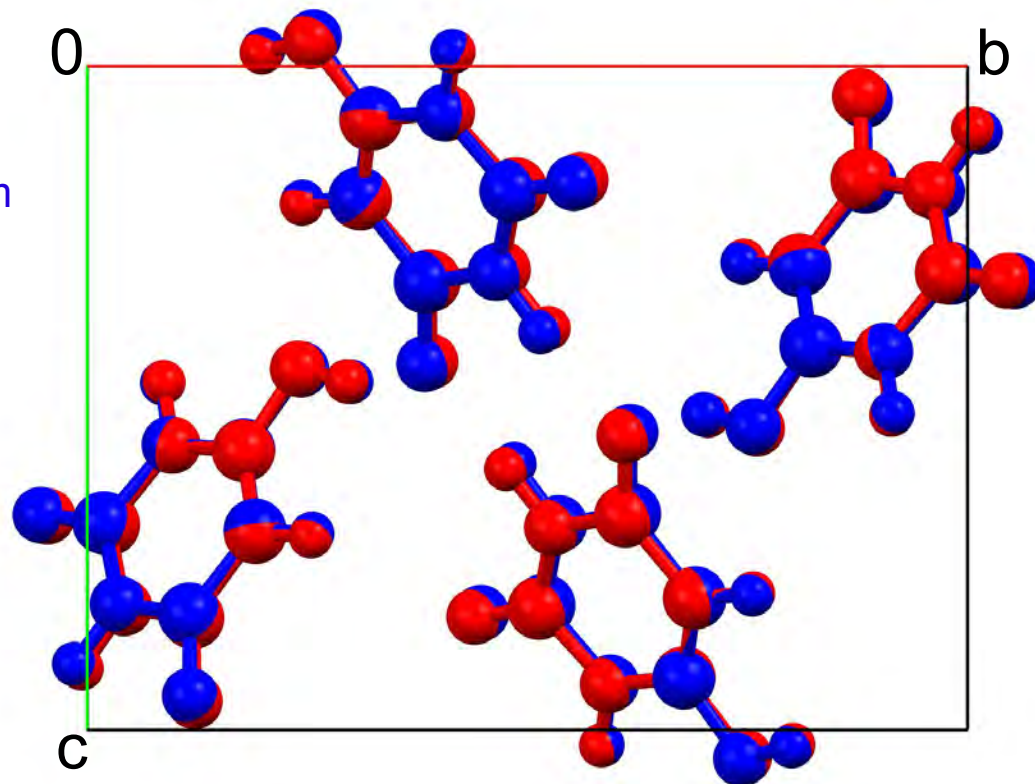
PDF structure

$a = 11.869 \text{ \AA}$
 $b = 8.941 \text{ \AA}$
 $c = 4.836 \text{ \AA}$
 $\alpha = 90.00^\circ$
 $\beta = 95.13^\circ$
 $\gamma = 90.00^\circ$

Crystal structure from
Rietveld refinement

$a = 11.876 \text{ \AA}$
 $b = 8.915 \text{ \AA}$
 $c = 4.834 \text{ \AA}$
 $\alpha = 90.00^\circ$
 $\beta = 95.08^\circ$
 $\gamma = 90.00^\circ$

$R_{wp}^{PDF} = 20.93 \%$



Structure determination from scratch by fit to the PDF data

Test with $Z' = 2$ ($P \bar{1}$, $Z = 4$)

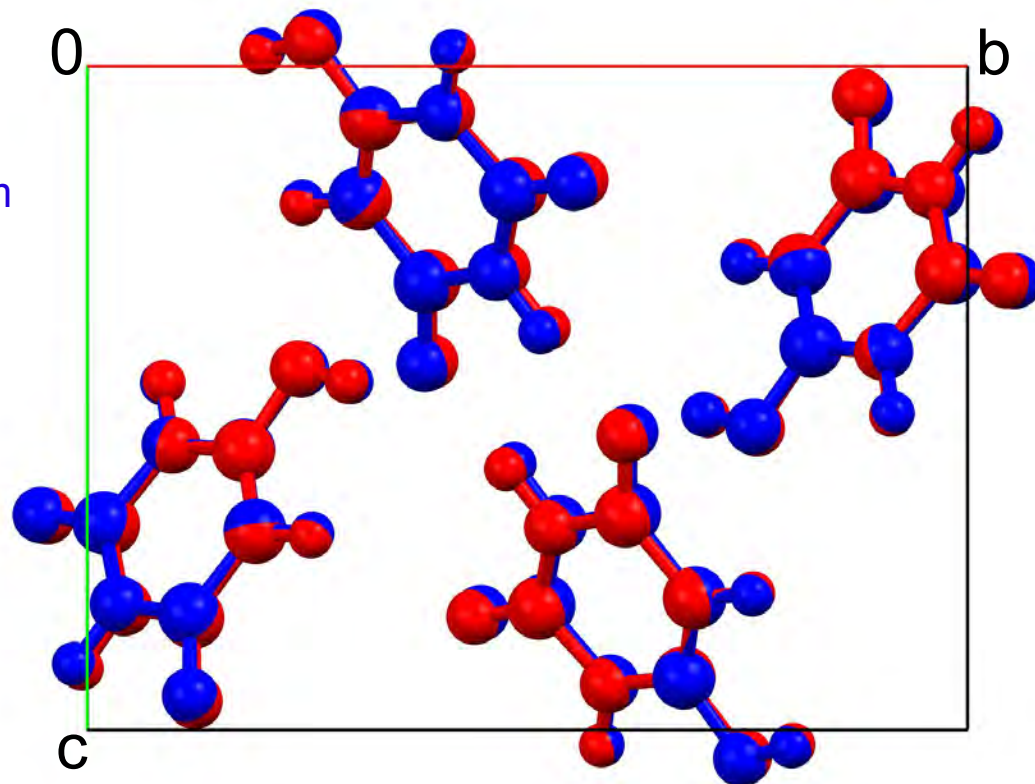
PDF structure

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 $\alpha = 90.00^\circ$
 $\beta = 95.08^\circ$
 $\gamma = 90.00^\circ$

$R_{wp}^{PDF} = 20,93 \%$



Second best solution:

- Similar lattice parameters

$R_{wp}^{PDF} = 24,62 \%$

- 2 of 4 molecules rotated by 180° (Typical effect, also observed by real-space methods)

Conclusion

Organic crystal structures can be solved and refined by a fit to the PDF curve, if the molecular structure is given. Even if the lattice parameters and space group are unknown.

Method development ongoing

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Dr. Carina Schlesinger

Stefan Habermehl

Dr. Stefan Brühne

Edith Alig

Dr. Lothar Fink

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ESRF, PSI, APS, NSLS, Elettra, Desy

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